

Laboratory of Mathematical Chemistry
Bourgas University, Bulgaria

**Using CATABOL to predict persistency,
biodegradation pathways and stable
degradants**

International Science Forum, *Computational Toxicology*

US EPA, May 21-23, 2007

Outlook

- **QSAR and Complexity of Chemical Structure**
- **Toxicity as a result of metabolic activation**
- **Metabolism logic**
- **Probabilistic approach to modeling metabolism**
- **CATABOL for simulating microbial degradation**
- **Performance and reliability of predicted metabolites**
- **Biodegradation kinetic models**
- **Simulating the effect of gene modification on metabolism**

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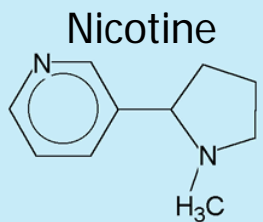


2D-3D Migration

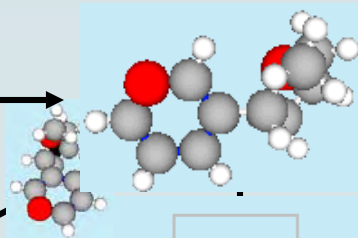
Molecular Flexibility



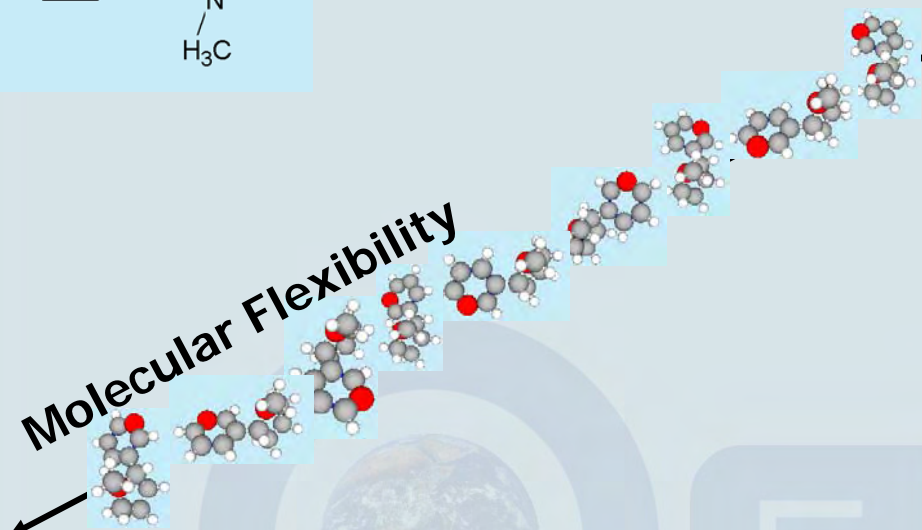
LMC
FSE
Laboratory of Mathematical Chemistry



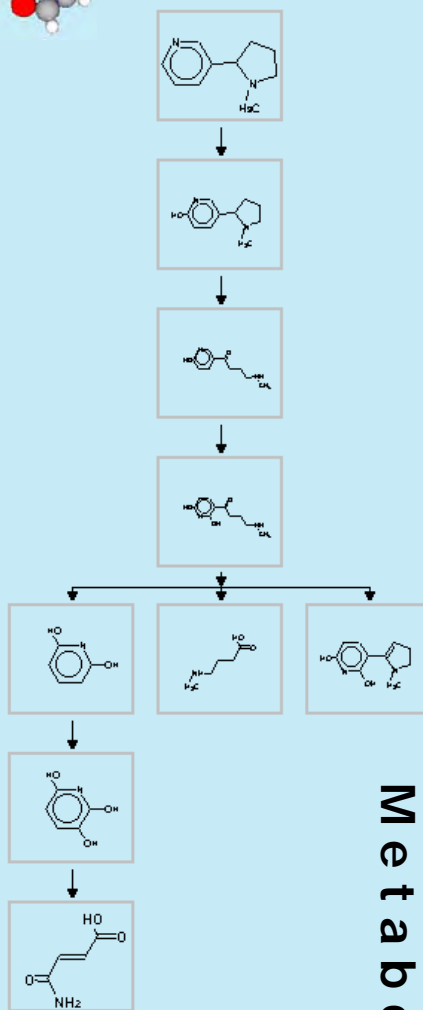
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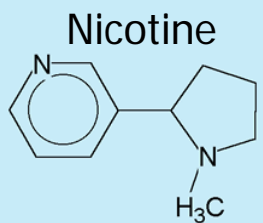


Microbial Degradation

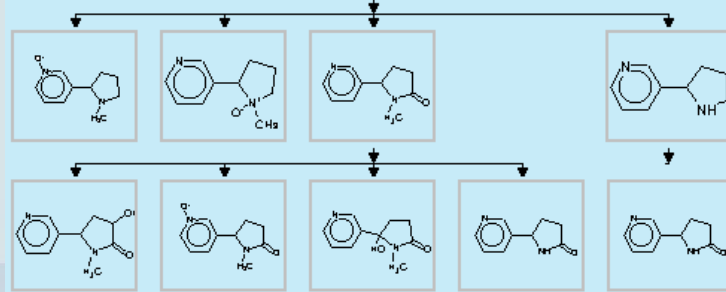
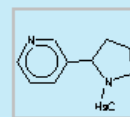
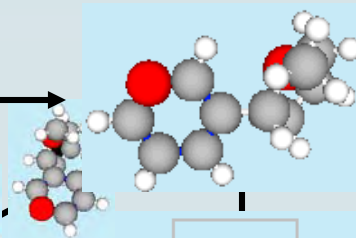


Metabolism





2D-3D Migration

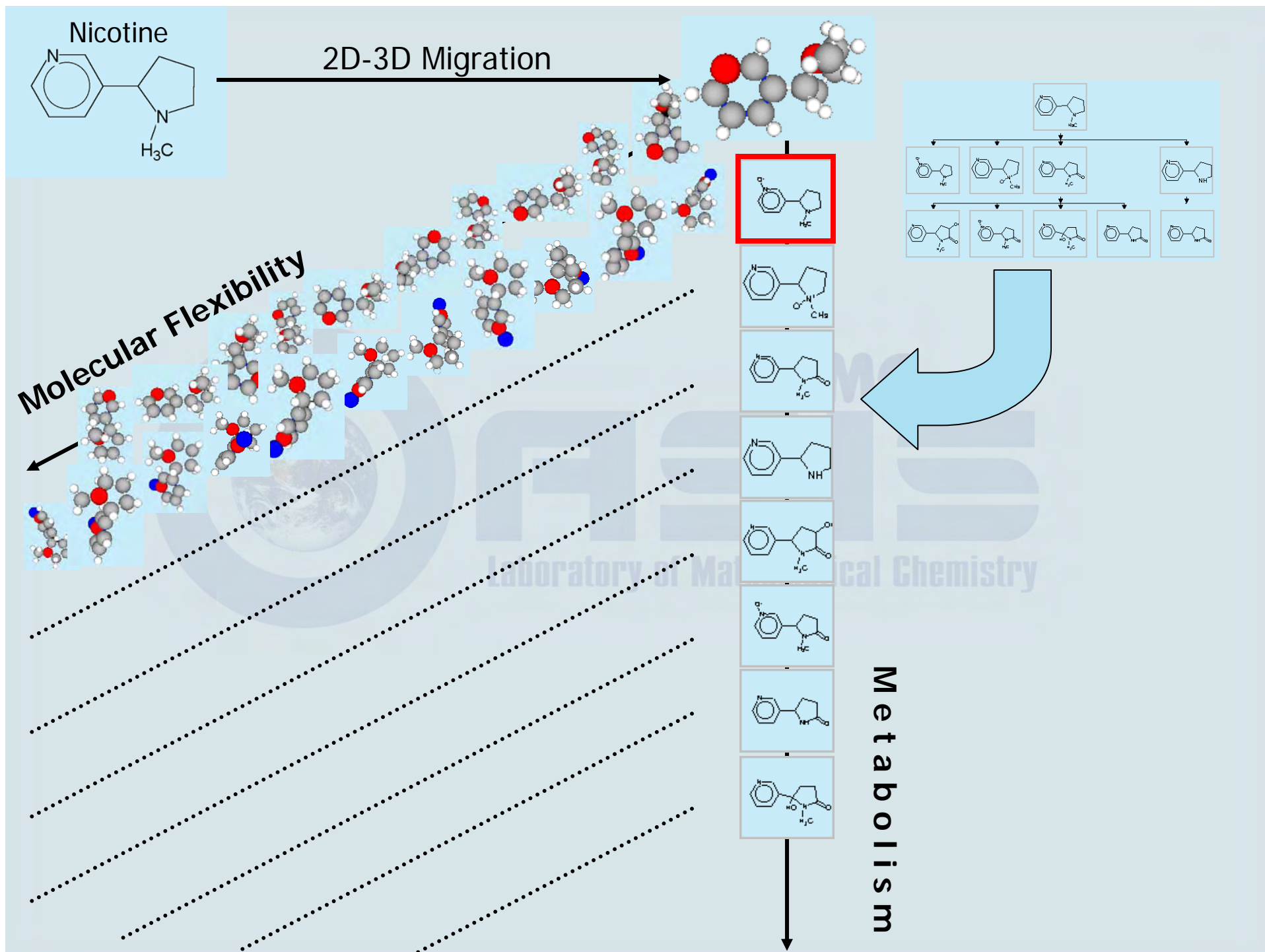


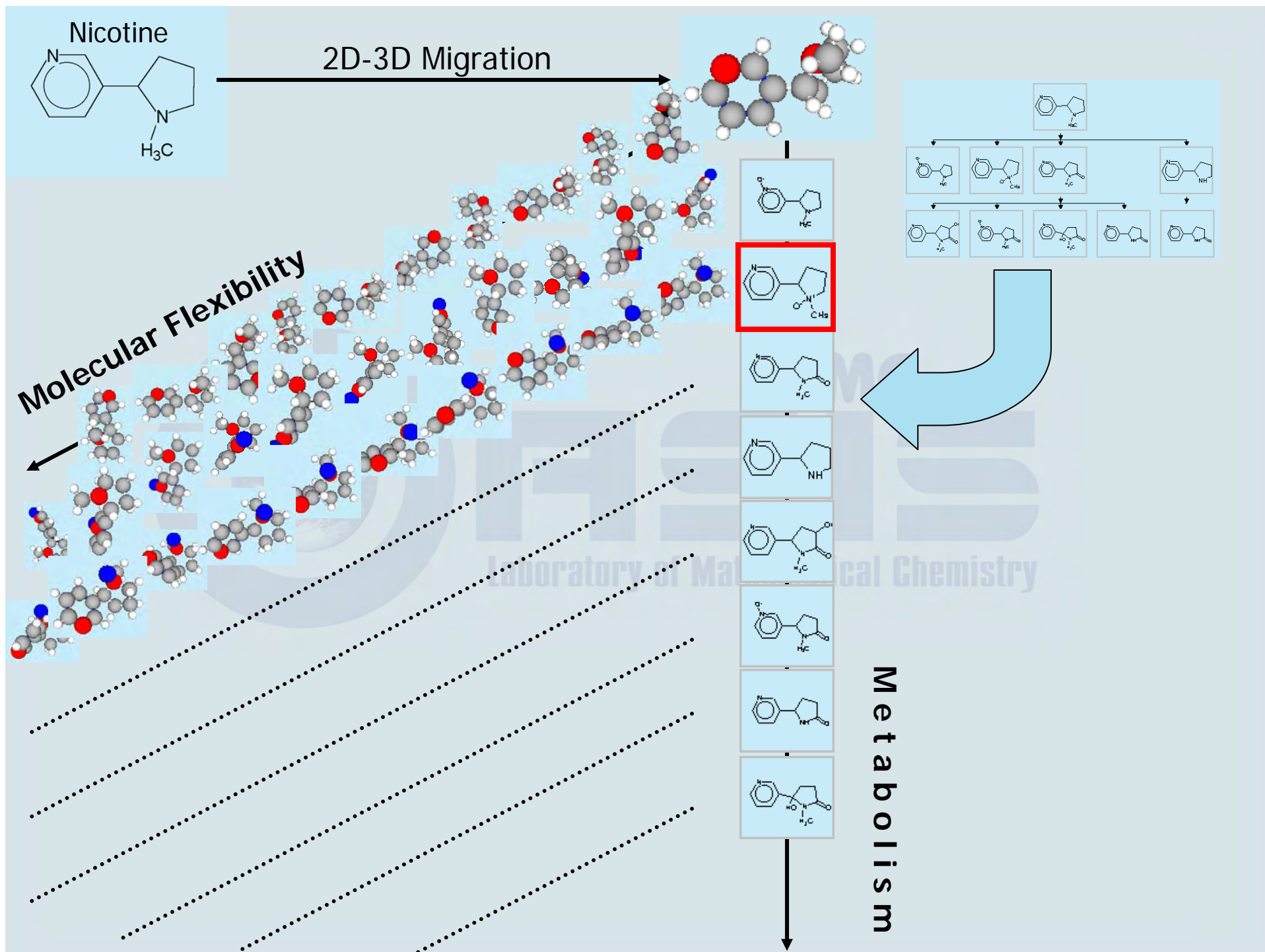
Metabolism

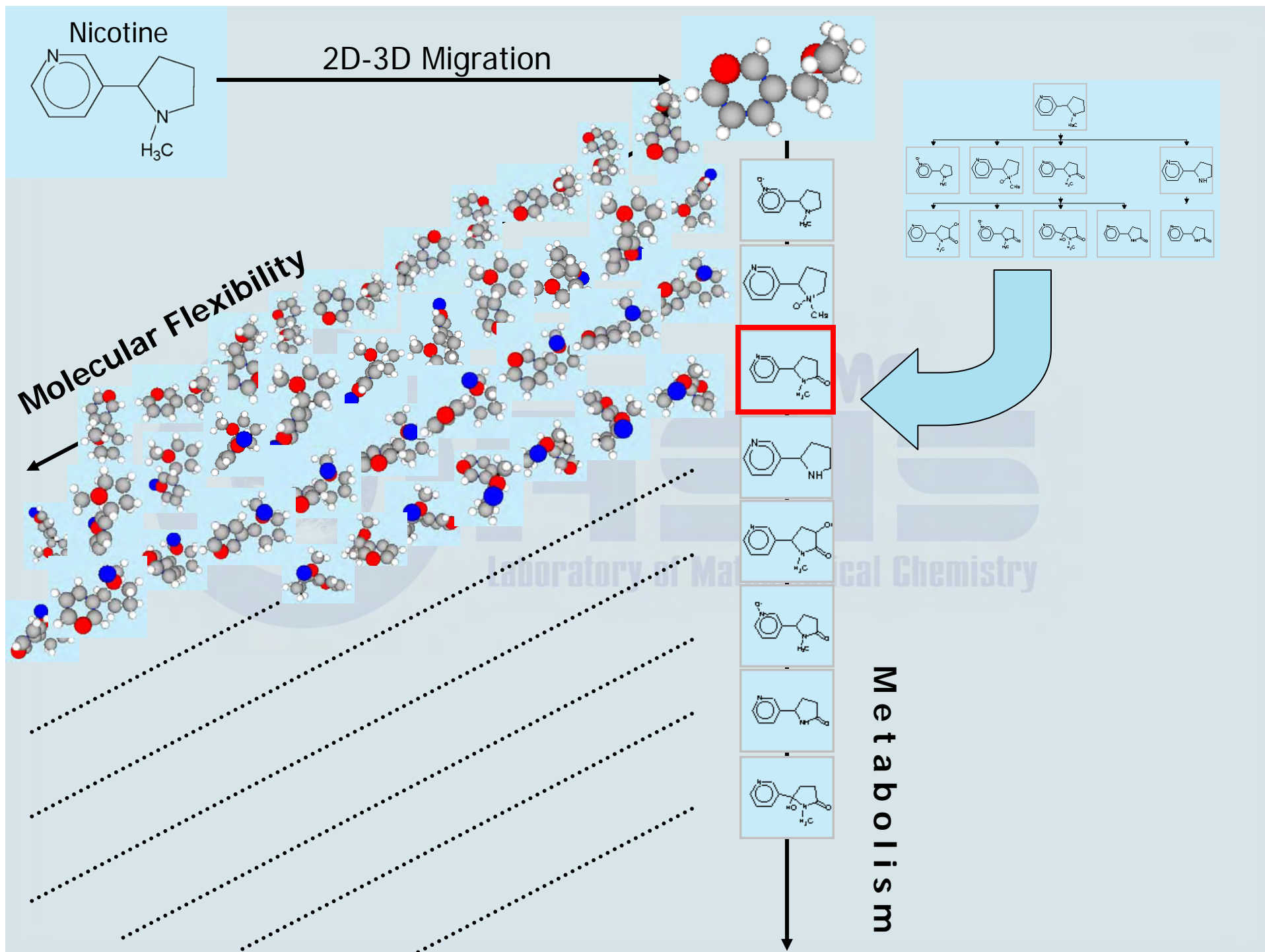
Rat Liver Metabolism

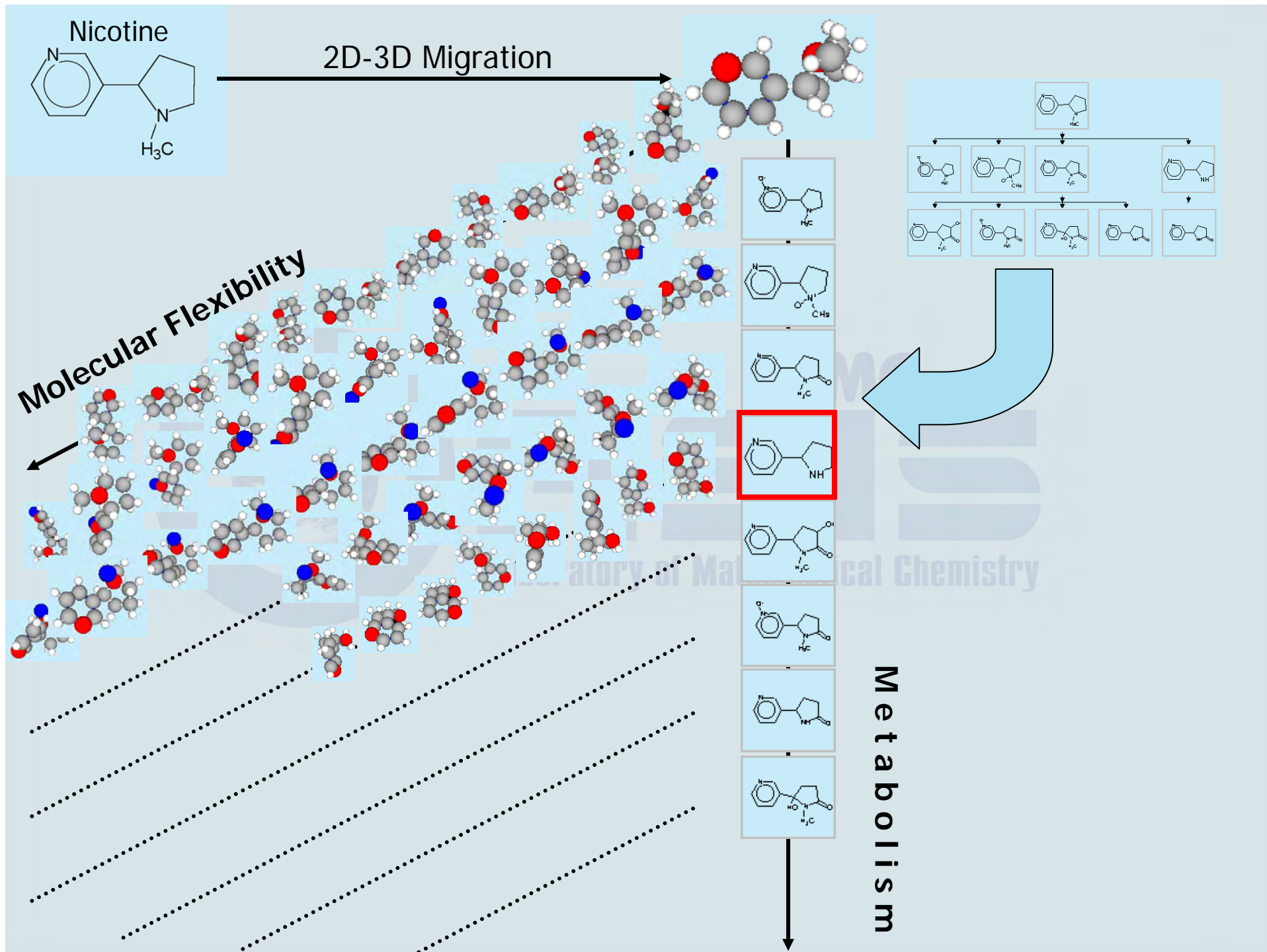
Molecular Flexibility

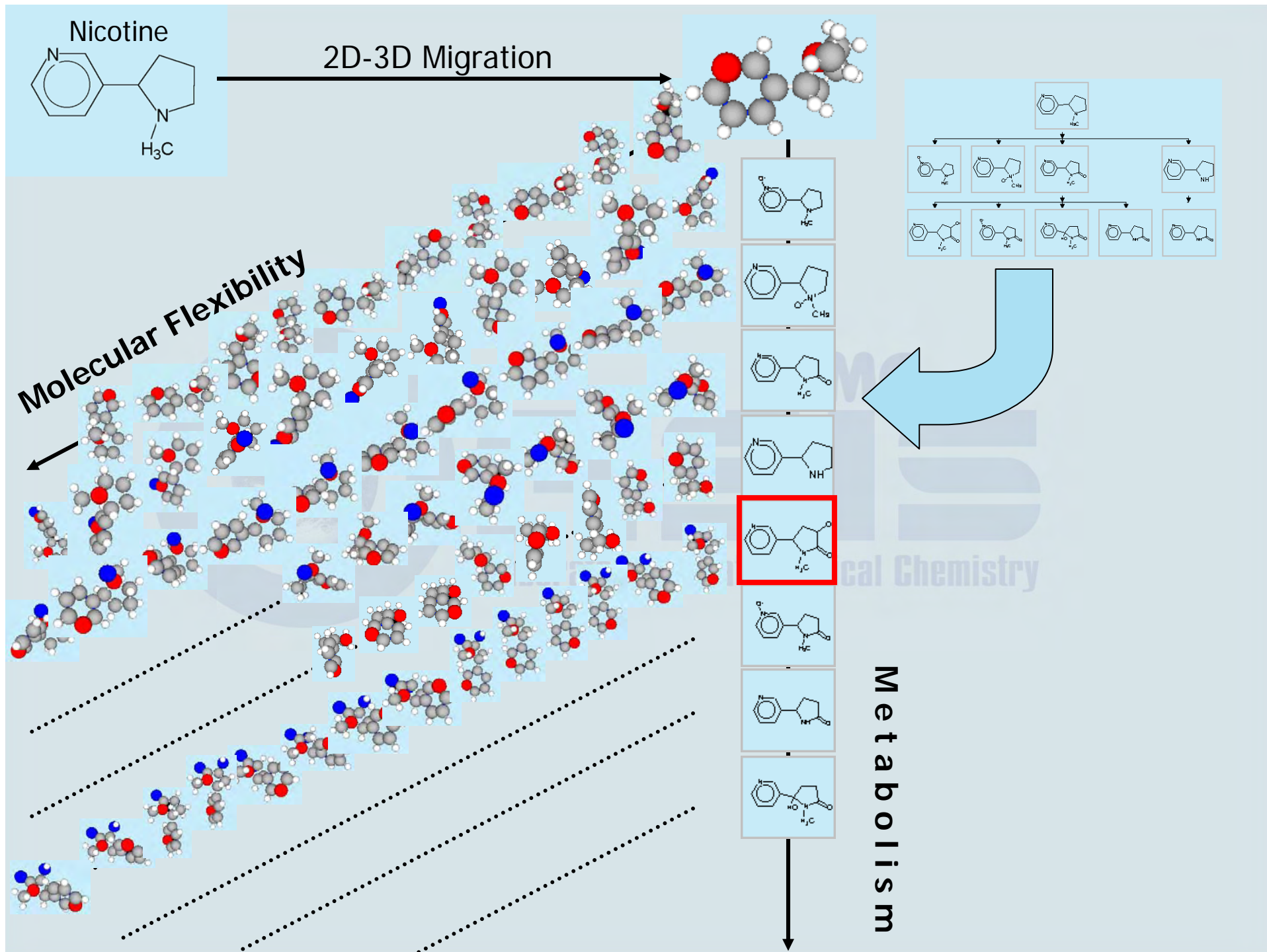
Laboratory of Mathematical Chemistry

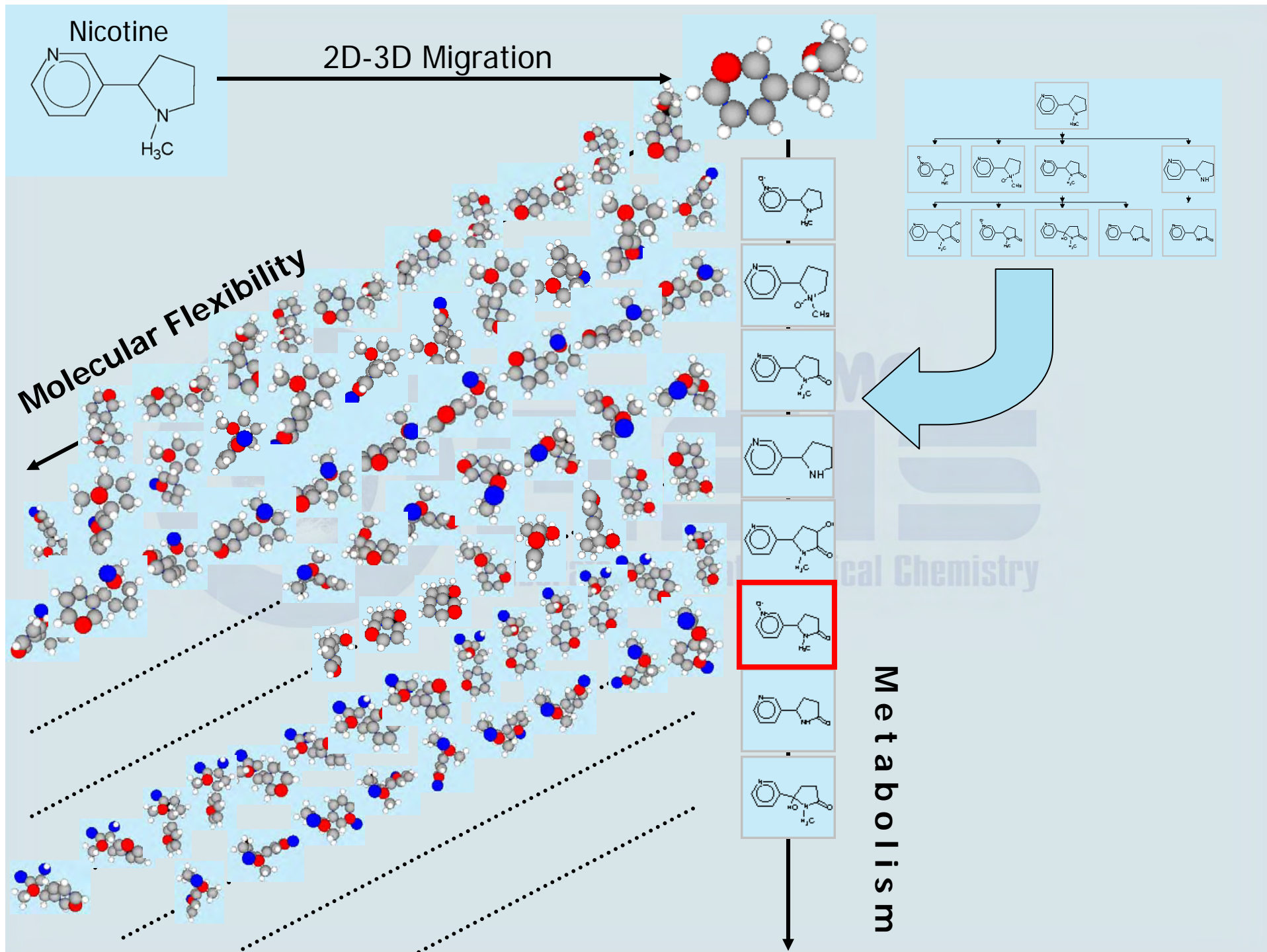


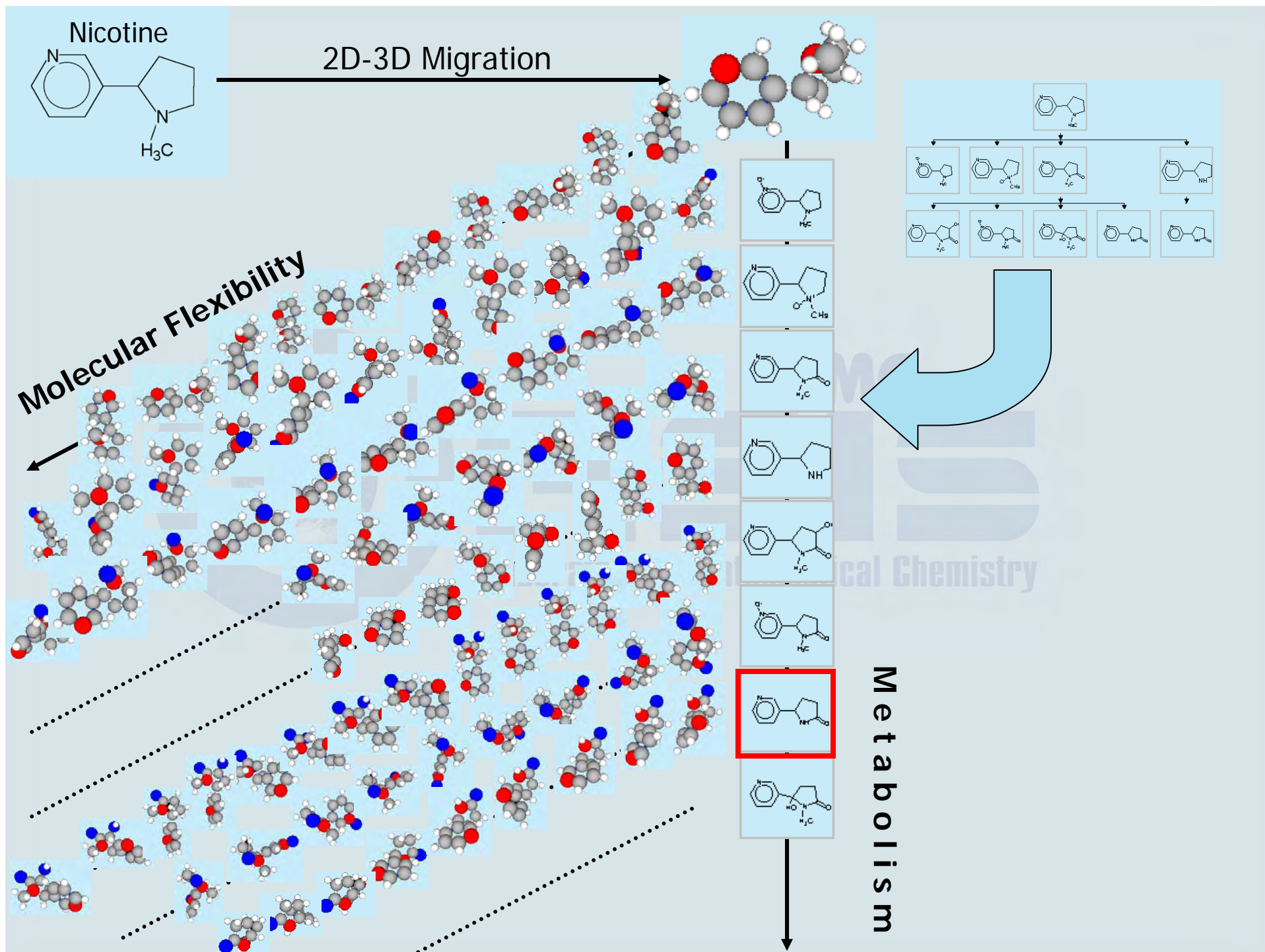


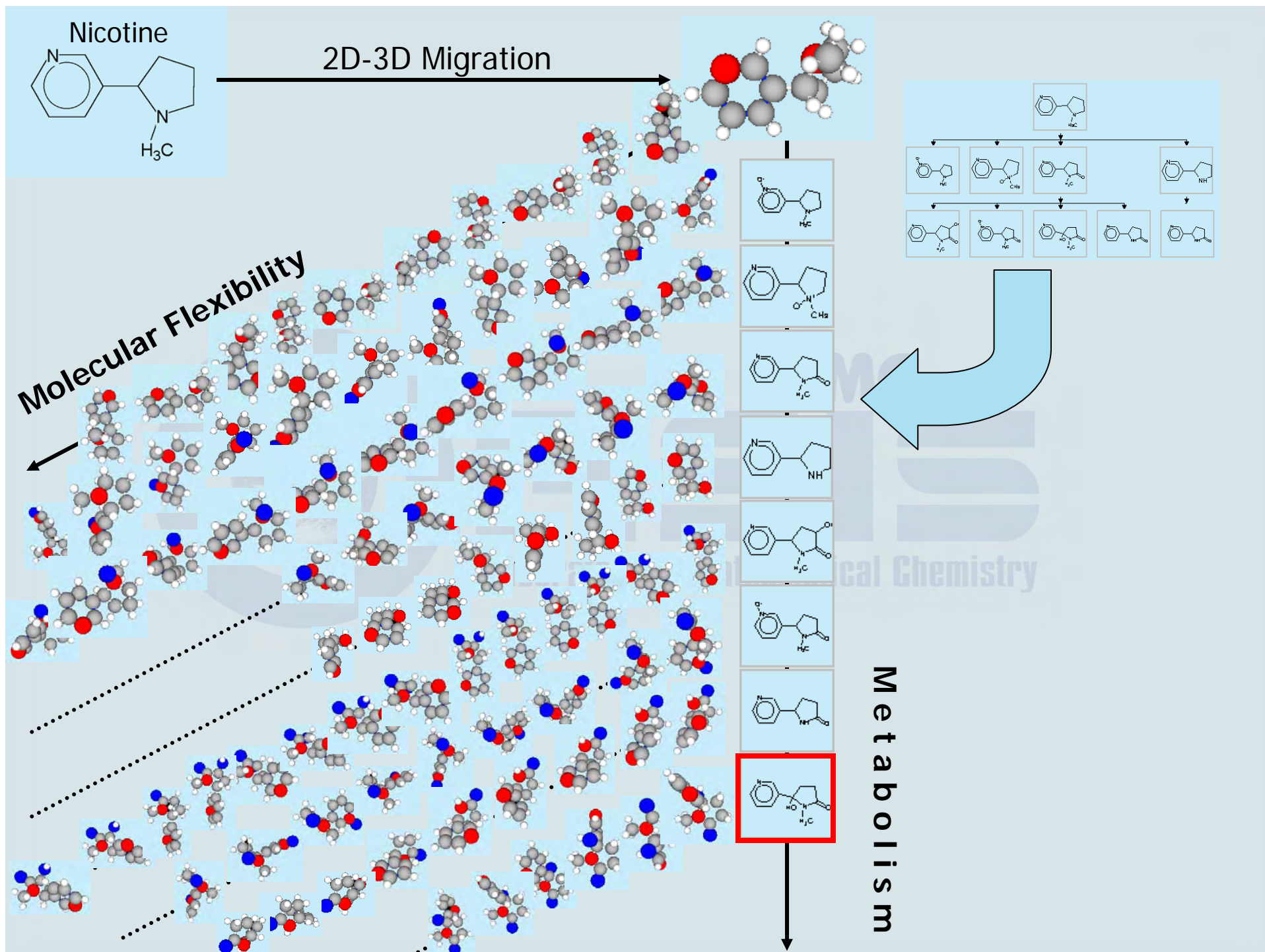


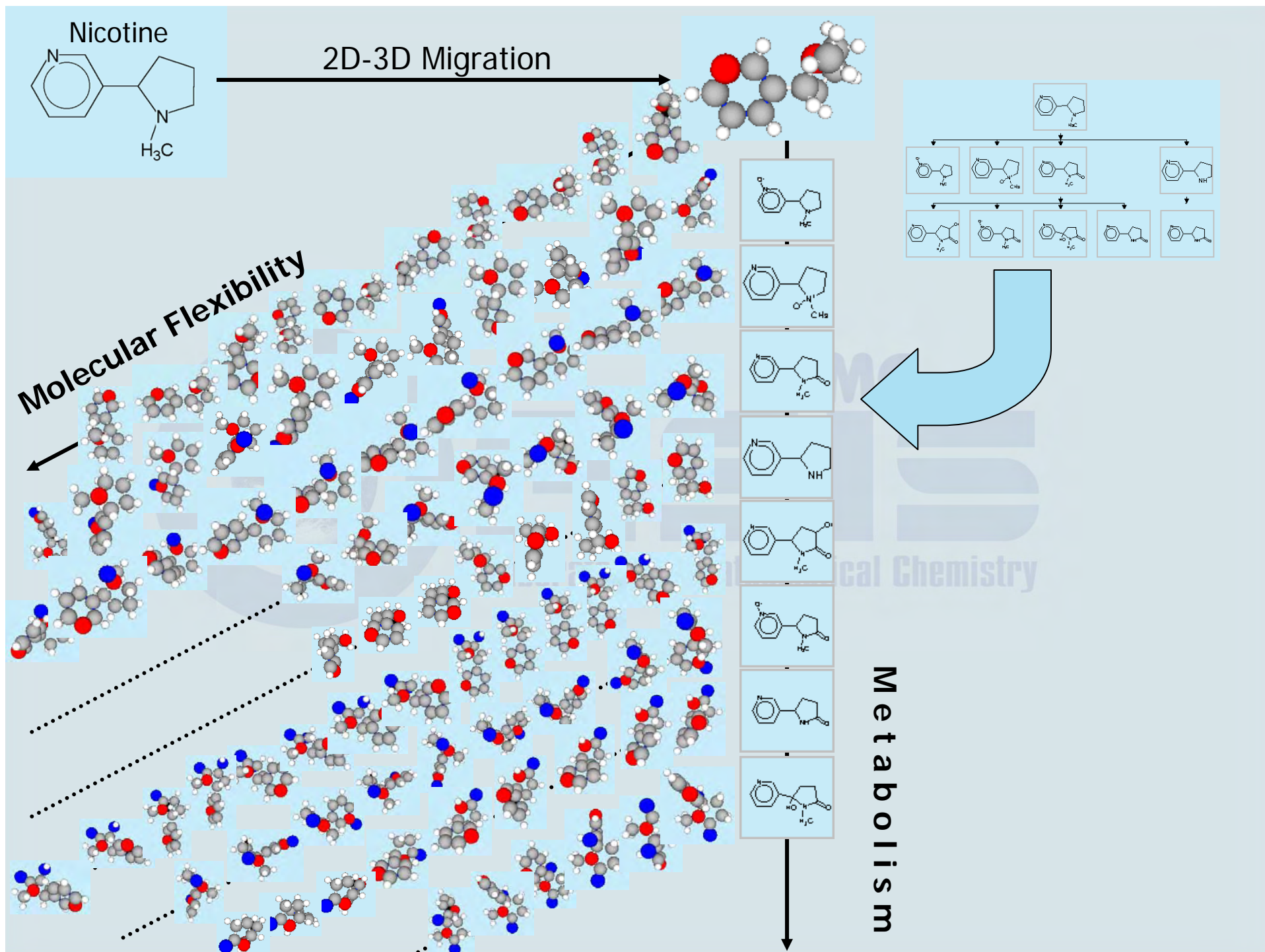








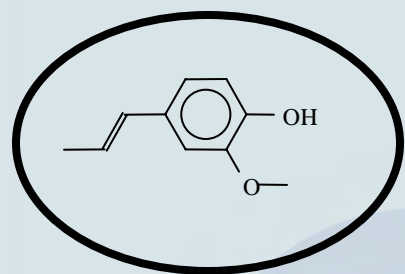




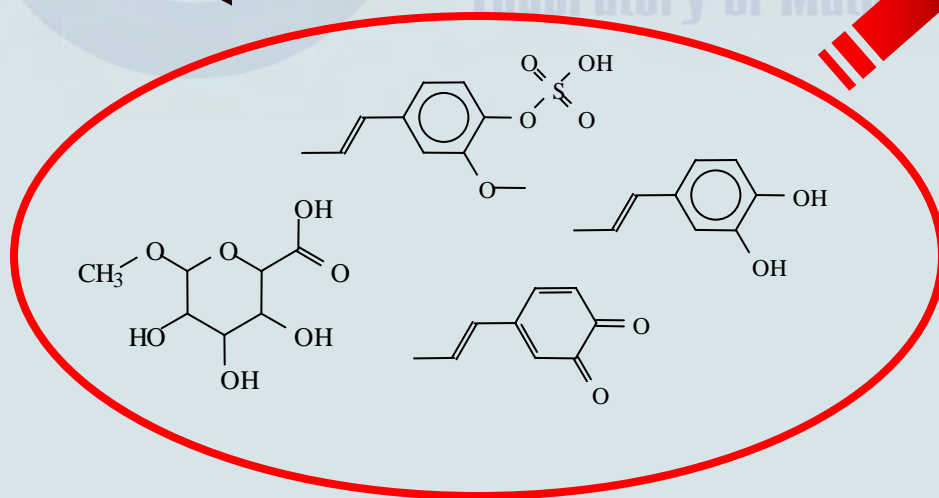
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(Q)SARs



Metabolism



Toxicodynamics

Biodegradation
Bioaccumulation
Acute Toxicity
Chronic Toxicity
Hormone Toxicity
Skin sensitization
Mutagenicity
...

The OASIS QSAR Concept :

To analyze toxicity as a result of metabolic activation

Combining on same modeling platform:

- **Toxicokinetics – specific metabolism**
- **Toxicodynamic – interaction with macromolecules**

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Metabolism

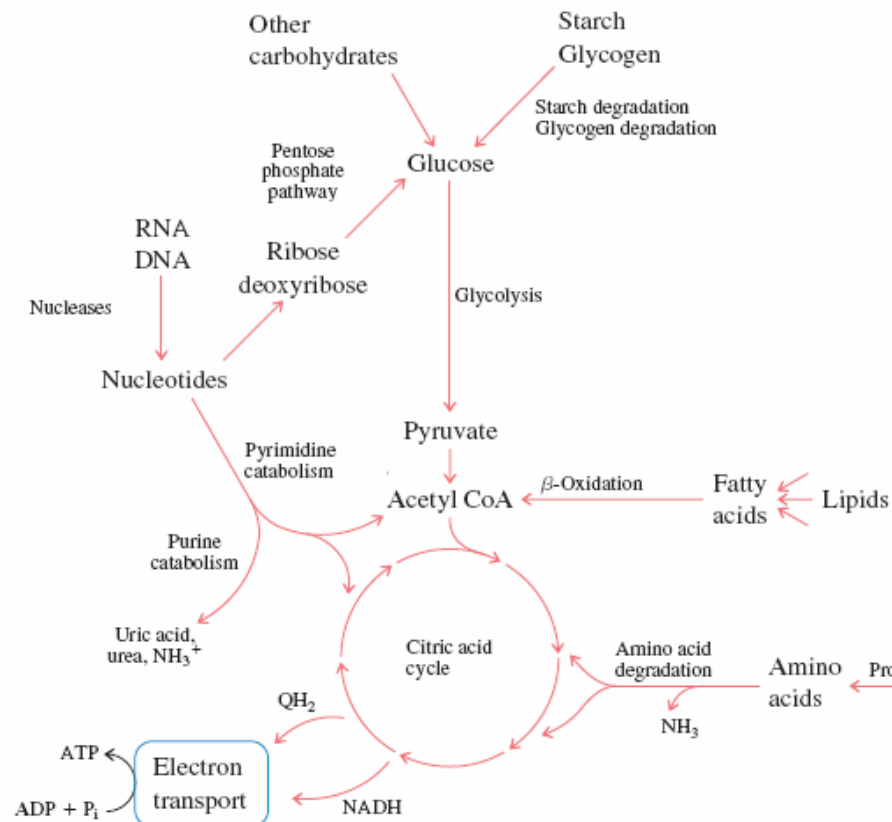
Energy-generating component: Catabolism

Produce energy (as ATP) and simple oxidized compounds

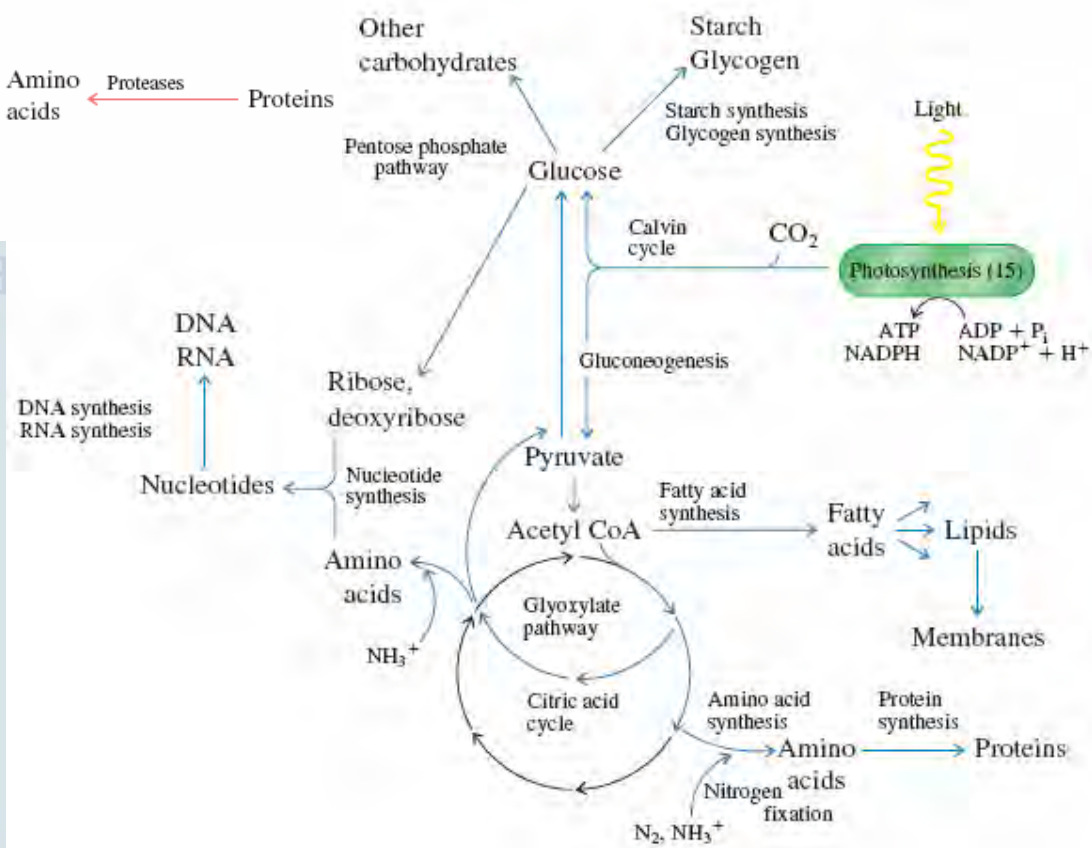
Energy-consuming component: Anabolism

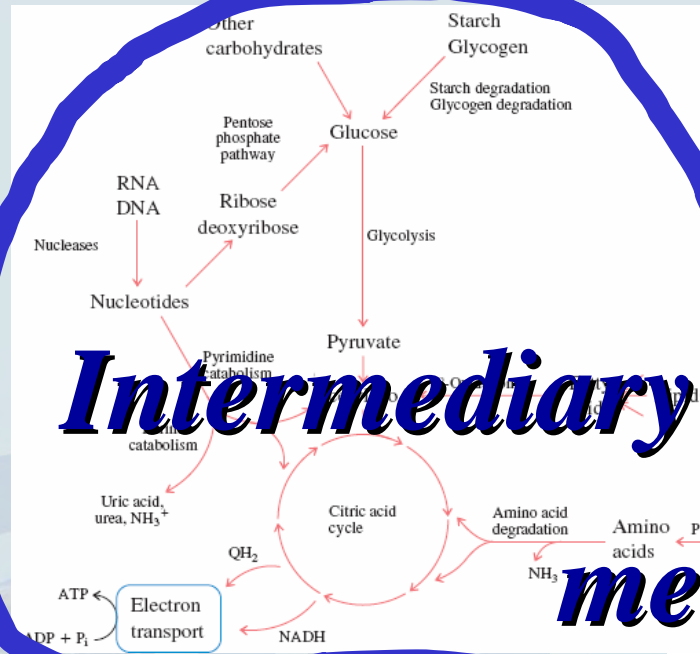
Build cell material





Catabolism

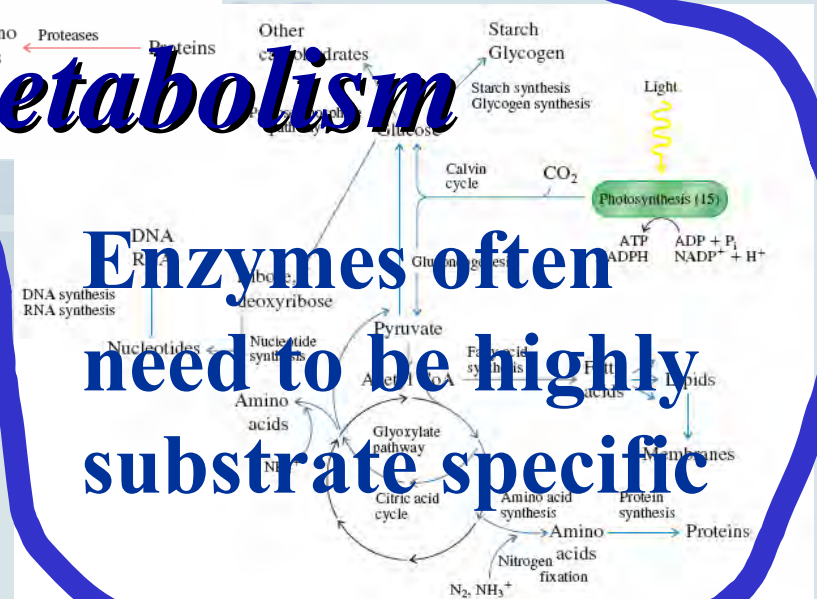




Intermediary

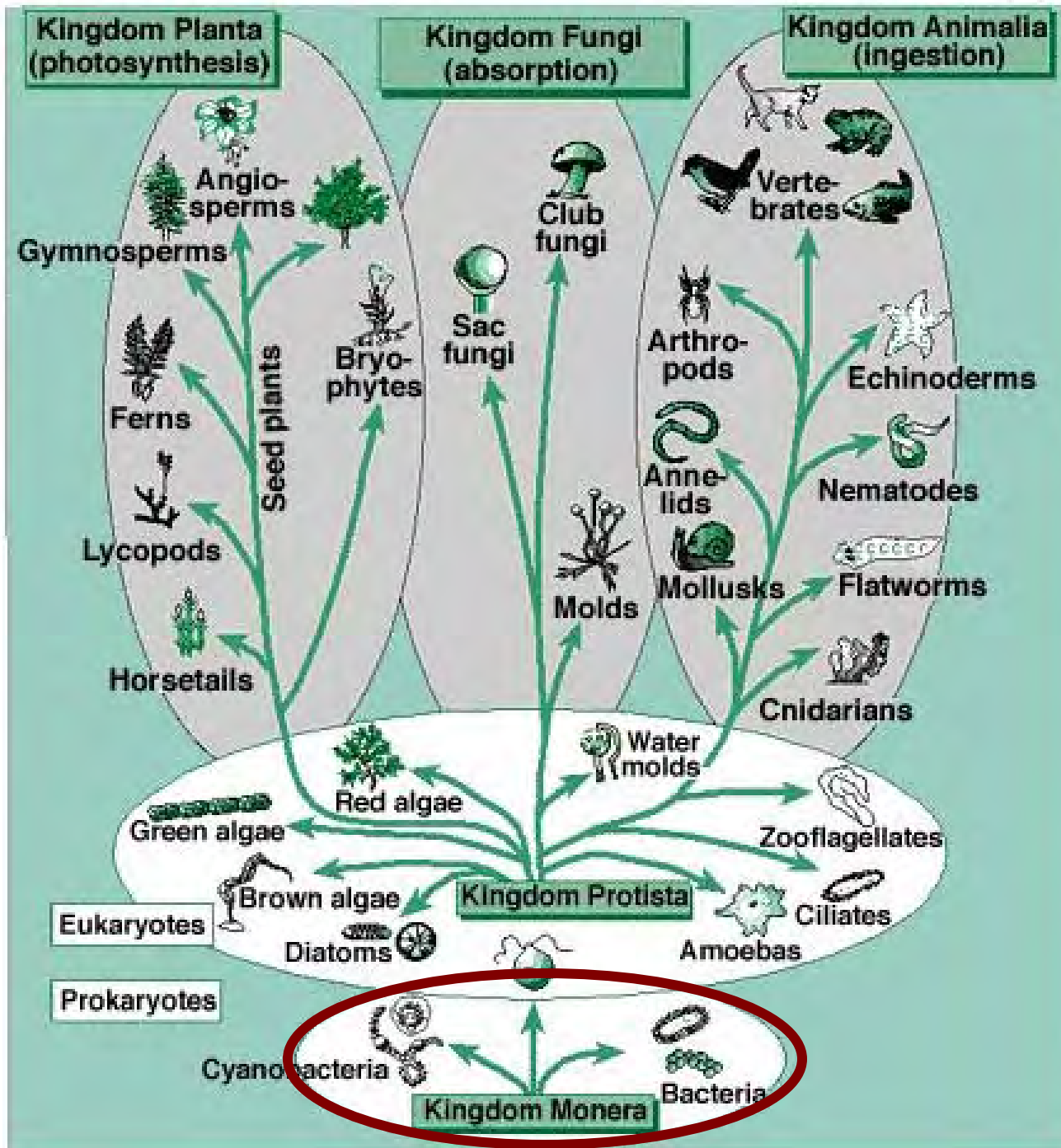
metabolism

**Non-intermediary
metabolism**



**Enzymes often
need to be highly
substrate specific**

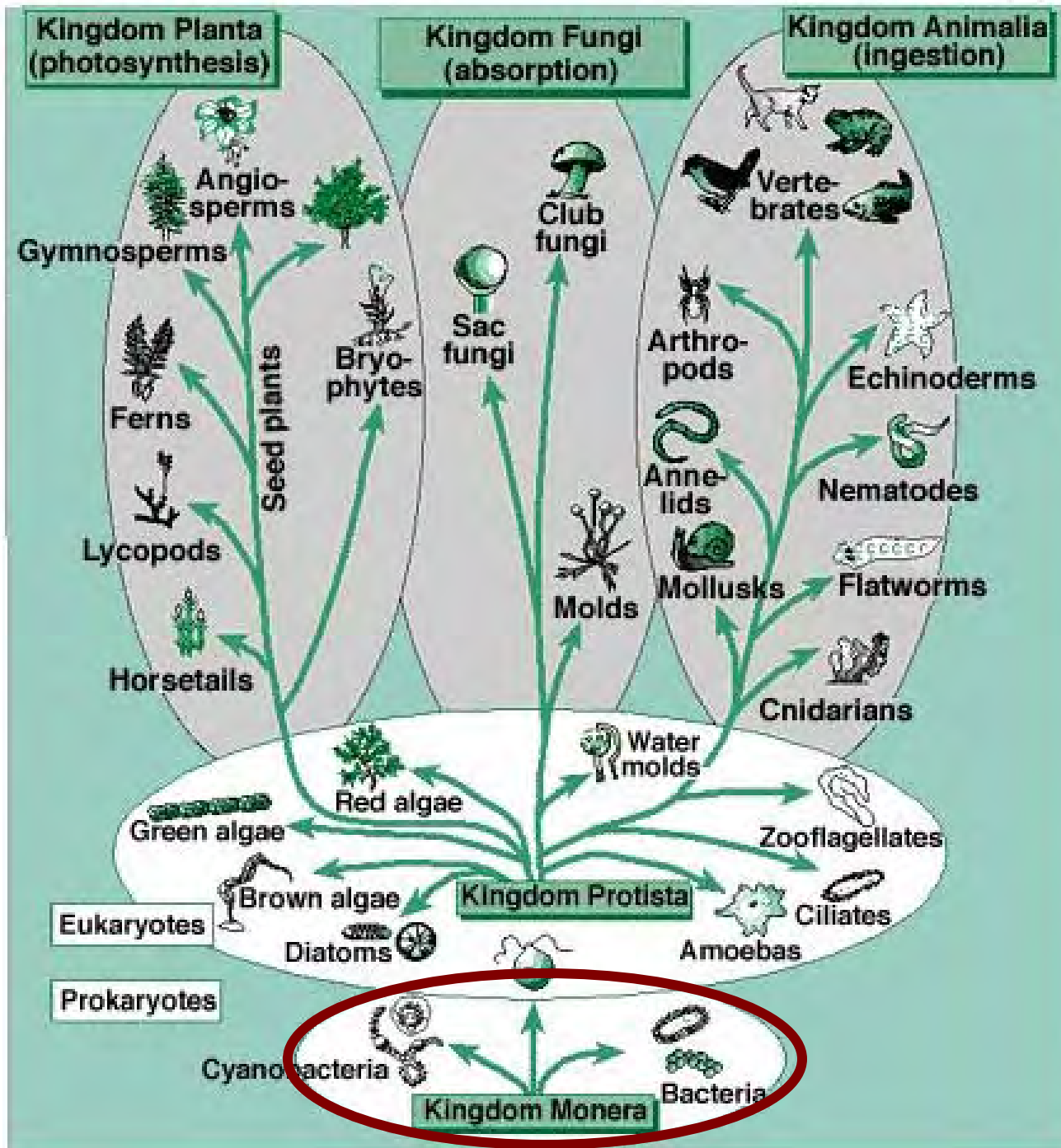
1. Response to environment
2. Response to human health



Prokaryotes

Kingdom Monera

Chemistry

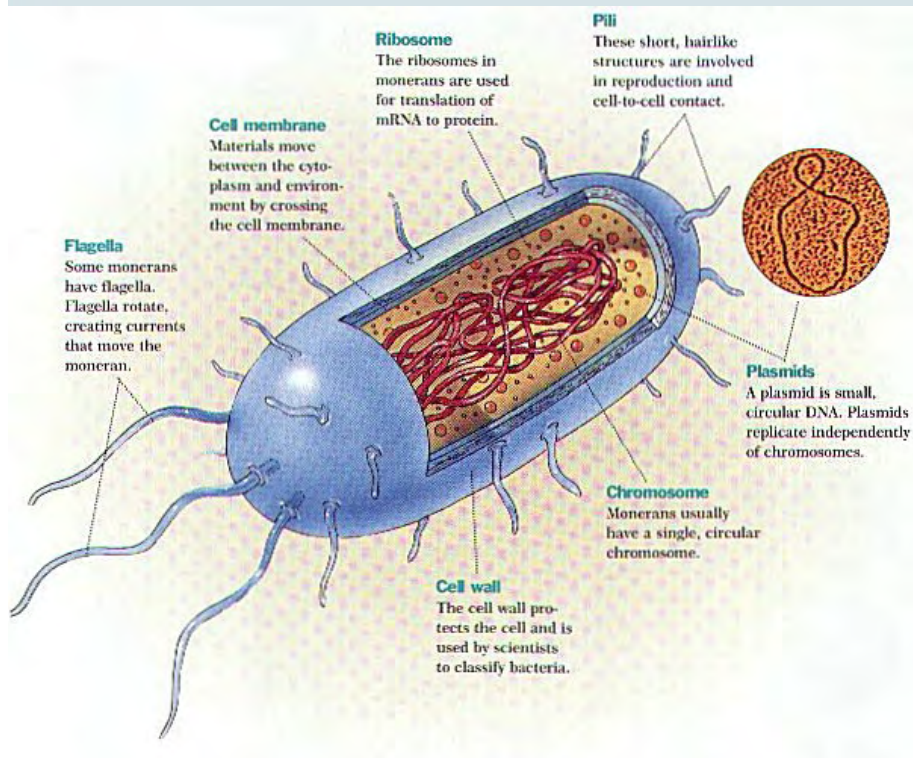


Prokaryotes

Kingdom Monera

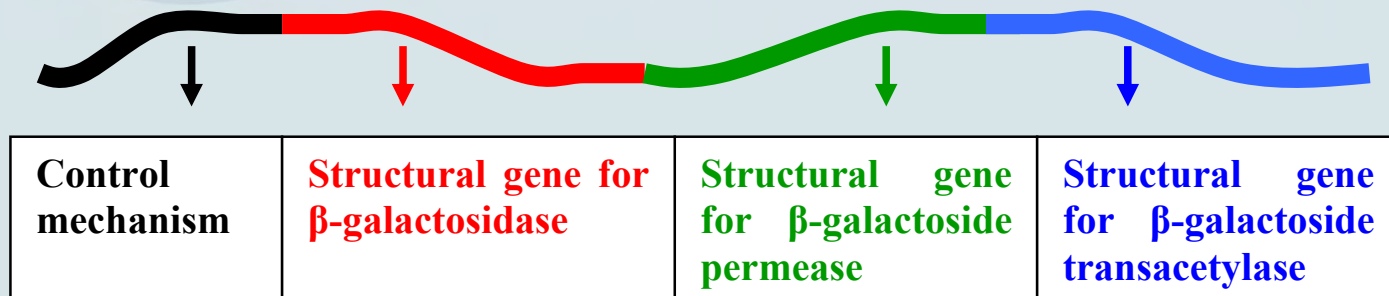
Chemistry

The Kingdom Monera (Prokaryotes)

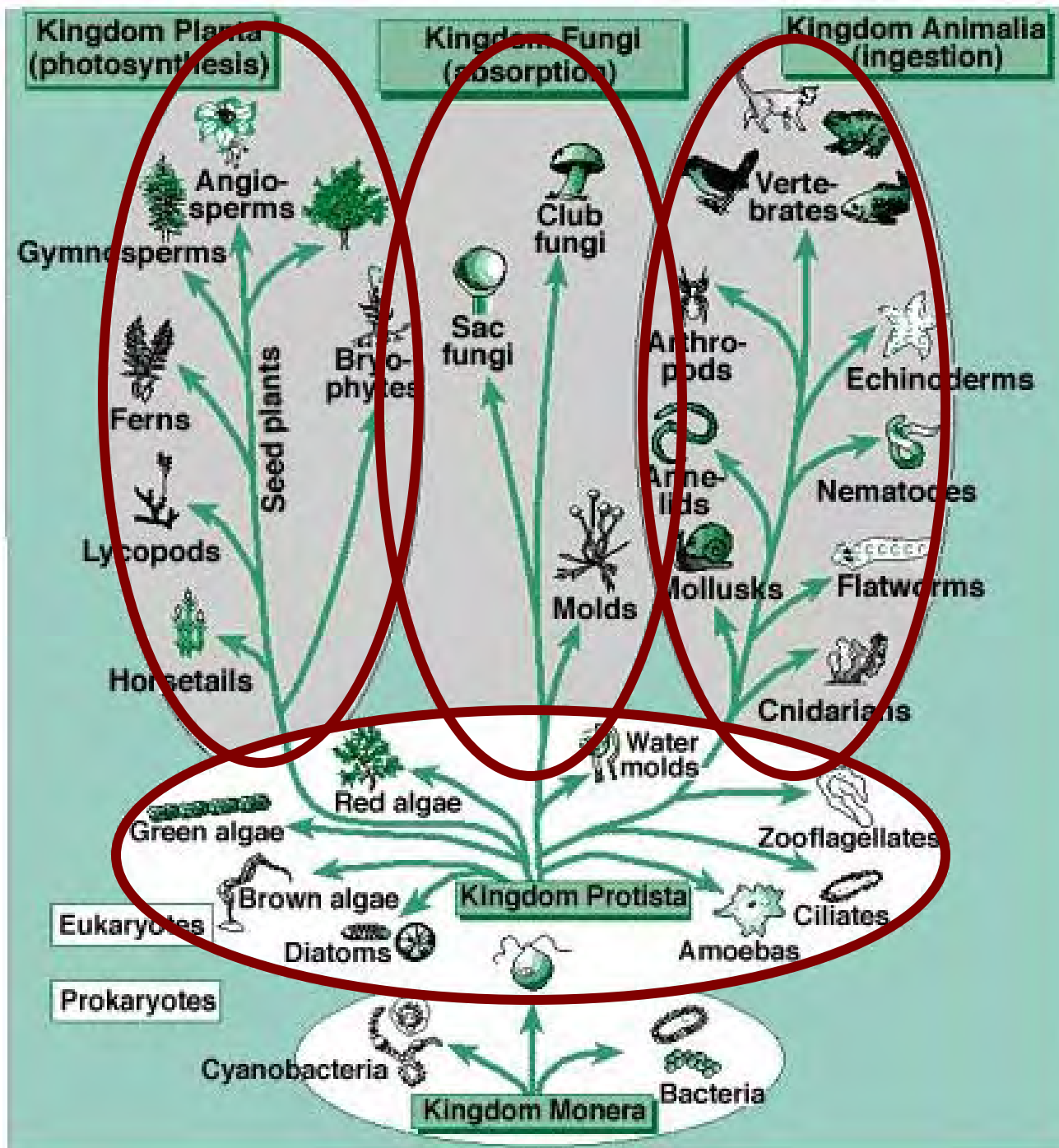


Bacteria tend to group genes that are functionally related together on the chromosome.

Operon - grouping of genes in bacteria under the control of the same regulatory system.



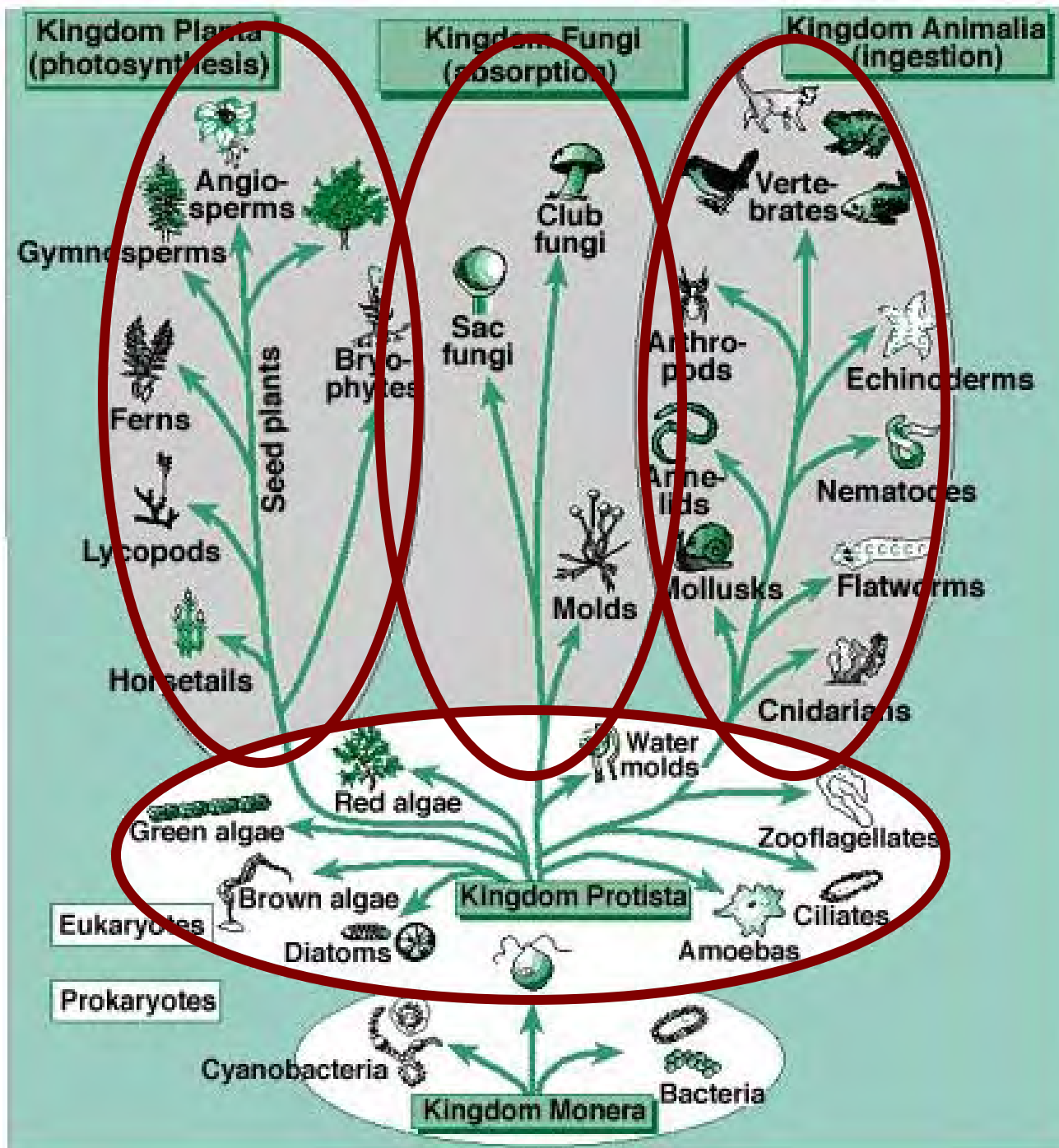
Lactose operon: contains genes that encode enzymes responsible for lactose metabolism



Eukaryotes

Kingdoms:
 Protista
 Planta
 Fungi
 Animalia

Chemistry

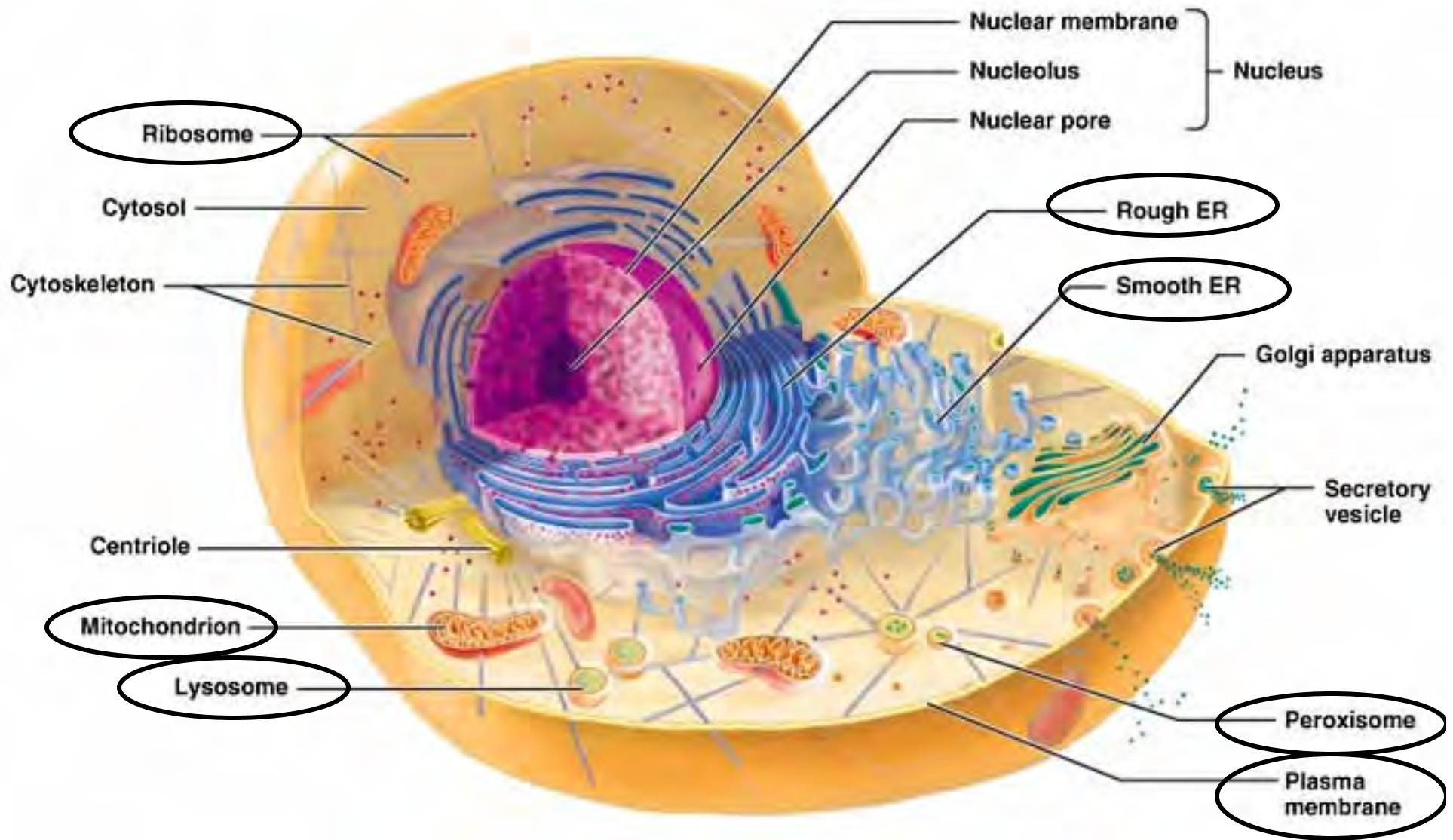


Eukaryotes

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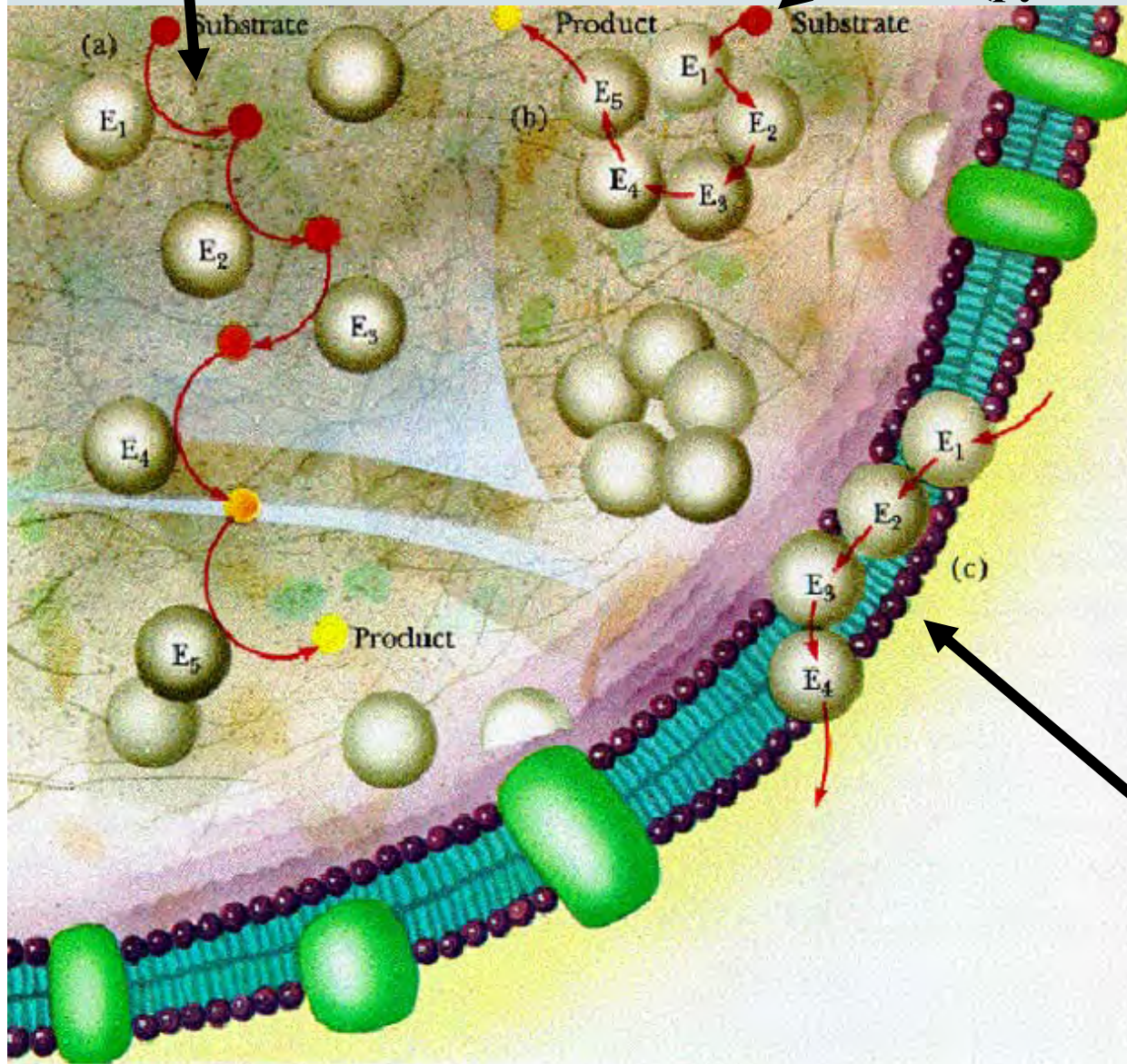
Chemistry

Eukaryotes



**Separate enzymes with freely
diffusing pathway intermediates
(glycolysis)**

**Cytosolic multienzyme complex with
intermediates channelled between
enzymes
(pyruvate dehydrogenase complex)**



**Membrane bound
multienzyme complex
(electron transport system)**

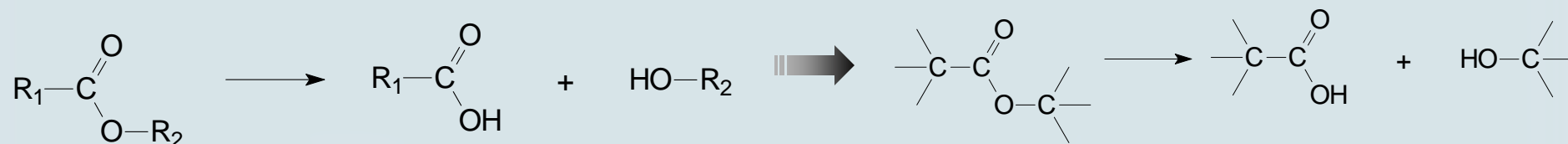
Summary

1. The application of metabolic transformations is strongly organized.
2. This is a premise for development of metabolic simulators as a hierarchically organized list of reactions

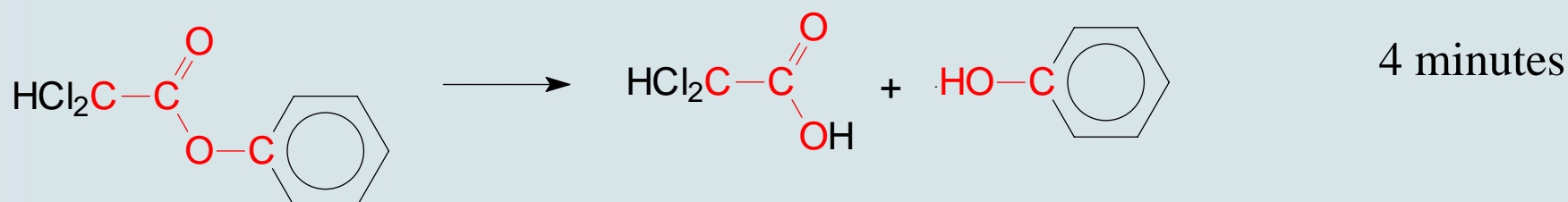
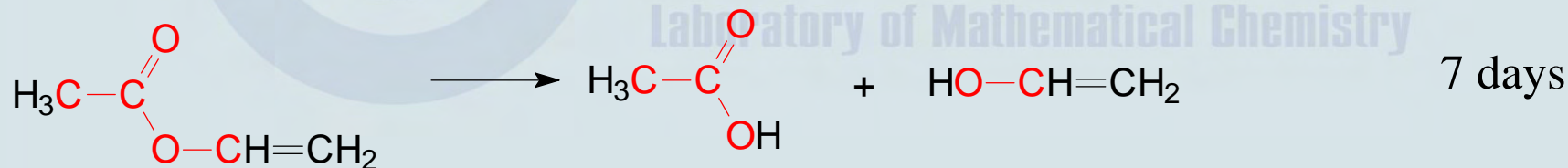
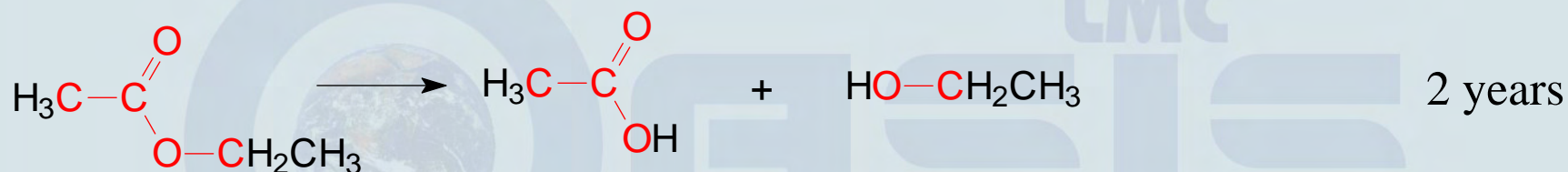
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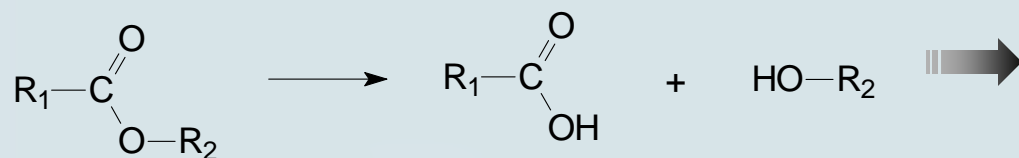
Simulation of molecular transformations



**Half-lives,
25°C, pH = 7**



Simulation of molecular transformations



#	Transformation rules	Rate
1	$X_2C-\overset{\overset{O}{\parallel}}{C}-O-C\{sp^2\} \longrightarrow X_2C-\overset{\overset{O}{\parallel}}{C}-OH + HO-C\{sp^2\}$ <p style="text-align: center;">X: F, Cl, Br</p>	High
2	$-C-\overset{\overset{O}{\parallel}}{C}-O-C\{sp^2\} \longrightarrow -C-\overset{\overset{O}{\parallel}}{C}-OH + HO-C\{sp^2\}$ <p style="text-align: center;">not CX₂; X: F, Cl, Br</p>	Moderate
3	$-C-\overset{\overset{O}{\parallel}}{C}-O-C- \longrightarrow -C-\overset{\overset{O}{\parallel}}{C}-OH + HO-C-$ <p style="text-align: center;">not C{sp²} and CX₂; X: F, Cl, Br</p>	Low

Simulators of metabolism

Rule based systems

BESS (P&G and Michigan State University)

META (MultiCASE Inc)

METEOR (Lhasa Ltd)

CATABOL (P&G and LMC, Bourgas As. Zlatarov University)

TIMES (LMC, Bourgas As. Zlatarov University)

PPS (UM-BBD, <http://umbbd.msi.umn.edu/>)

MEPPS (under development, Lhasa Ltd)

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Simulators of metabolism

Rule based systems

METEOR (Lhasa Ltd)

PPS (UM-BBD)

	Very likely
	Likely
	Neutral
	Unlikely
	Very unlikely
	Unknown

Simulators of metabolism

Rule based systems

CATABOL (P&G and LMC, Bourgas As. Zlatarov University)

TIMES (LMC, Bourgas As. Zlatarov University)

Probability $\left\{ \begin{array}{l} \mathbf{k} \\ \mathbf{t}_{1/2} \end{array} \right\}$

What is a *METABOLIC SIMULATOR*?

Prioritized list of molecular transformations and substructure matching engine applying it:

- Including spontaneous and enzymatic reactions
- Probabilistic scheme for prioritization of reactions
- Organ/tissue specific

Illustrating Basic Principles of the Metabolic Simulators

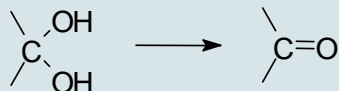


Simulator of metabolism

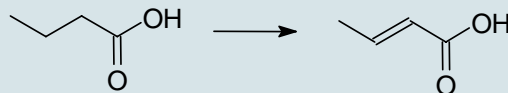
Principal transformations

Metabolites

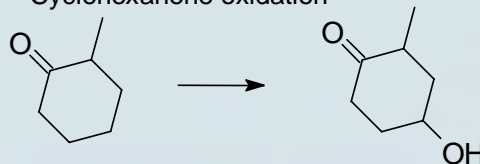
Geminal diol decomposition



β -oxidation



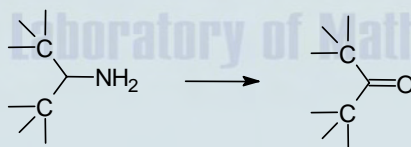
Cyclohexanone oxidation



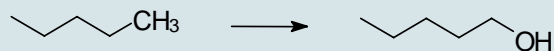
Ester hydrolysis



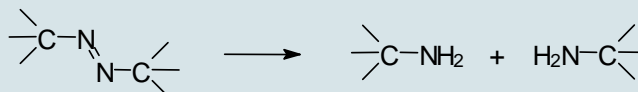
Amine decomposition



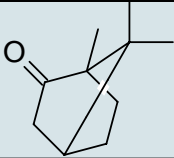
ω -Oxidation

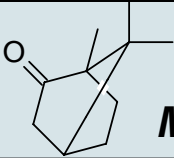

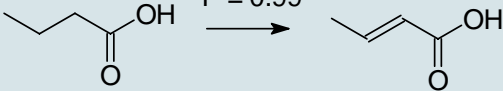
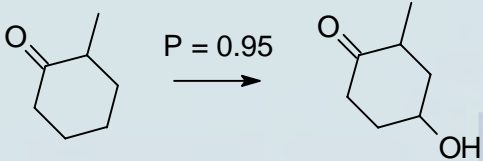
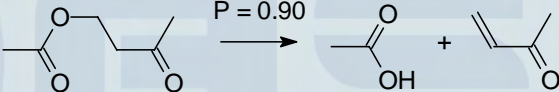
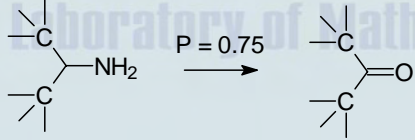
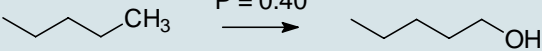
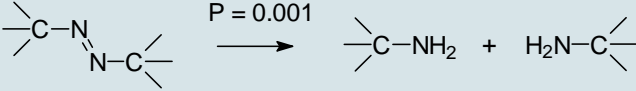


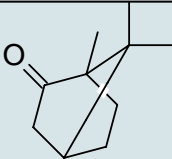
Azo-bond cleavage

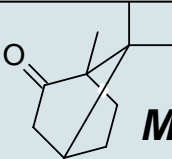


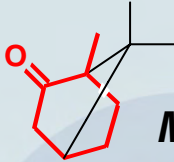
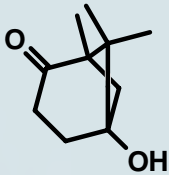
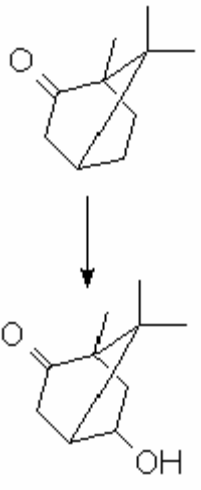
Substrate	Principal transformations	Metabolites
	Geminal diol decomposition	
	$\begin{array}{c} \diagup \text{OH} \\ \\ \text{C} \\ \\ \diagdown \text{OH} \end{array} \xrightarrow{\text{P} = 1.00} \begin{array}{c} \diagup \\ \\ \text{C}=\text{O} \\ \\ \diagdown \end{array}$	
	β -oxidation	
	$\text{CH}_3\text{CH}_2\text{CH}_2\text{C}(=\text{O})\text{OH} \xrightarrow{\text{P} = 0.99} \text{CH}_3\text{CH}=\text{CHC}(=\text{O})\text{OH}$	
	Cyclohexanone oxidation	
	$\text{Cyclohexanone} \xrightarrow{\text{P} = 0.95} \text{Cyclohexanecarboxylic acid}$	
	Ester hydrolysis	
	$\text{CH}_3\text{COOCH}_2\text{CH}_2\text{C}(=\text{O})\text{CH}_3 \xrightarrow{\text{P} = 0.90} \text{CH}_3\text{COOH} + \text{CH}_2=\text{CHC}(=\text{O})\text{CH}_3$	
	Amine decomposition	
	$\begin{array}{c} \diagup \\ \\ \text{C} \\ \\ \text{C}-\text{NH}_2 \\ \\ \text{C} \\ \\ \diagdown \end{array} \xrightarrow{\text{P} = 0.75} \begin{array}{c} \diagup \\ \\ \text{C} \\ \\ \text{C}=\text{O} \\ \\ \text{C} \\ \\ \diagdown \end{array}$	
	ω -Oxidation	
	$\text{CH}_3(\text{CH}_2)_4\text{CH}_3 \xrightarrow{\text{P} = 0.40} \text{CH}_3(\text{CH}_2)_4\text{OH}$	
	Azo-bond cleavage	
	$\begin{array}{c} \diagup \\ \\ \text{C}-\text{N} \\ \quad \diagdown \\ \text{N}-\text{C} \\ \quad \diagup \end{array} \xrightarrow{\text{P} = 0.001} \begin{array}{c} \diagup \\ \\ \text{C}-\text{NH}_2 \end{array} + \text{H}_2\text{N}-\begin{array}{c} \diagup \\ \\ \text{C} \end{array}$	

Substrate	Principal transformations	Metabolites
	Geminal diol decomposition $\begin{array}{c} \diagup \text{OH} \\ \\ \text{C} \\ \\ \diagdown \text{OH} \end{array} \xrightarrow{P = 1.00} \begin{array}{c} \diagup \\ \\ \text{C}=\text{O} \\ \\ \diagdown \end{array}$	
	β -oxidation $\text{CH}_3\text{CH}_2\text{CH}_2\text{COOH} \xrightarrow{P = 0.99} \text{CH}_3\text{CH}=\text{CHCOOH}$	
	Cyclohexanone oxidation $\text{Cyclohexanone} \xrightarrow{P = 0.95} \text{4-Hydroxycyclohexanone}$	
	Ester hydrolysis $\text{CH}_3\text{COOCH}_2\text{CH}_2\text{COCH}_3 \xrightarrow{P = 0.90} \text{CH}_3\text{COOH} + \text{CH}_3\text{CH}=\text{CHCOCH}_3$	
	Amine decomposition $\begin{array}{c} \diagup \\ \\ \text{C} \\ \\ \text{C}-\text{NH}_2 \\ \\ \text{C} \\ \\ \diagdown \end{array} \xrightarrow{P = 0.75} \begin{array}{c} \diagup \\ \\ \text{C} \\ \\ \text{C}=\text{O} \\ \\ \text{C} \\ \\ \diagdown \end{array}$	
	ω -Oxidation $\text{CH}_3(\text{CH}_2)_4\text{CH}_3 \xrightarrow{P = 0.40} \text{CH}_3(\text{CH}_2)_4\text{OH}$	
	Azo-bond cleavage $\begin{array}{c} \diagup \\ \\ \text{C}-\text{N} \\ \quad \diagdown \\ \text{N}-\text{C} \\ \quad \diagup \end{array} \xrightarrow{P = 0.001} \begin{array}{c} \diagup \\ \\ \text{C}-\text{NH}_2 \end{array} + \text{H}_2\text{N}-\begin{array}{c} \diagup \\ \\ \text{C} \end{array}$	

Substrate	Principal transformations	Metabolites
	<p>Geminal diol decomposition</p> <p>$P = 1.00$</p> <p>Match? - No!</p>	
	<p>β-oxidation</p> <p>$P = 0.99$</p>	
	<p>Cyclohexanone oxidation</p> <p>$P = 0.95$</p>	
	<p>Ester hydrolysis</p> <p>$P = 0.90$</p>	
	<p>Amine decomposition</p> <p>$P = 0.75$</p>	
	<p>ω-Oxidation</p> <p>$P = 0.40$</p>	
	<p>Azo-bond cleavage</p> <p>$P = 0.001$</p>	

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	$\begin{array}{c} \diagup \text{OH} \\ \\ \text{C} \\ \\ \diagdown \text{OH} \end{array} \xrightarrow{P = 1.00} \begin{array}{c} \diagup \\ \\ \text{C}=\text{O} \\ \\ \diagdown \end{array}$	
	β-oxidation	
	$\text{CH}_3\text{CH}_2\text{CH}_2\text{COOH} \xrightarrow{P = 0.99} \text{CH}_3\text{CH}=\text{CHCOOH}$	
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	$\text{Cyclohexanone} \xrightarrow{P = 0.95} \text{Cyclohexanecarboxylic acid}$	
	Ester hydrolysis	
	$\text{CH}_3\text{COOCH}_2\text{CH}_2\text{COCH}_3 \xrightarrow{P = 0.90} \text{CH}_3\text{COOH} + \text{CH}_3\text{COCH}=\text{CH}_2$	
	Amine decomposition	
	$\begin{array}{c} \diagup \\ \\ \text{C} \\ \\ \text{C}-\text{NH}_2 \\ \\ \diagdown \end{array} \xrightarrow{P = 0.75} \begin{array}{c} \diagup \\ \\ \text{C} \\ \\ \text{C}=\text{O} \\ \\ \diagdown \end{array}$	
	ω-Oxidation	
	$\text{CH}_3(\text{CH}_2)_4\text{CH}_3 \xrightarrow{P = 0.40} \text{CH}_3(\text{CH}_2)_4\text{OH}$	
	Azo-bond cleavage	
	$\begin{array}{c} \diagup \\ \\ \text{C}-\text{N} \\ \quad \diagdown \\ \text{N}-\text{C} \\ \quad \diagup \end{array} \xrightarrow{P = 0.001} \begin{array}{c} \diagup \\ \\ \text{C}-\text{NH}_2 \\ \end{array} + \text{H}_2\text{N}-\begin{array}{c} \diagup \\ \\ \text{C} \\ \\ \diagdown \end{array}$	

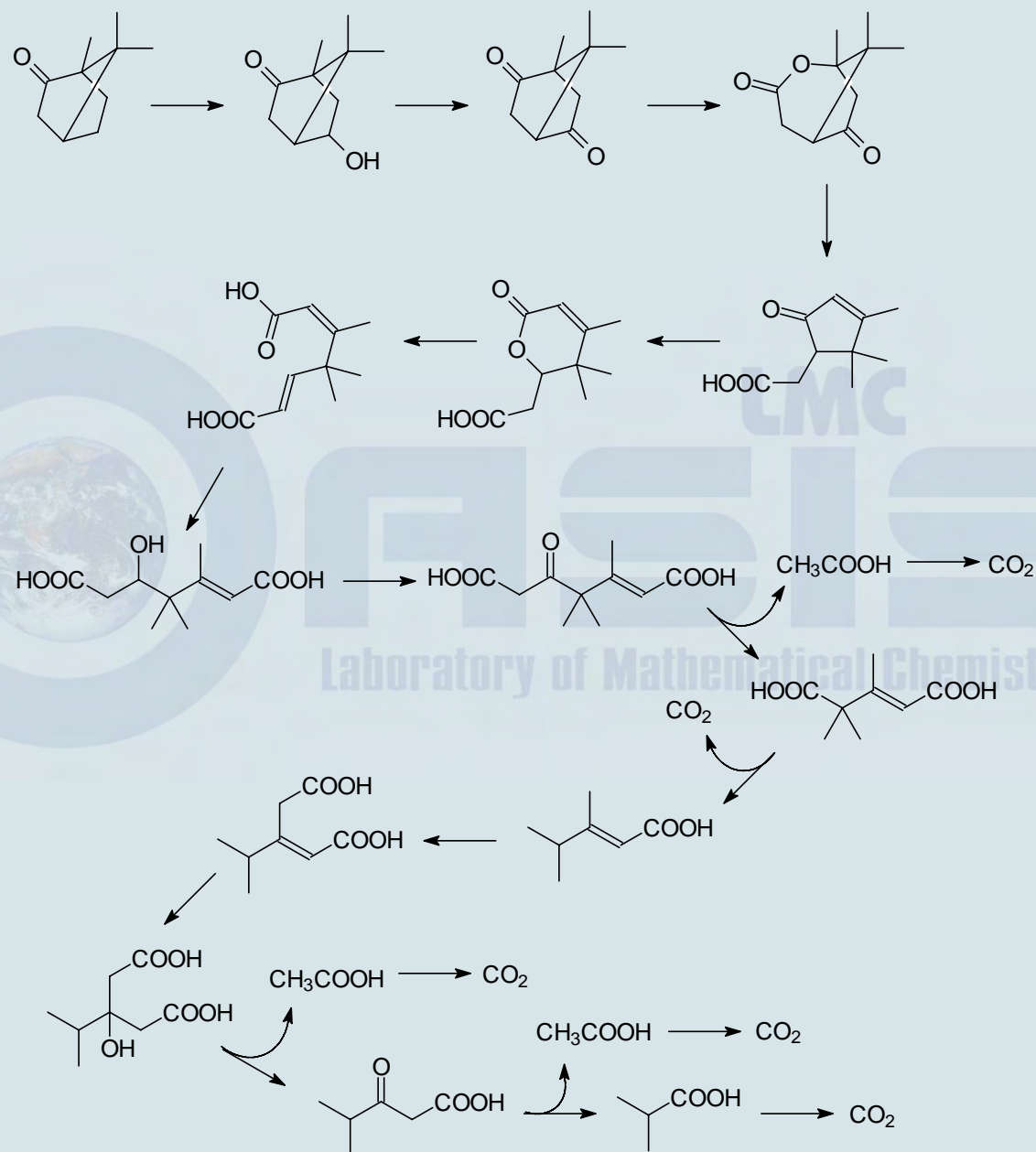
Substrate	Principal transformations	Metabolites
	Geminal diol decomposition $\begin{array}{c} \diagup \text{OH} \\ \\ \text{C} \\ \\ \diagdown \text{OH} \end{array} \xrightarrow{P = 1.00} \begin{array}{c} \diagup \\ \\ \text{C}=\text{O} \\ \\ \diagdown \end{array}$	
	β -oxidation $\text{CH}_3\text{CH}_2\text{CH}_2\text{COOH} \xrightarrow{P = 0.99} \text{CH}_3\text{CH}=\text{CHCOOH}$	
	Cyclohexanone oxidation $\text{Cyclohexanone} \xrightarrow{P = 0.95} \text{Cyclohexanol}$	
	Ester hydrolysis $\text{CH}_3\text{COOCH}_2\text{CH}_2\text{COCH}_3 \xrightarrow{P = 0.90} \text{CH}_3\text{COOH} + \text{CH}_3\text{CH}=\text{CHCOCH}_3$	
	Amine decomposition $\begin{array}{c} \\ \text{C} \\ \quad \\ \text{C} \quad \text{NH}_2 \\ \quad \\ \text{C} \end{array} \xrightarrow{P = 0.75} \begin{array}{c} \\ \text{C} \\ \quad \\ \text{C} \quad \text{C}=\text{O} \\ \quad \\ \text{C} \end{array}$	
	ω -Oxidation $\text{CH}_3(\text{CH}_2)_4\text{CH}_3 \xrightarrow{P = 0.40} \text{CH}_3(\text{CH}_2)_4\text{OH}$	
	Azo-bond cleavage $\begin{array}{c} \\ \text{C}-\text{N} \\ \quad \\ \text{N}-\text{C} \\ \end{array} \xrightarrow{P = 0.001} \begin{array}{c} \\ \text{C}-\text{NH}_2 \\ \end{array} + \text{H}_2\text{N}-\begin{array}{c} \\ \text{C} \\ \end{array}$	

Substrate	Principal transformations	Metabolites
	Geminal diol decomposition $\begin{array}{c} \diagup \text{OH} \\ \\ \text{C} \\ \\ \diagdown \text{OH} \end{array} \xrightarrow{P = 1.00} \begin{array}{c} \diagup \\ \\ \text{C}=\text{O} \\ \\ \diagdown \end{array}$	
	β -oxidation $\text{CH}_3\text{CH}_2\text{CH}_2\text{COOH} \xrightarrow{P = 0.99} \text{CH}_3\text{CH}=\text{CHCOOH}$	
	Cyclohexanone oxidation $\text{Cyclohexanone} \xrightarrow{P = 0.95} \text{Cyclohexanol}$	
Match? - Yes!	RESULT	
	Ester hydrolysis $\text{CH}_3\text{COOCH}_2\text{CH}_2\text{COCH}_3 \xrightarrow{P = 0.90} \text{CH}_3\text{COOH} + \text{CH}_2=\text{CHCOCH}_3$	
	Amine decomposition $\begin{array}{c} \diagup \\ \\ \text{C}-\text{NH}_2 \\ \\ \text{C} \end{array} \xrightarrow{P = 0.75} \begin{array}{c} \diagup \\ \\ \text{C}=\text{O} \\ \\ \text{C} \end{array}$	
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	Azo-bond cleavage $\begin{array}{c} \diagup \\ \\ \text{C}-\text{N}=\text{N}-\text{C} \\ \\ \diagdown \end{array} \xrightarrow{P = 0.001} \begin{array}{c} \diagup \\ \\ \text{C}-\text{NH}_2 \\ \\ \diagdown \end{array} + \text{H}_2\text{N}-\begin{array}{c} \diagup \\ \\ \text{C} \\ \\ \diagdown \end{array}$	

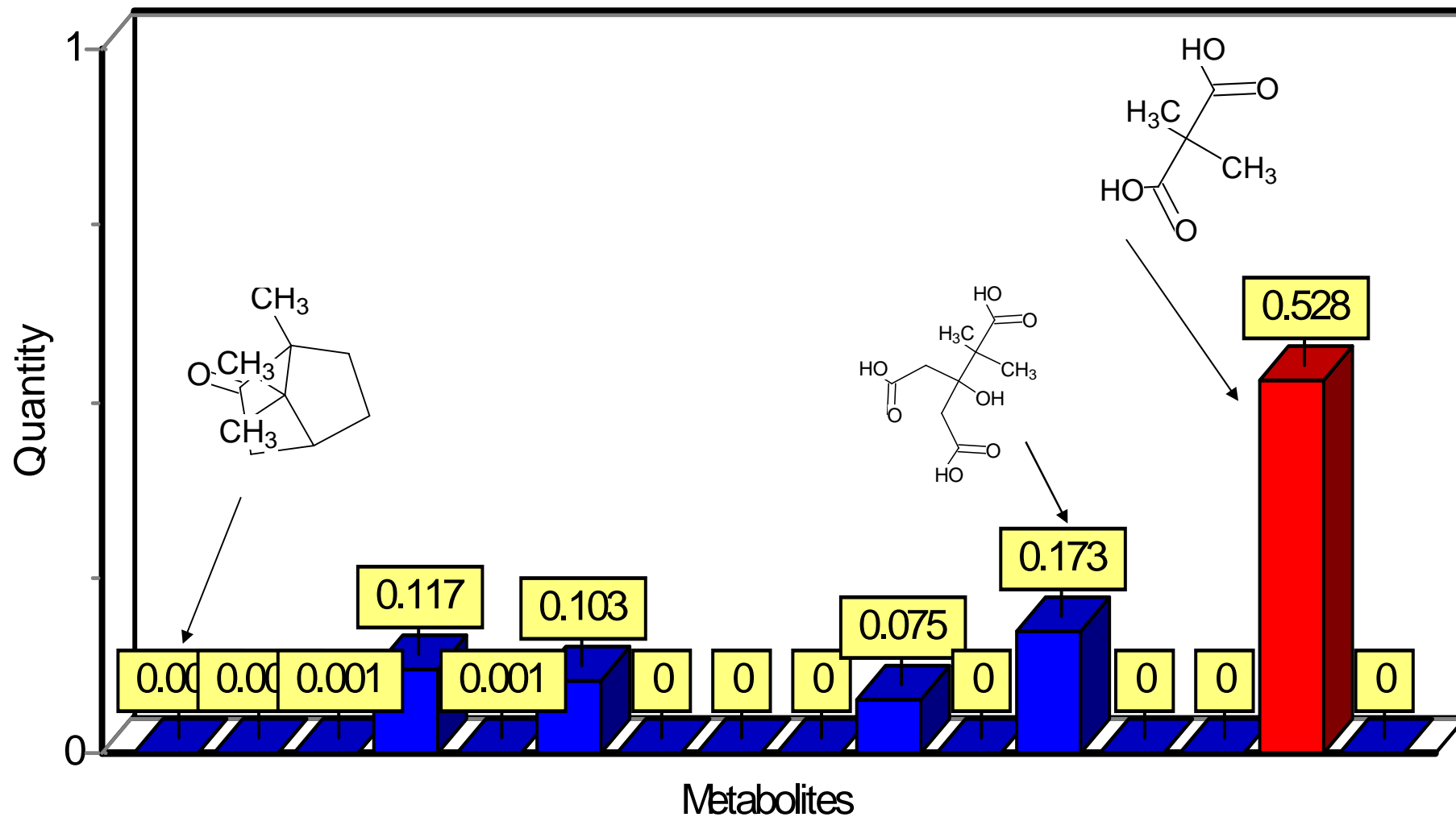
Outlook

- **QSAR and Complexity of Chemical Structure**
- **Toxicity as a result of metabolic activation**
- **Metabolism logic**
- **Probabilistic approach to modeling metabolism**
- **CATABOL for simulating microbial degradation**
- **Performance and reliability of predicted metabolites**
- **Biodegradation kinetic models**
- **Simulating the effect of gene modification on metabolism**

Predicted biotransformation pathway for camphor



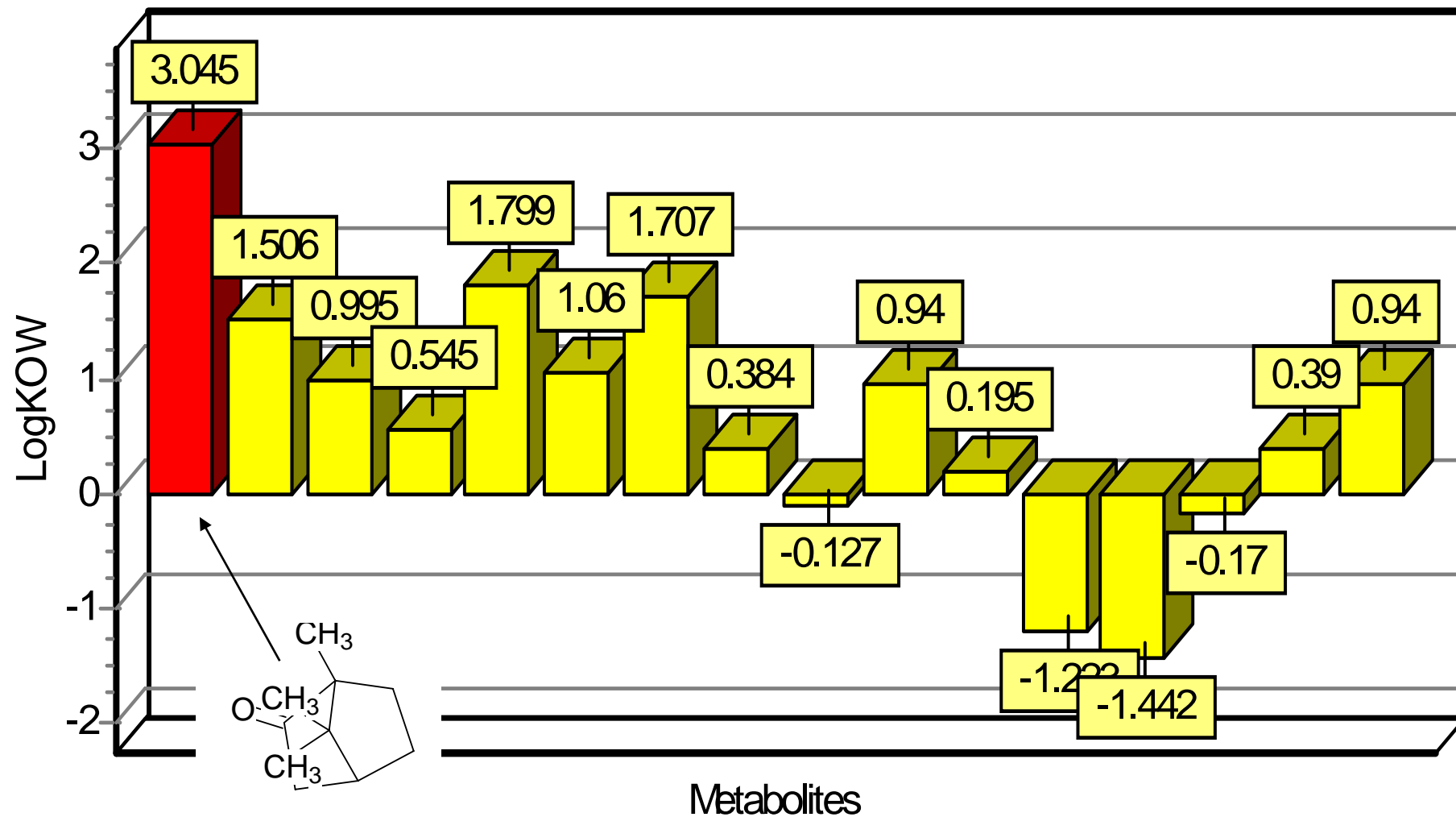
Metabolites distribution



C=C(O)(O)C(C)(C)C(=O)O; Q=0.528212; Pb=0.529; Ps=0.528; logKOW=0.390

Transformation 394 Level 13

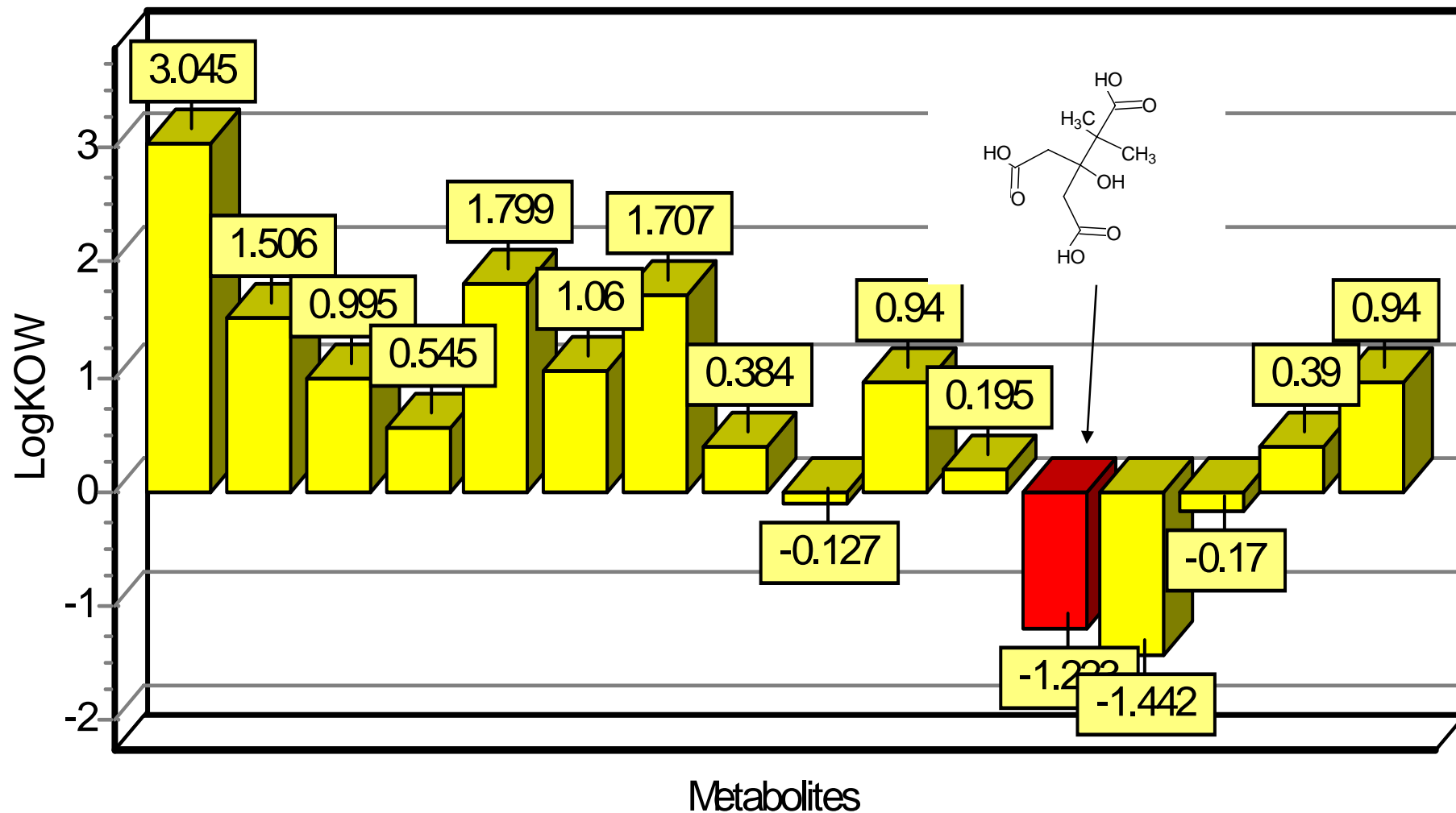
Metabolites distribution



C1(=O)C2(C)C(C)(C)C(OC2)C1; Q=0.001000; Pb=1.000; Ps=0.001; logKOW=3.045

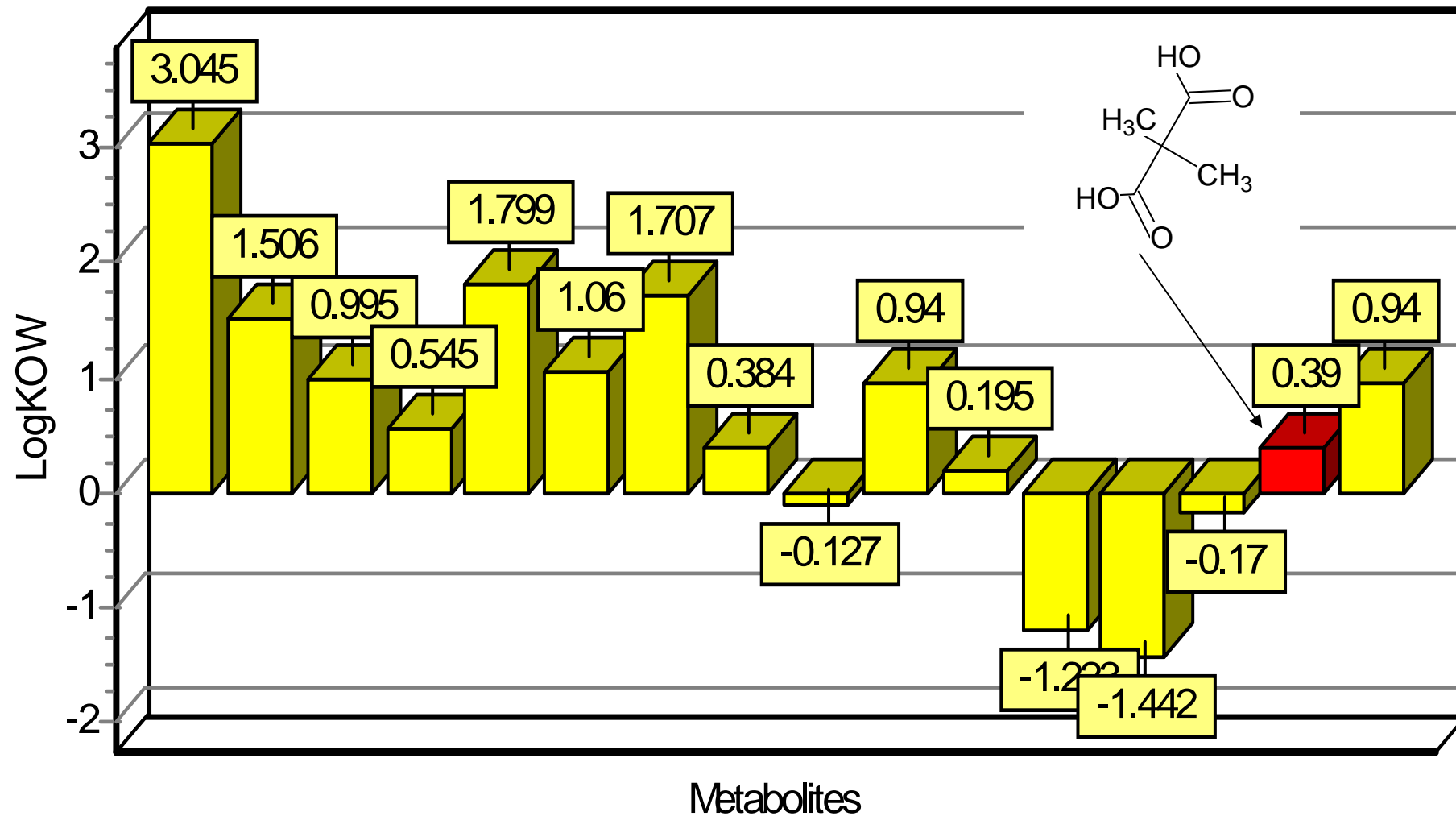
Transformation 164 Level 0

Metabolites distribution



C(=O)OCC(C)C(=O)OCC(=O)OCC(=O)O; Q=0.172507; P₀=0.701; P_S=0.173; logKOW=-1.223
Transformation 281 Level 11

Metabolites distribution

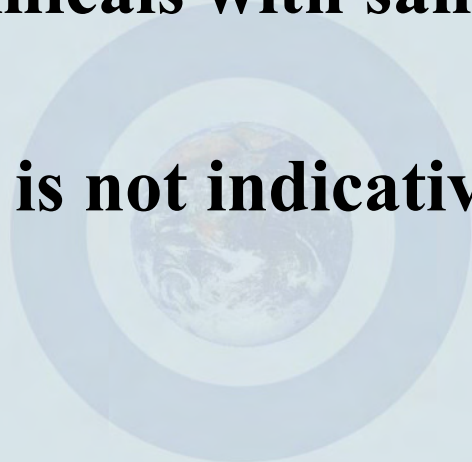


C(=O)O(C)C(C)C(=O)O; Q=0.528212; Pb=0.529; Ps=0.528; logKOW=0.390

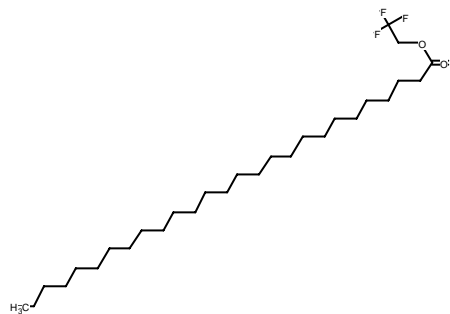
Transformation 394 Level 13

Persistency Evaluation Problems Solved by CATABOL

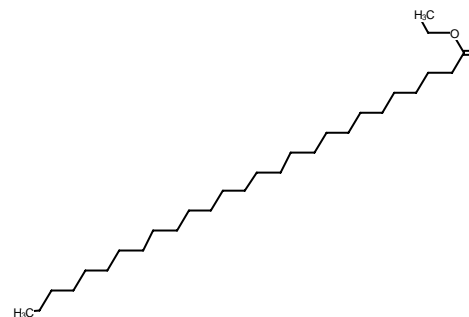
- Chemicals with same BOD could have different fate
- BOD is not indicative for obtaining stable degradants



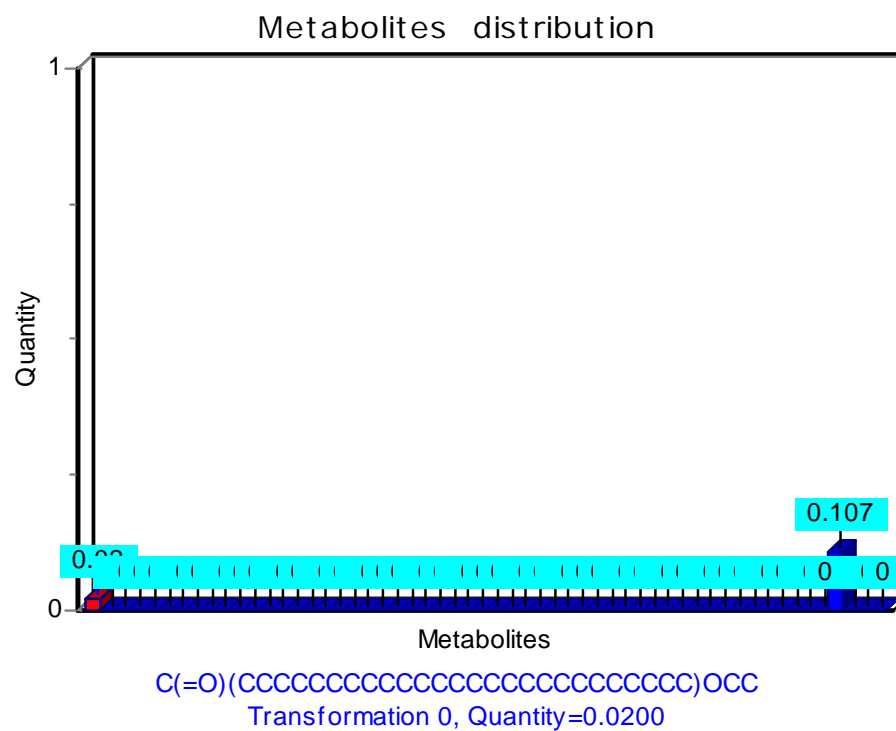
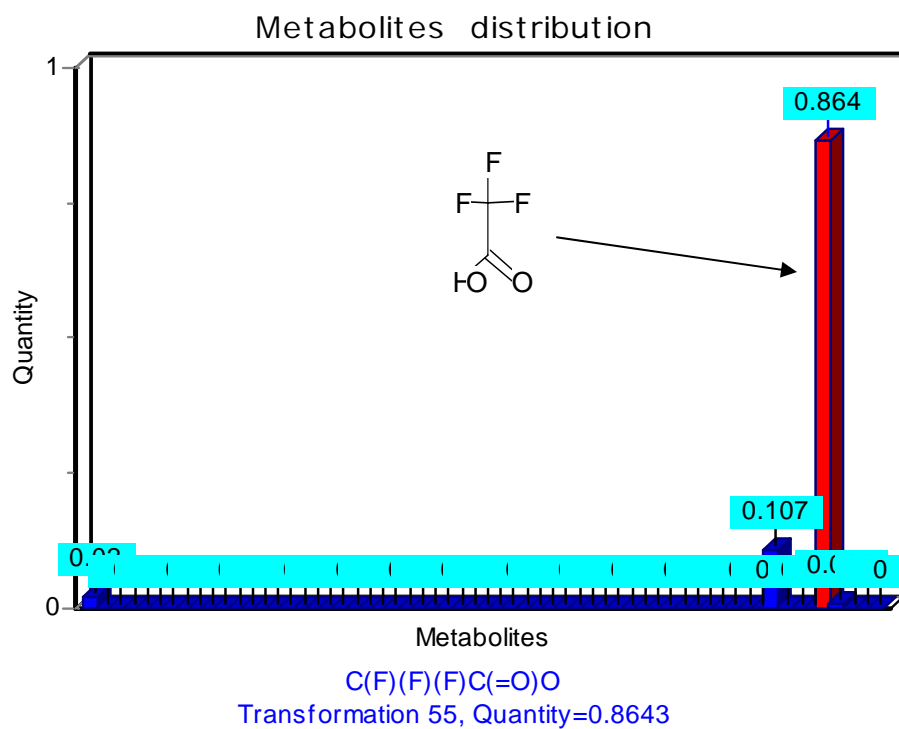
LMC
MATHSIS
Laboratory of Mathematical Chemistry



BOD = 95%



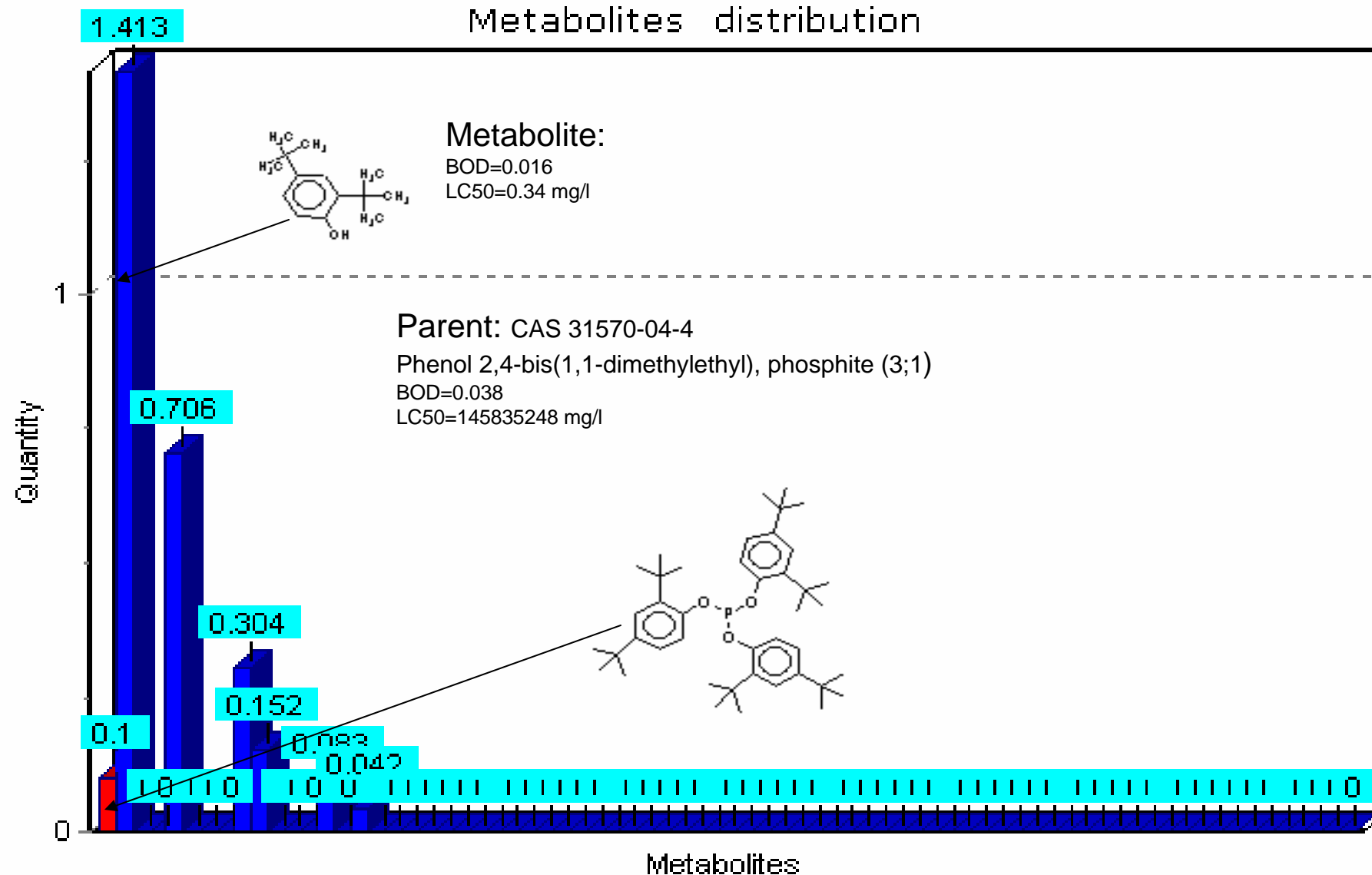
BOD = 97%



Persistency Evaluation Problems Solved by CATABOL

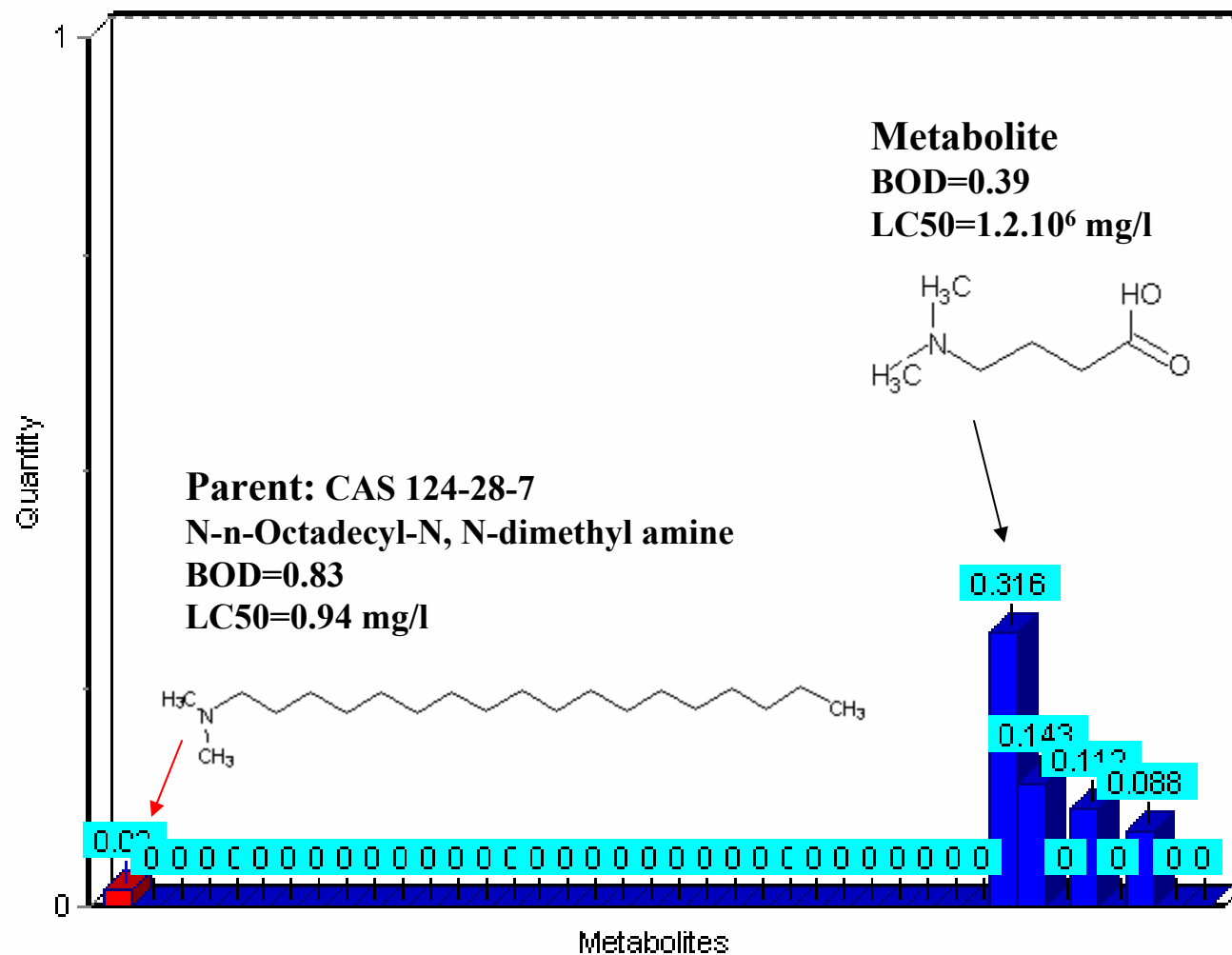
- Chemicals with same BOD could have different fate
- BOD is not indicative for obtaining stable degradants
- Toxicity of chemicals could be due to the stable degradants

Metabolites distribution



C(C)(C)(C)c1c(OP(Oc2c(C(C)(C)C)cc(C(C)(C)C)cc2)Oc2c(C(C)(C)C)cc(C(C)(C)C)cc2)ccc(C(C)(C)C)c1
Transformation 0, Quantity=0.1000

Metabolites distribution



C(CCCCCCCCCCCCCCCCCCCC)N(C)C
 Transformation 0, Quantity=0.0200

Stable degradants in chemical legislation

Laboratory of Mathematical Chemistry

Japanese NITE

Chemical Substances Control Law (CSCL)



The Features of the BCF Data Set under CSCL

Relationship between Biodegradation Test and Bioconcentration Test

Biodegradaton Test			Decision	Next Test
Method	Test Result			
Biodegradati on Test (OECD 301C)	1	Parent Residual > 40% and Metabolite Residual < 1%	Not Readily Biodegradable	Bioconcentraiotn for Parent
	2	Parent Residual ≥ 1% and Metabolite Residual ≥ 1%		Bioconcentraiotn for Parent and Metabolite(s) (≥1%)
	3	Parent Residual < 1% and Metabolite Residual ≥ 1%		Bioconcentraiotn for Metabolite(s) (≥1%)
	4	Parent Residual ≤ 40% and Metabolite Residual < 1%	Readily Biodegradable	Nothing (Reguration Free)

Bioconcentration Test

OECD305C: The test fish (carp) is exposed to two concentrations of the test chemical substance in water under flow-through conditions.

All tests are conducted by GLP laboratories and their test results are reviewed by the joint council.

The Features of the BCF Data Set under CSCL

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	2	Parent Residual ≥ 1% and Metabolite Residual ≥ 1%		Bioconcentraiotn for Parent and Metabolite(s) (≥1%)
	3	Parent Residual < 1% and Metabolite Residual ≥ 1%		Bioconcentraiotn for Metabolite(s) (≥1%)
	4	Parent Residual ≤ 40% and Metabolite Residual < 1%	Readily Biodegradable	Nothing (Reguration Free)

Bioconcentration Test

OECD305C: The test fish (carp) is exposed to two concentrations of the test chemical substance in water under flow-through conditions.

All tests are conducted by GLP laboratories and their test results are reviewed by the joint council.

CATABOL - mathematical formalism

BOD or CO₂ production

Modeled

$$BOD = \sum_n \Delta_n^{O_2} \prod_{m=1}^n P_m$$

$$CO_2 = \sum_n \Delta_n^{CO_2} \prod_{m=1}^n P_m$$

Theoretical

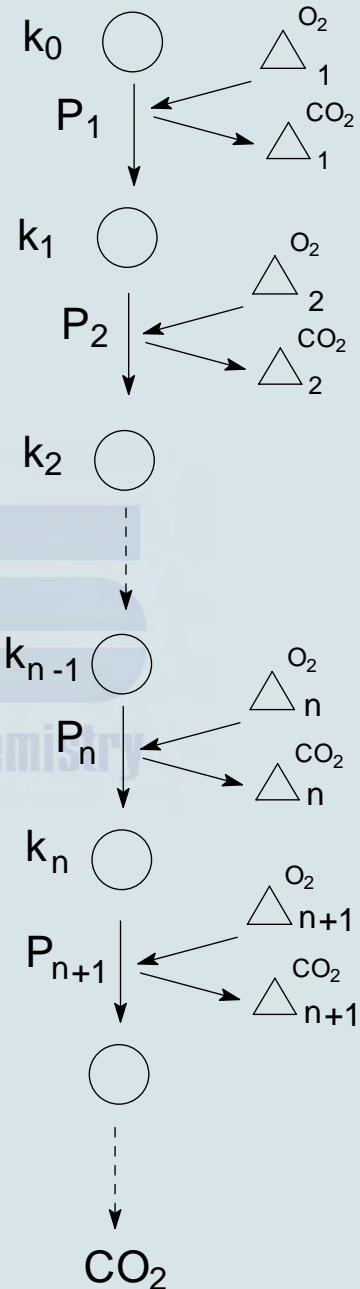
$$BOD^{Th} = \sum_n \Delta_n^{O_2}$$

$$CO_2^{Th} = \sum_n \Delta_n^{CO_2}$$

Biodegradability - % of theoretical

$$BOD^{Calc} = \frac{\sum_n \Delta_n^{O_2} \prod_{m=1}^n P_m}{\sum_n \Delta_n^{O_2}} 100, \%$$

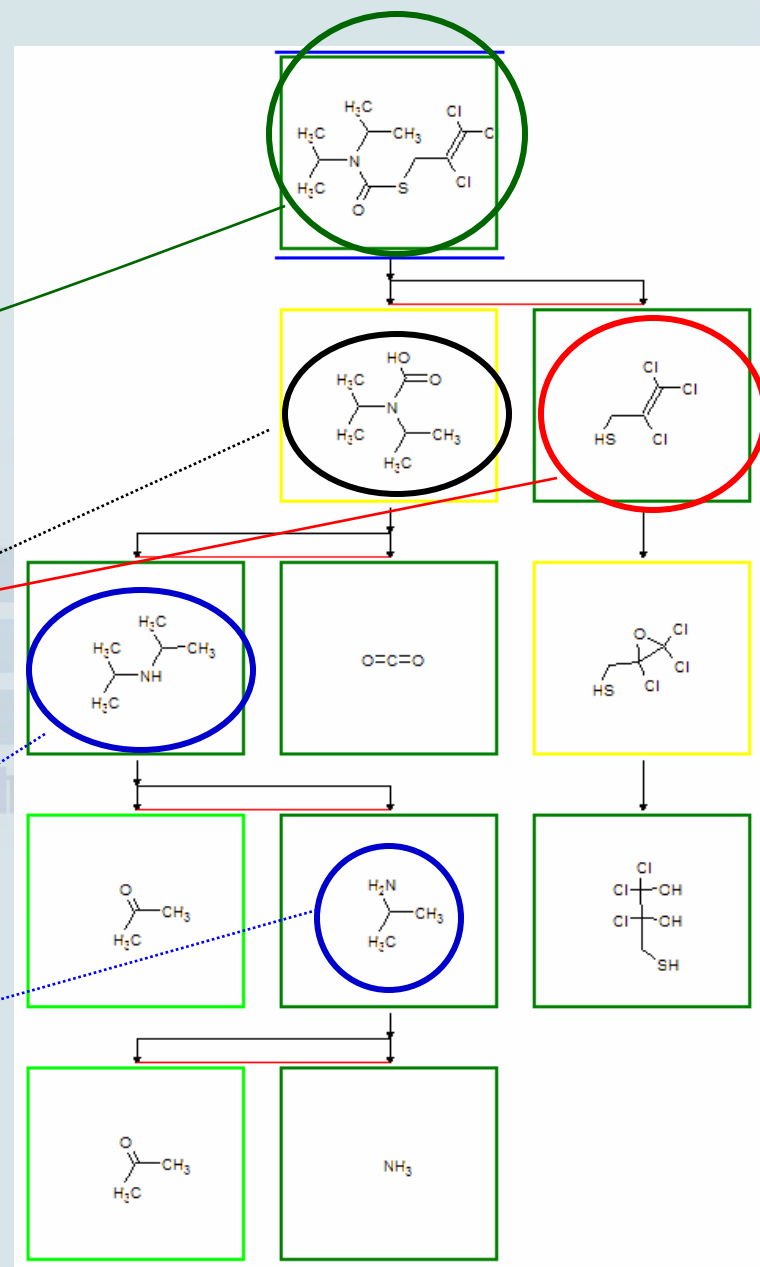
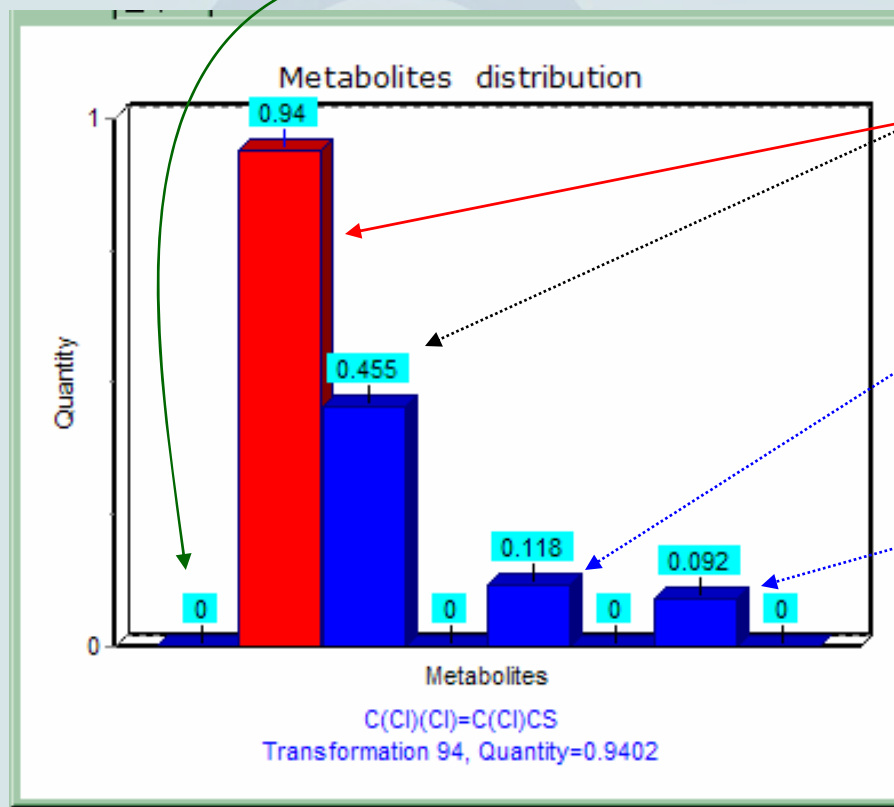
$$ThCO_2^{Calc} = \frac{\sum_n \Delta_n^{CO_2} \prod_{m=1}^n P_m}{\sum_n \Delta_n^{O_2}} 100, \%$$



CATABOL - mathematical formalism

Quantities of metabolites

$$Q_n^{Calc} = (1 - P_{n+1}) \prod_{m=1 \rightarrow n} P_m, \text{ mol/mol parent}$$



CATABOL - mathematical formalism

First order kinetics

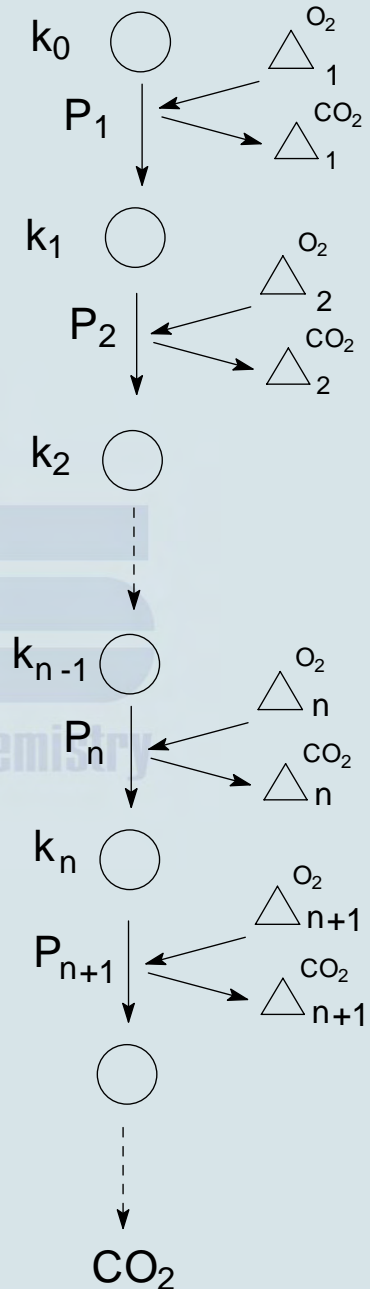
$$BOD = 100(1 - \exp(-kt))$$

$$t_{1/2} = \ln(2)/k$$

$$k = -\ln(1 - BOD_{28-d}^{Calc} / 100) / 28$$

Ultimate half-life

$$t_{1/2} = \frac{\ln(2)}{-\ln(1 - BOD_{28-d}^{Calc} / 100) / 28}$$



CATABOL - mathematical formalism

First order kinetics

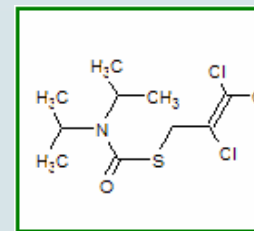
$$BOD = 100(1 - \exp(-kt))$$

$$t_{1/2} = \ln(2)/k$$

$$k = -\ln(1 - BOD_{28-d}^{Calc} / 100) / 28$$

Ultimate half-life

$$t_{1/2} = \frac{\ln(2)}{-\ln(1 - BOD_{28-d}^{Calc} / 100) / 28}$$



$$BOD^{Calc} = 31\%$$

$$k = 0.013 \text{ day}^{-1}$$

$$t_{1/2} = 52 \text{ days}$$

CATABOL – training data

Observed BOD – Estimation of transformation probabilities

MITI I test (OECD 301 C)

745 training chemicals

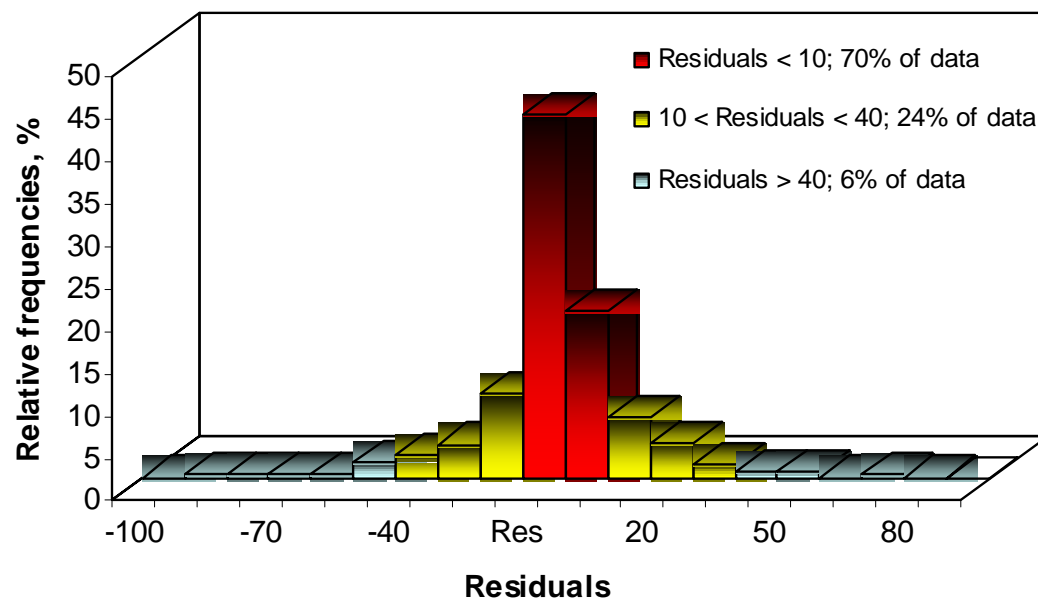
347 external validation chemicals

$$\min_P RSS = \sum_{n=1}^N \left(BOD_n^{Obs.} - BOD_n^{Calc.} \right)^2$$

Coefficient of determination $R^2 = 0.69$

Sensitivity (ready biodegradable) - 86%

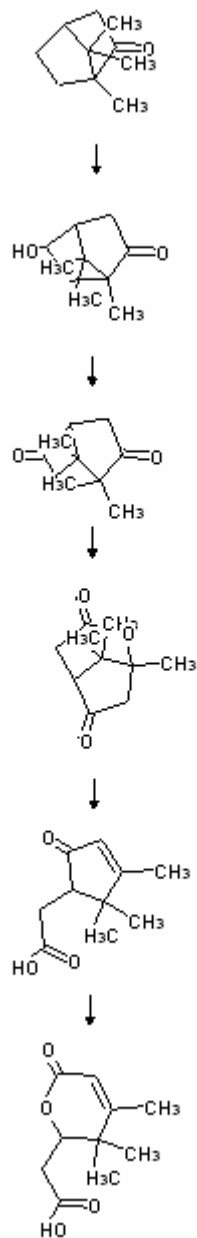
Specificity (not ready biodegradable) - 91%



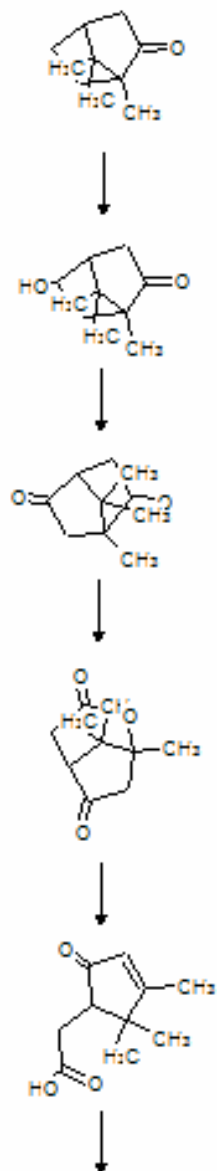
Outlook

- **QSAR and Complexity of Chemical Structure**
- **Toxicity as a result of metabolic activation**
- **Metabolism logic**
- **Probabilistic approach to modeling metabolism**
- **CATABOL for simulating microbial degradation**
- **Performance and reliability of predicted metabolites**
- **Biodegradation kinetic models**
- **Simulating the effect of gene modification on metabolism**

Observed



Predicted

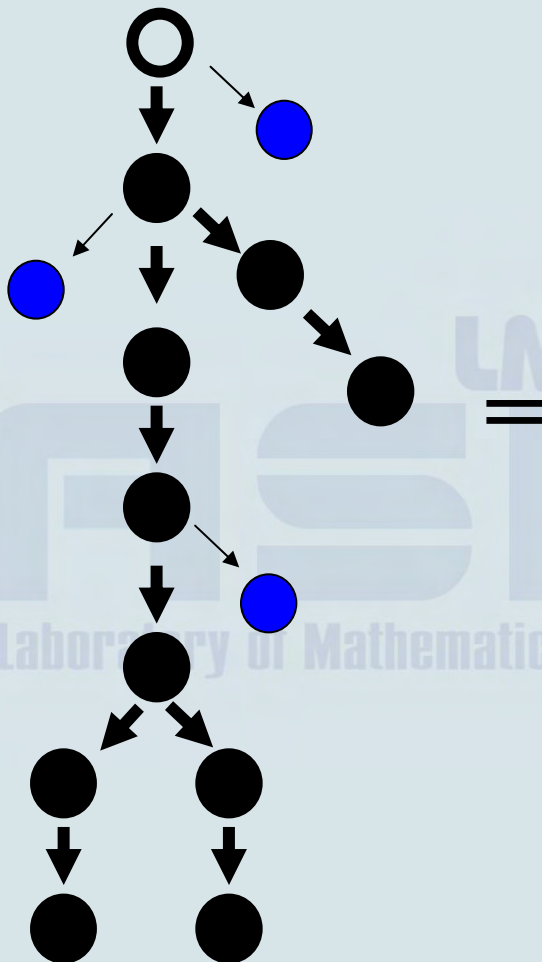
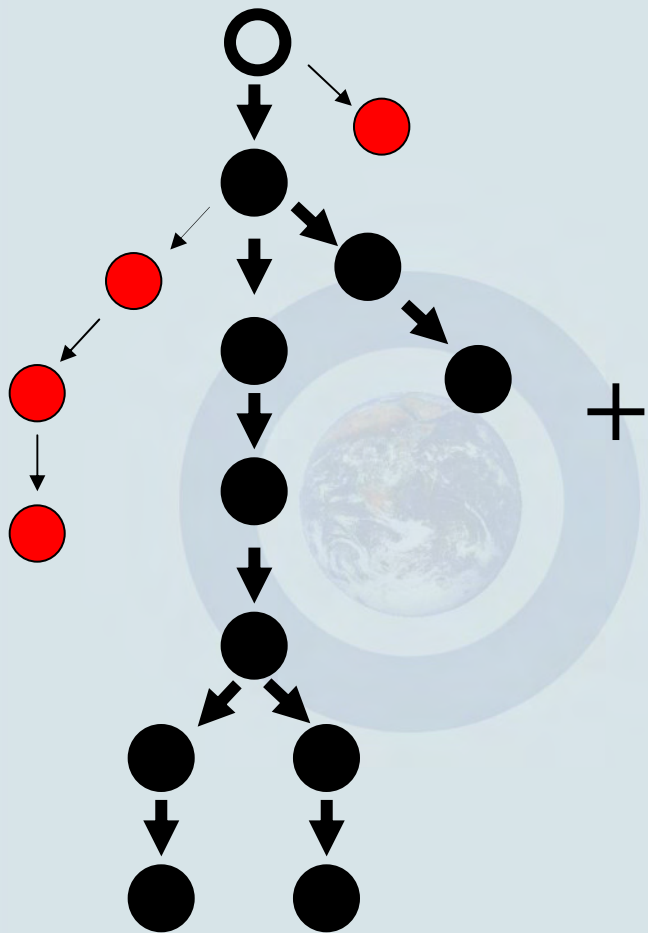


How good is reproducibility of the observed catabolism?

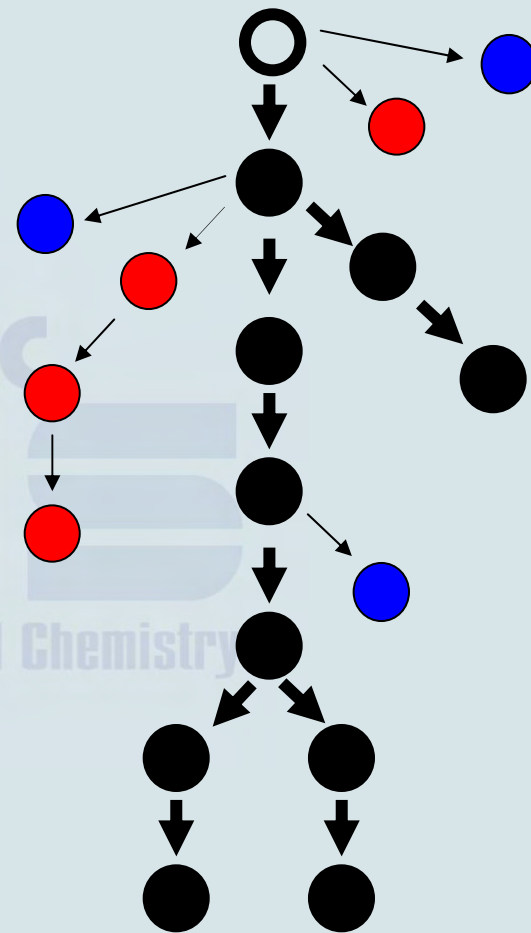
LMC
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Observed versus simulated pathways

Observed and predicted catabolism

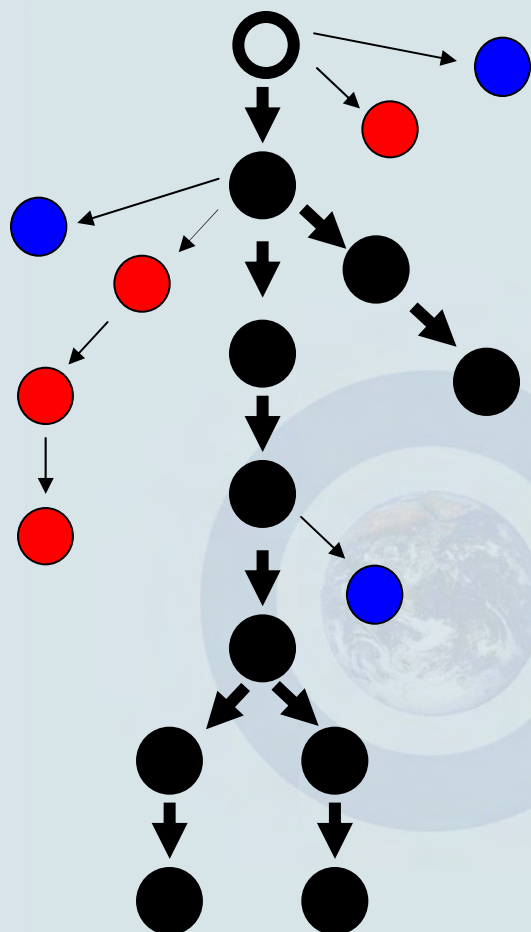


Union of pathways



- - Observed and predicted metabolites, $S_{\text{Obs}} \cap S_{\text{Pred}}$
- - Observed and not predicted metabolites, $S_{\text{Obs}} - S_{\text{Pred}}$ or $S_{\text{Obs}} \setminus S_{\text{Pred}}$
- - Predicted and not observed metabolites, $S_{\text{Pred}} - S_{\text{Obs}}$ or $S_{\text{Pred}} \setminus S_{\text{Obs}}$

Union of pathways



Probability that the metabolite is observed,
given that the metabolite is predicted (**predictability**)

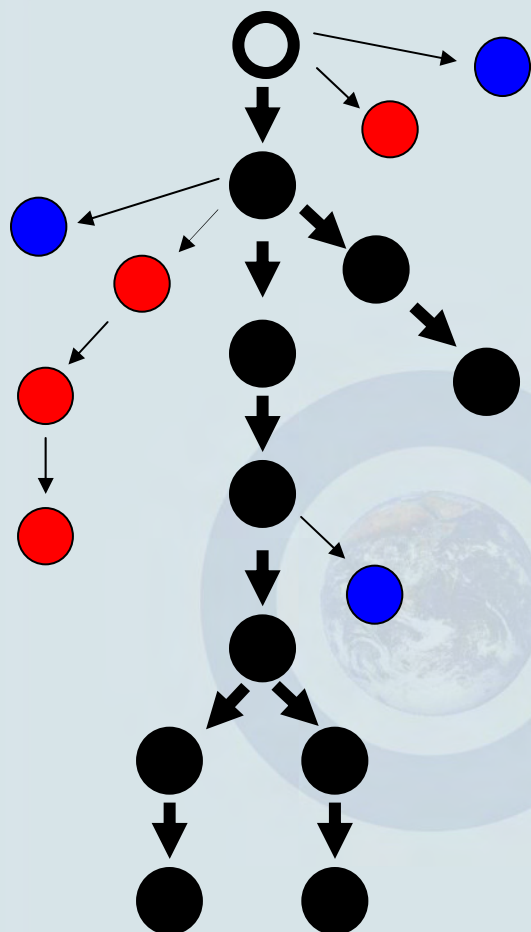
$$\frac{Card(S_{\text{Obs}} \cap S_{\text{Pred}})}{Card(S_{\text{Pred}})} = \frac{\text{●}}{\text{●} + \text{●}}$$

Probability that the metabolite is observed,
given that the metabolite is truly observed (**sensitivity**)

$$\frac{Card(S_{\text{Obs}} \cap S_{\text{Pred}})}{Card(S_{\text{Obs}})} = \frac{\text{●}}{\text{●} + \text{●}}$$

- - Observed and predicted metabolites, $S_{\text{Obs}} \cap S_{\text{Pred}}$
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- - Predicted and not observed metabolites, $S_{\text{Pred}} - S_{\text{Obs}}$ or $S_{\text{Pred}} \setminus S_{\text{Obs}}$

Union of pathways



Probability that the metabolite is not predicted
given that the metabolite is truly observed
(**false negatives**)

$$\frac{Card(S_{Obs} \setminus S_{Pred})}{Card(S_{Obs})} = \frac{\text{red circle}}{\text{black circle} + \text{red circle}}$$

Probability that the metabolite is predicted
given that the metabolite is truly not observed
(**false positives**)

$$\frac{Card(S_{Pred} \setminus S_{Obs})}{Card(S_{Pred})} = \frac{\text{blue circle}}{\text{black circle} + \text{blue circle}}$$

- - Observed and predicted metabolites, $S_{Obs} \cap S_{Pred}$
- - Observed and not predicted metabolites, $S_{Obs} - S_{Pred}$ or $S_{Obs} \setminus S_{Pred}$
- - Predicted and not observed metabolites, $S_{Pred} - S_{Obs}$ or $S_{Pred} \setminus S_{Obs}$

Reliability of generated metabolic pathway

$N_{i,succ}^{TR}$ - the numbers of successful applications of the transformation

$N_{i,fail}^{TR}$ - the numbers of unsuccessful applications of the transformation

Reliability of i -th transformation

$$R_i^{TR} = \frac{N_{i,succ}^{TR}}{N_{i,succ}^{TR} + N_{i,fail}^{TR}}$$

Reliability of predicted l -th metabolite

$$R_l^M = \prod_{j=1}^J R_j^{TR}$$

Reliability of k -th map

$$R_k^{Map} = \frac{\sum_{j=1}^K R_k^M}{K}$$

CATABOL – Applicability domain

Model: Bird or Fish?



Applicability
domain



Training data
External validation data
Modeled hypothesis
Statistics
Application of the model

Discrimination of
predictions

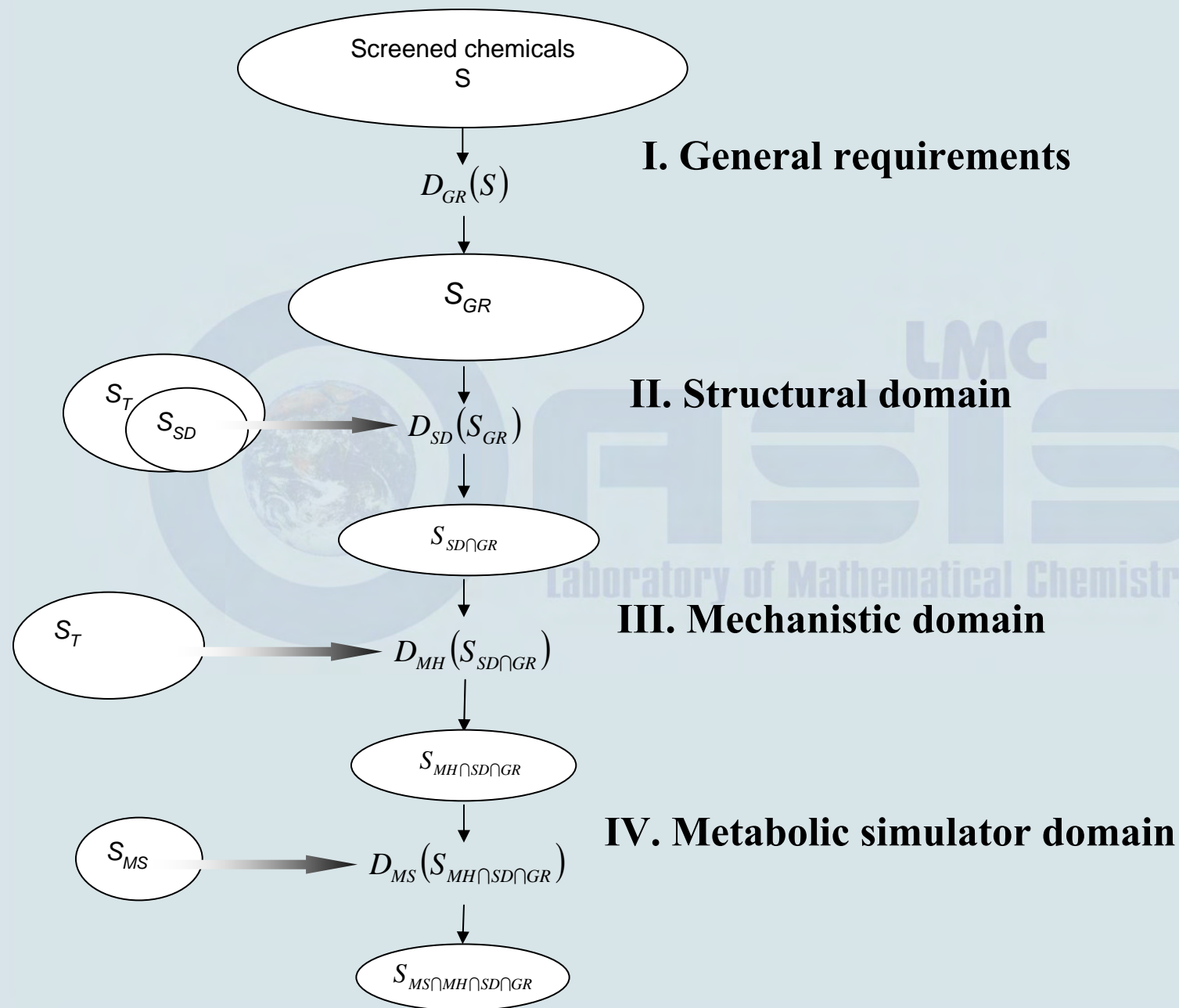


Out of domain

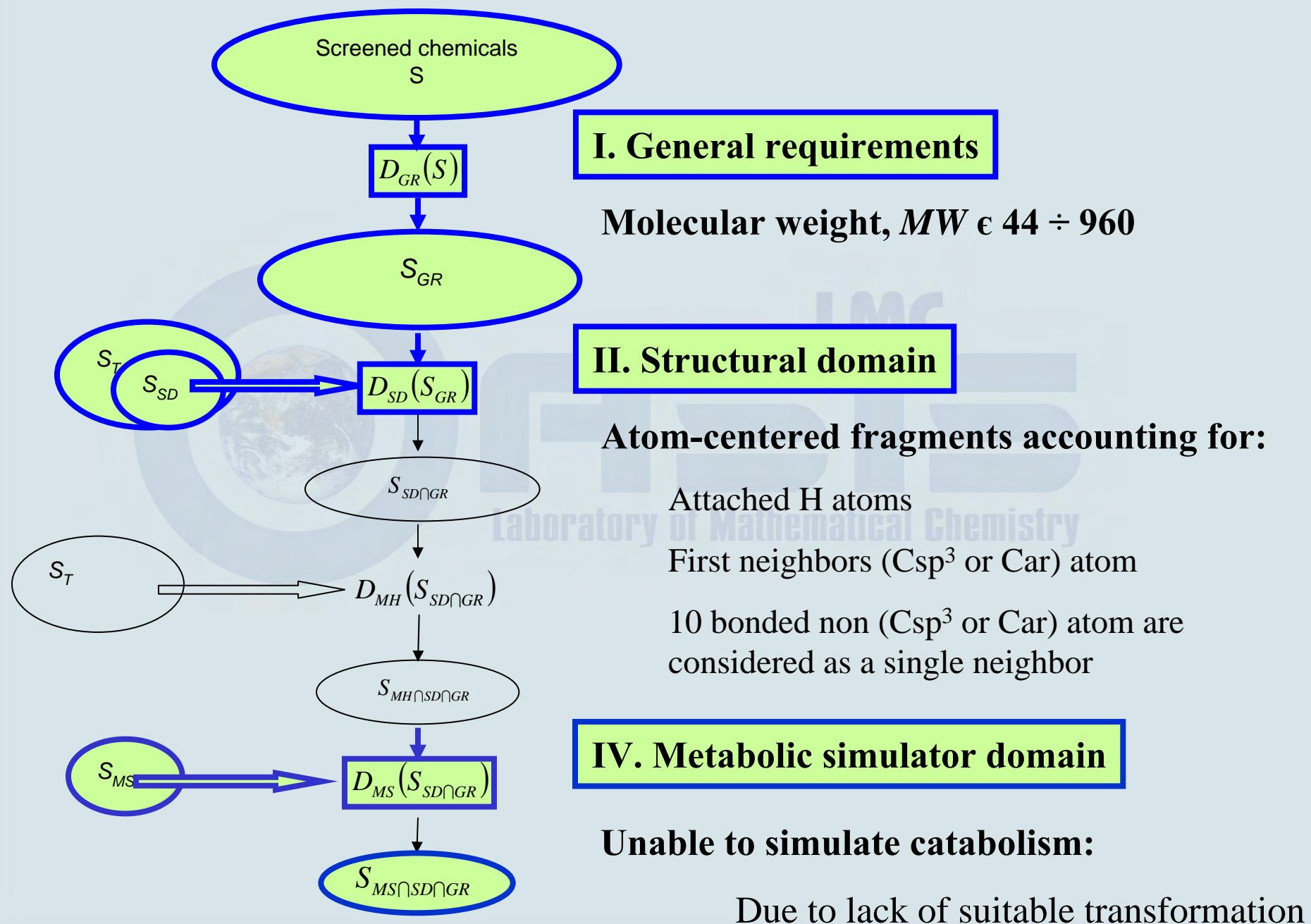
In domain



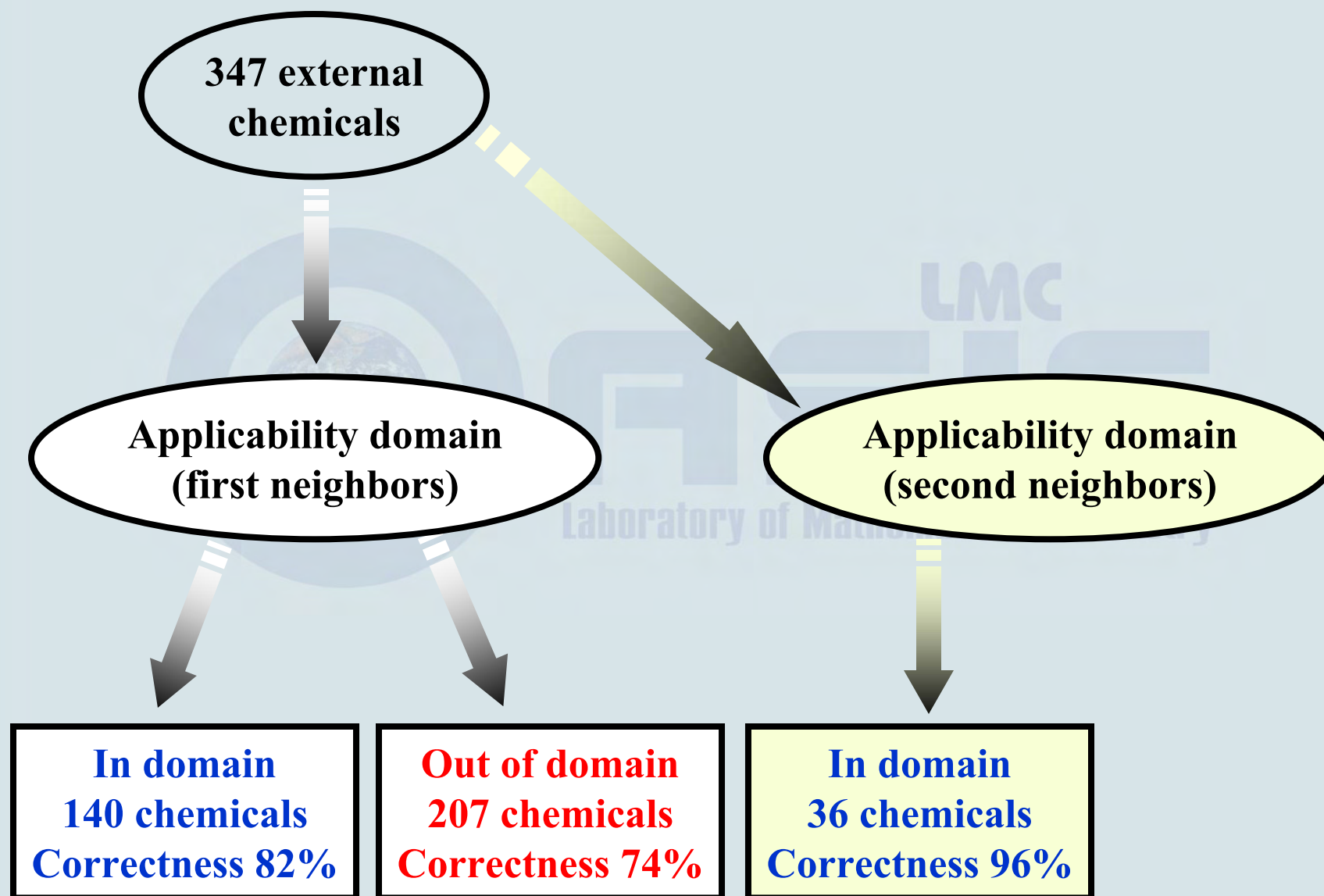
CATABOL – Applicability domain



CATABOL – Applicability domain

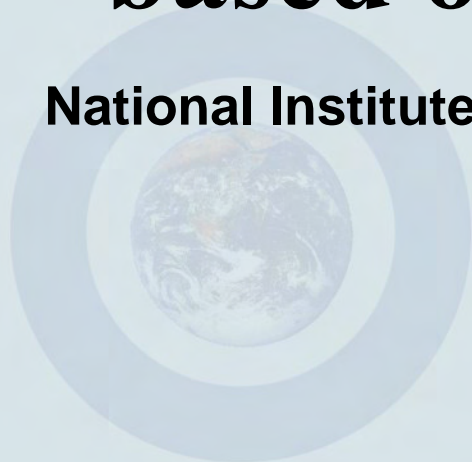


CATABOL – Applicability domain



External Validation of CATABOL based on OECD principle

**National Institute of Technology and Evaluation (NITE)
Japan, 2005**



Laboratory of Mathematical Chemistry

Used data for external validation

Test: OECD 301 C (MITI I)

Existing chemical: 338

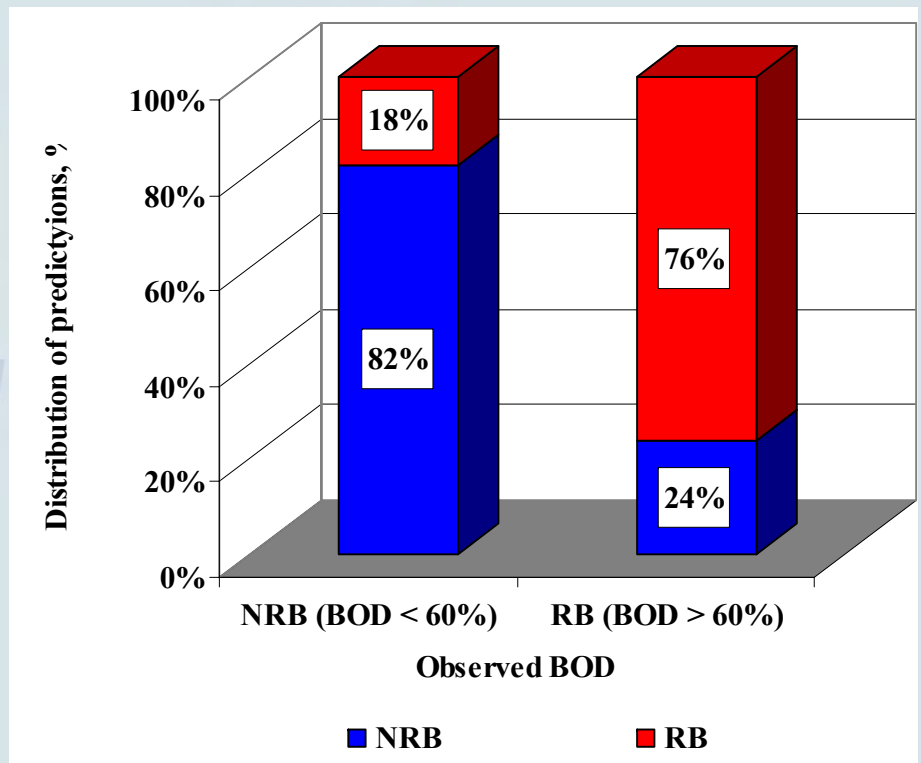
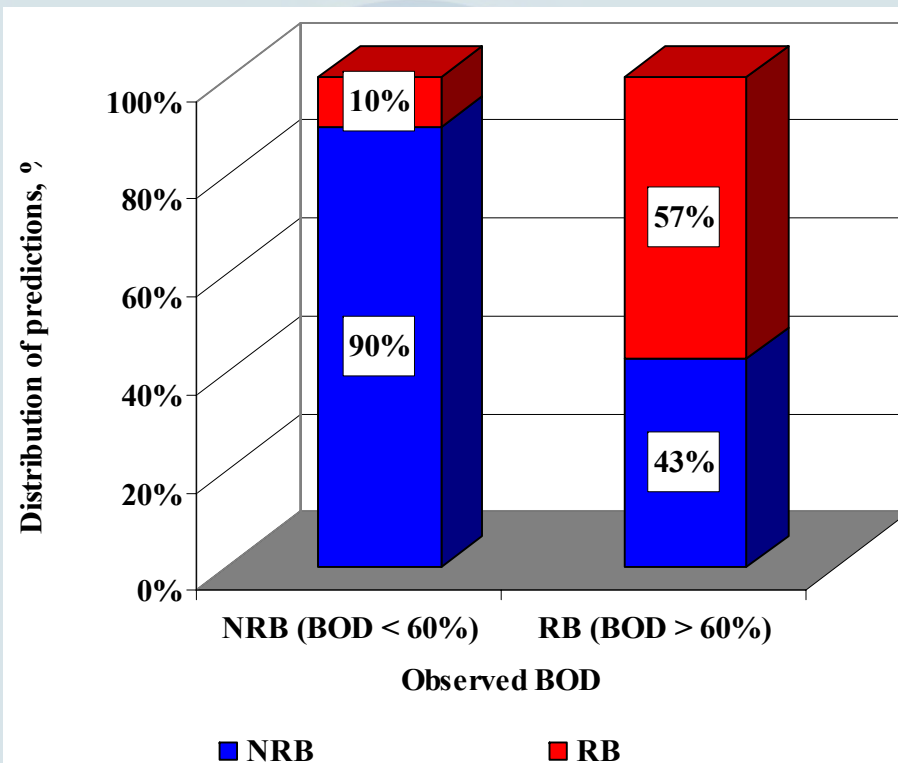
New chemicals: 1123

Total: 1461

Statistics

Without accounting for domain

Accounting for domain



Summary

The committee recommended that.....

- Four biodegradation models (BIOWIN5,6 CERI model CATABOL) are acceptable for using for the screening purpose.
- CATABOL should be used mainly because CATABOL is only model that is based on biodegradation pathway and provide many useful information to assess the biodegradability of a chemicals.
- In addition, CERI model and BIOWIN5 should be used to obtain prediction from different view point.

Summary

NITE

The committee recommended that.....

- Four biodegradation models (BIOWIN5,6 CERI model CATABOL) are acceptable for using for the screening purpose.
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- In addition, CERI model and BIOWIN5 should be used to obtain prediction from different view point.

Outlook

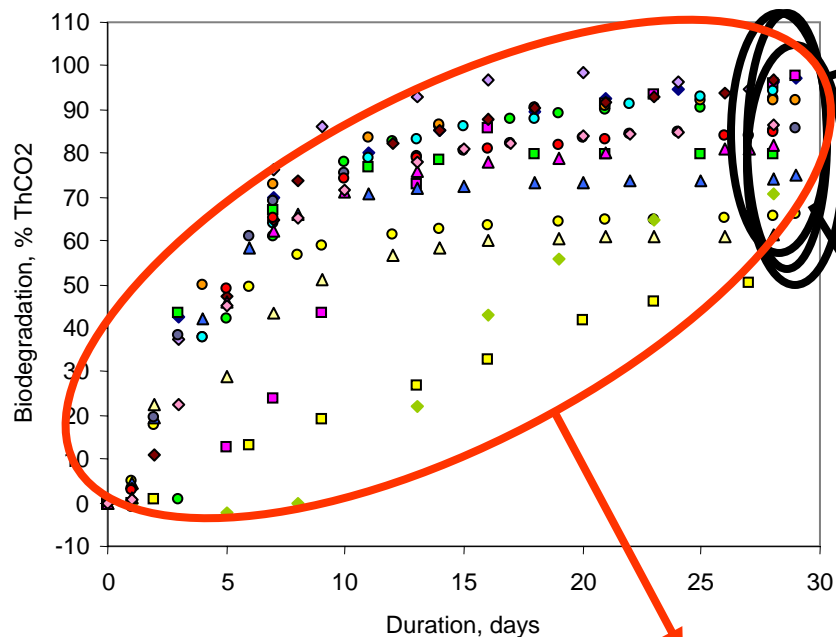
- **QSAR and Complexity of Chemical Structure**
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- **Biodegradation kinetic models**
- **Simulating the effect of gene modification on metabolism**

Kinetic Models

*Cooperation with BASF, ExxonMobil,
Givaudan, Dow Chemicals*

Laboratory of Mathematical Chemistry

CATABOL^{TD} – time dependent model



MITI I test: 1090 chemicals
%ThOD at 28th day

MITI I test: 266 chemicals
Metabolites quantities at 28th day

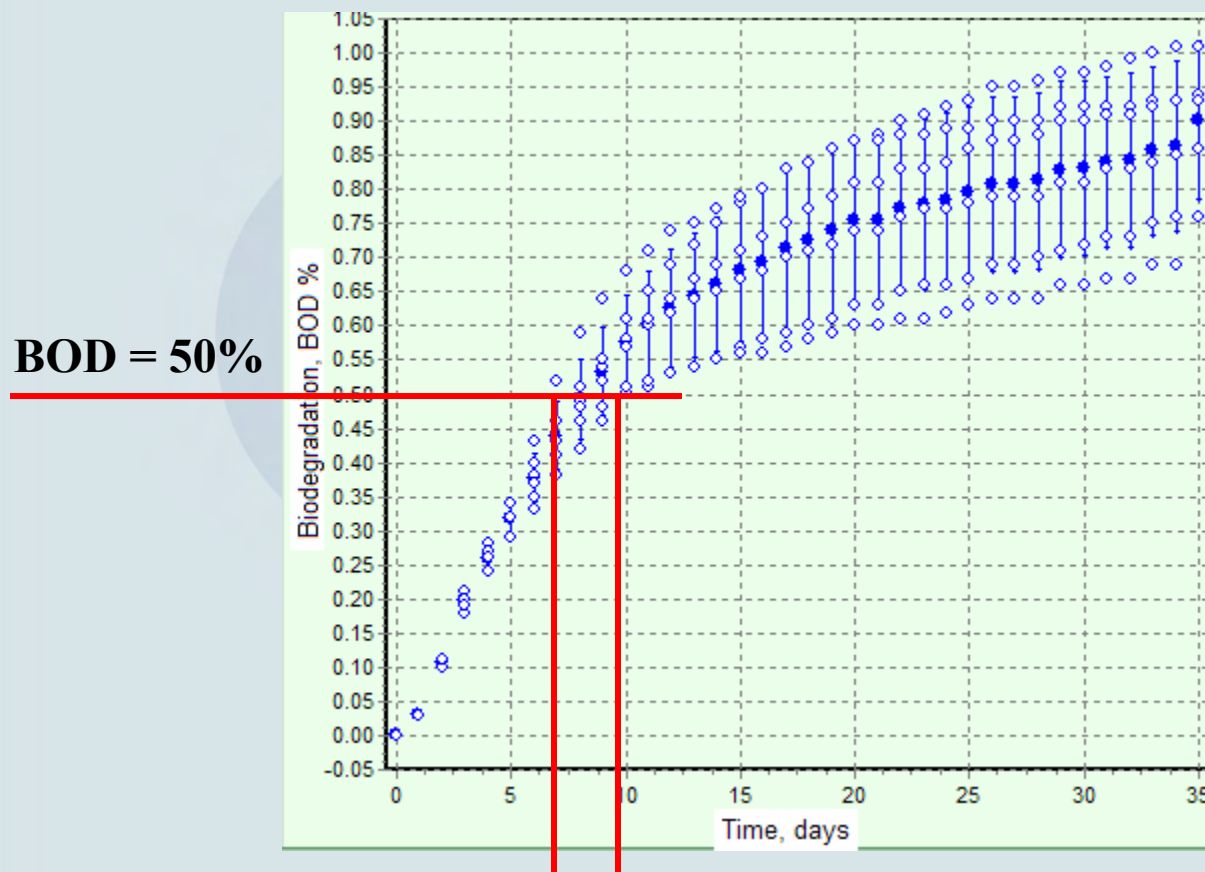
OECD 301B: 108 chemicals
%ThCO₂ at 28th day

BASF: 301A-F, ISO 14593: ~ 500 chemicals
Time series for %DOC, %ThOD or %ThCO₂

Metabolic studies with microorganisms: ~ 300 chemicals
(Pseudomonas, Spingomonas, Rhodococcus, etc.)
Documented metabolic pathways

CATABOL^{TD} – time dependent model

Training data



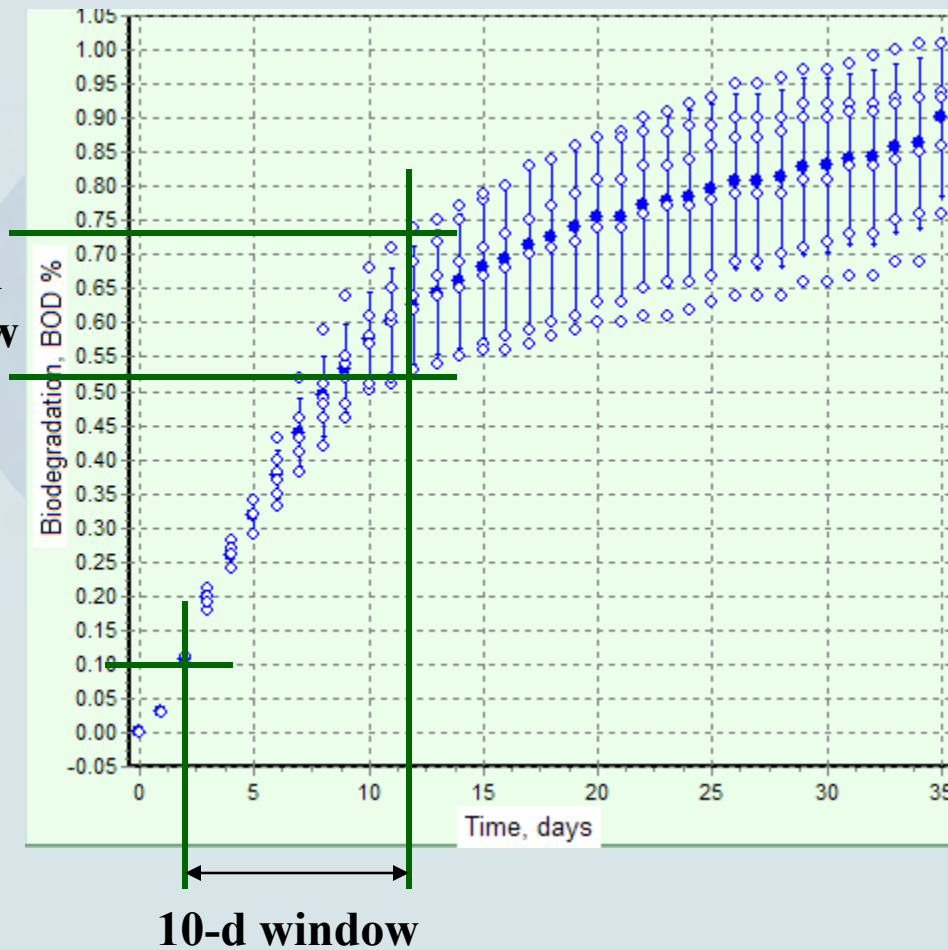
$t_{1/2}^{\text{Ultimate}}$

Ultimate half life

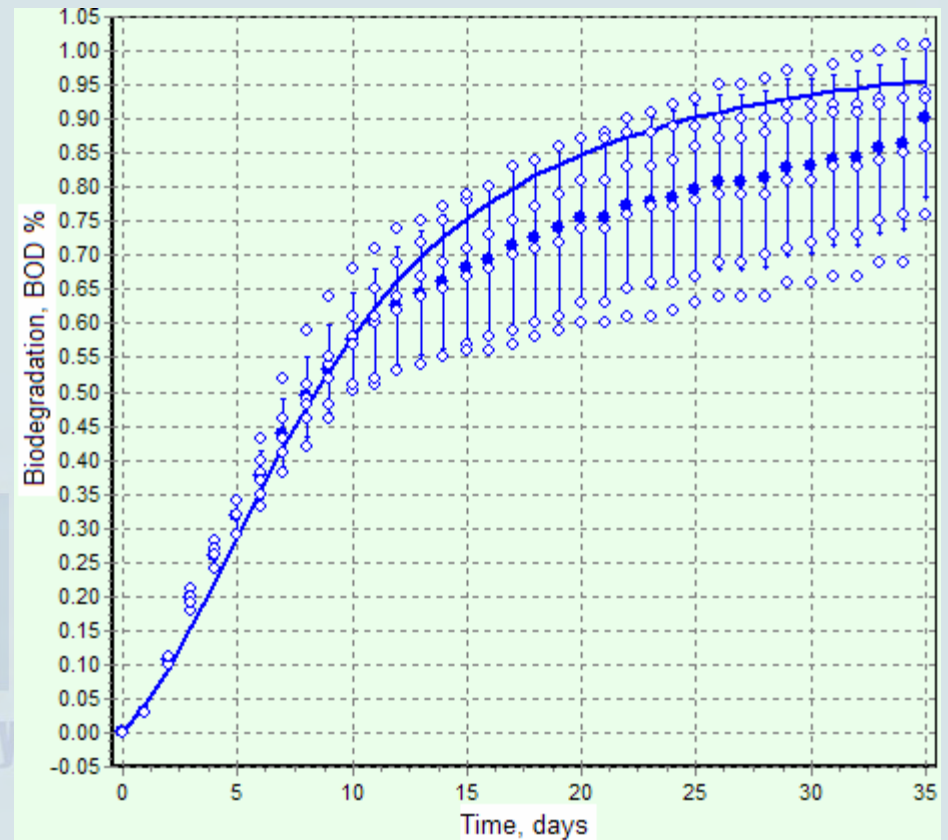
CATABOL^{TD} – time dependent model

Training data

**BOD within
10-d window**



CATABOL^{TD} – time dependent model



$$BOD^{Calc} = f(P_i) = f[P_i(k_i, t)]$$

$$\min_k RSS = \sum_n \sum_t \left(BOD_{n,t}^{Obs} - BOD_{n,t}^{Calc} \right)^2$$

P approximated by first order kinetics

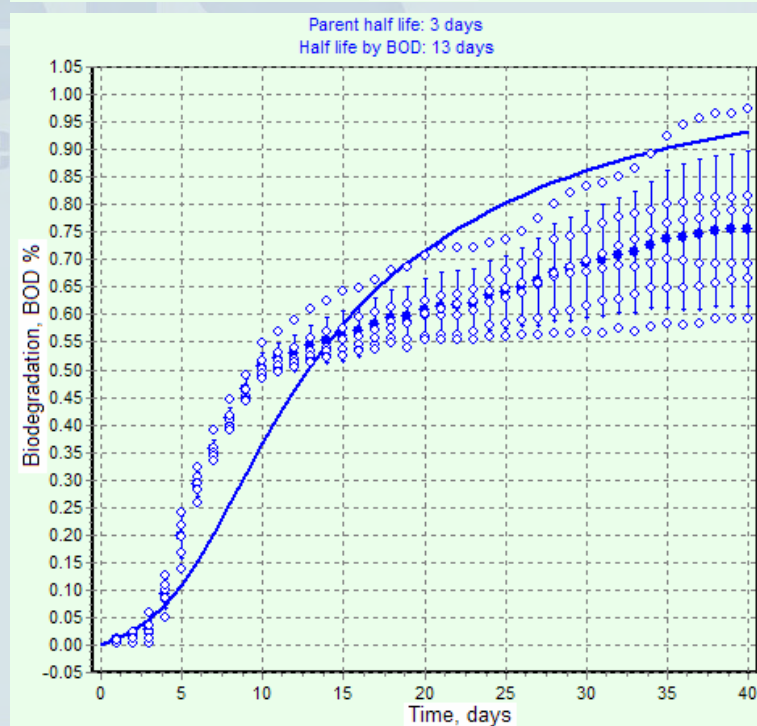
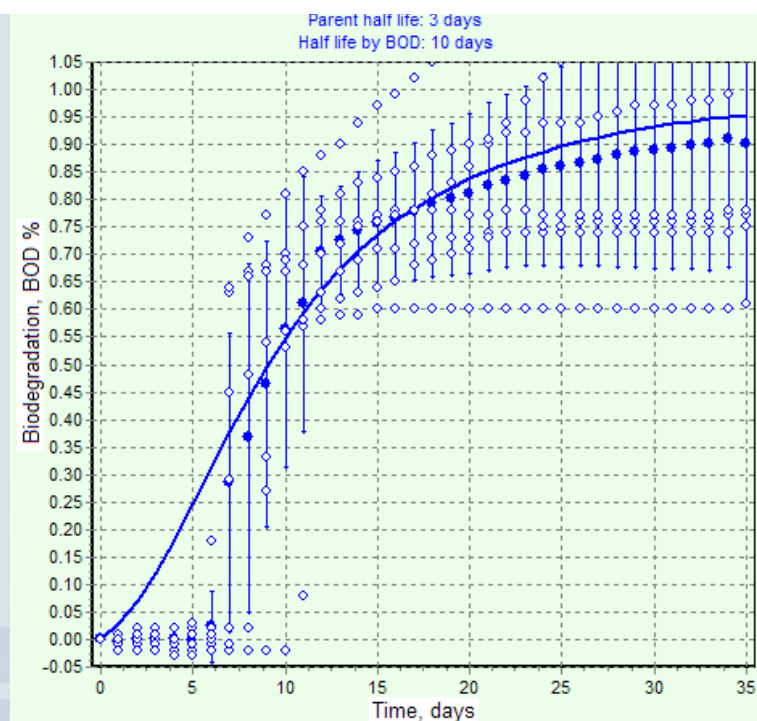
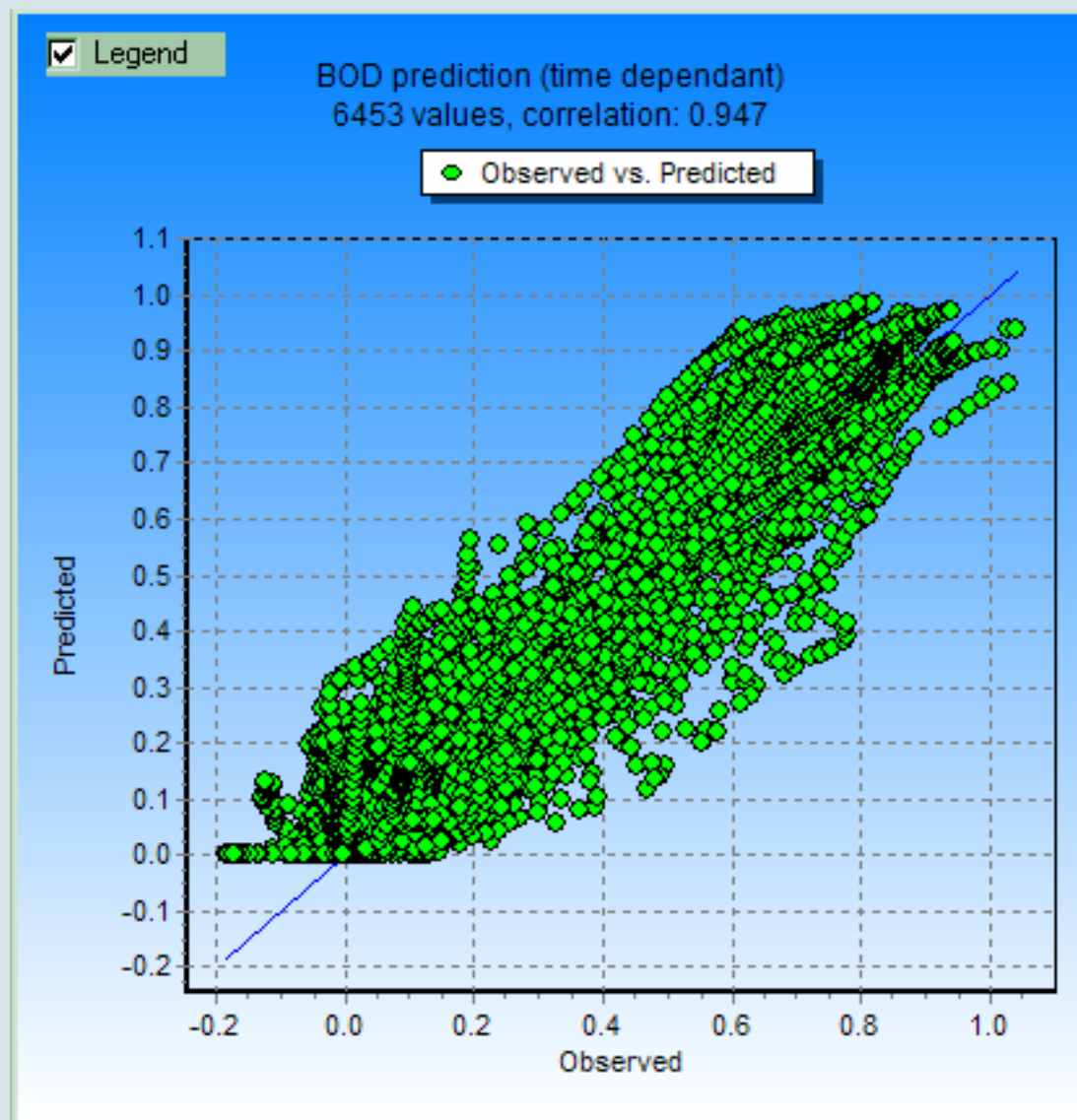
$$P_t = (1 - \exp(-kt))$$

P approximated by second order kinetics

$$P_t = 1 - \frac{1}{1 + kt[S]_0} =$$

$$= 1 - \frac{1}{1 + k't}$$

CATABOL^{TD} – time dependent model



CATABOL^{TD} – time dependent model

The model is able to predict within OECD 301F test:

- 1. Primary half-life – half-life of parent chemical**
- 2. Ultimate half-life – half-life by BOD**
- 3. Biodegradation at different days**
- 4. Metabolites quantity at different days**
- 5. Biodegradation within 10 days window**



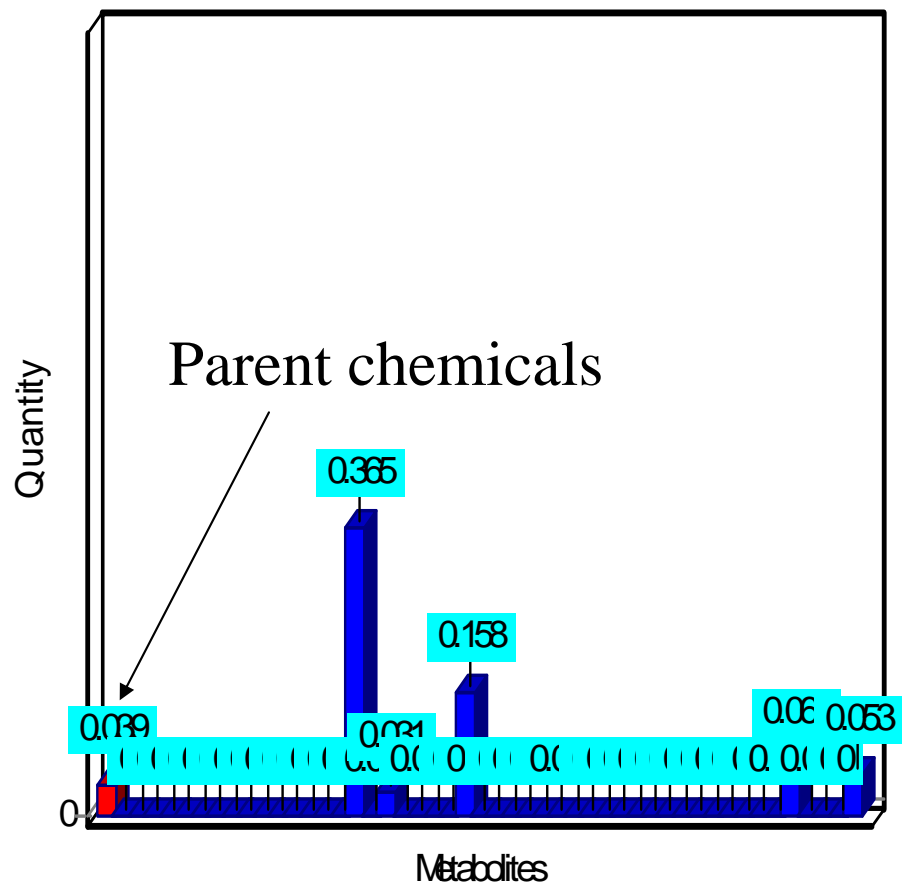
CATABOL^{TD} – time dependent model

The model is able to predict within OECD 301F test:

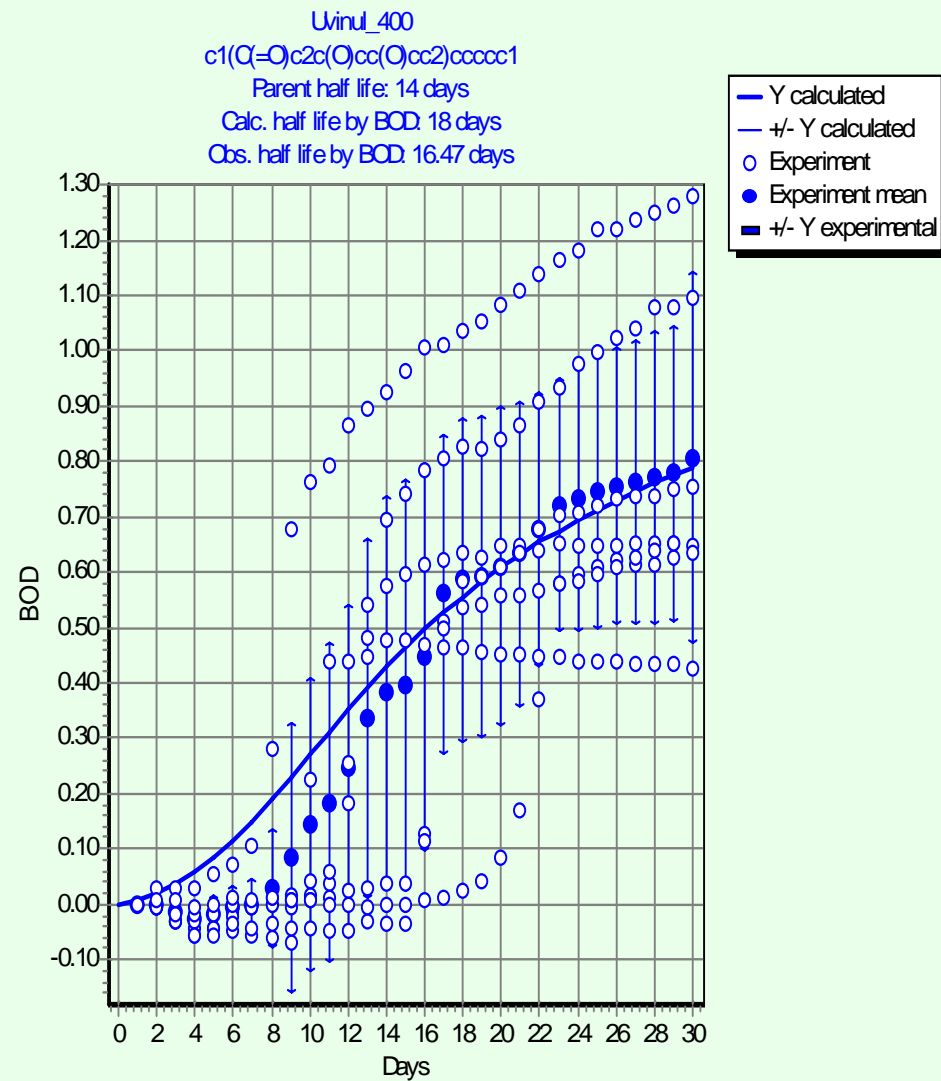
- 1. Primary half-life – half-life of parent chemical**
- 2. Ultimate half-life – half-life by BOD**
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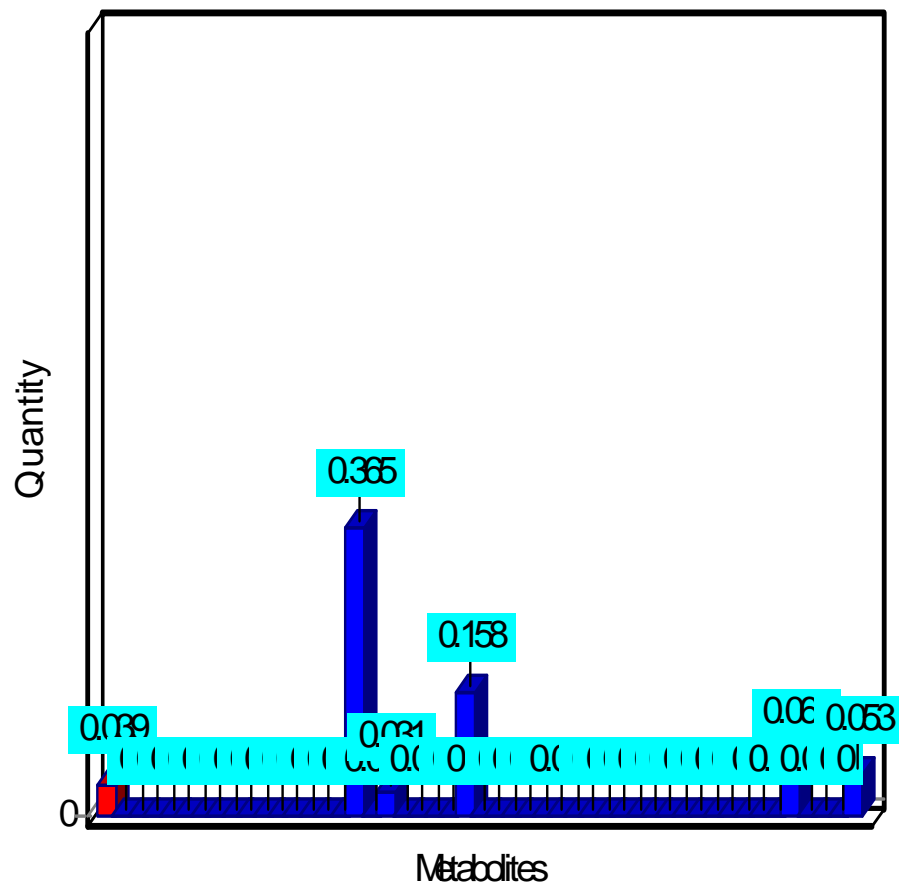
Metabolites distribution



c1(C=O)c2c(O)cc(O)cc2)ccccc1
Transformation 0, Quantity=0.0386



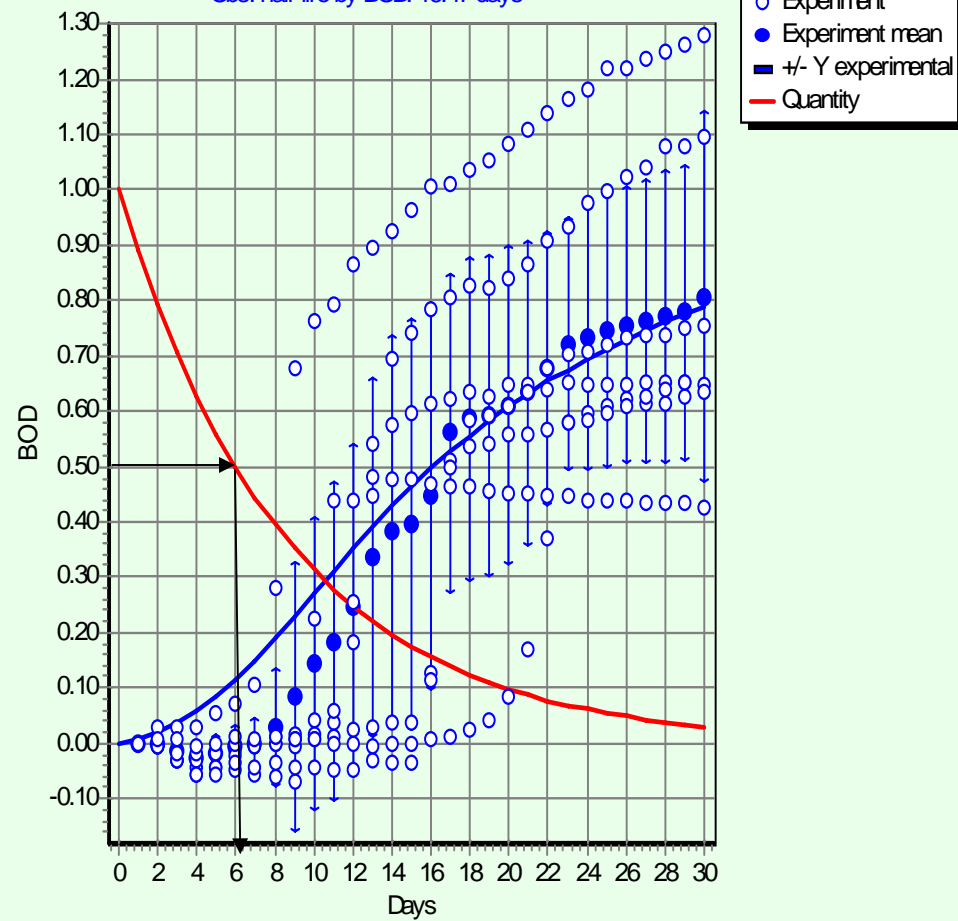
Metabolites distribution



c1(Q=Qc2x(Qx(Qx2)xxxxc1
Transformation0, Quantity=0.0386

Primary half live – 6 days

Calc. half life by BOD: 18 days
Obs. half life by BOD: 16.47 days



CATABOL^{TD} – time dependent model

The model is able to predict within OECD 301F test:

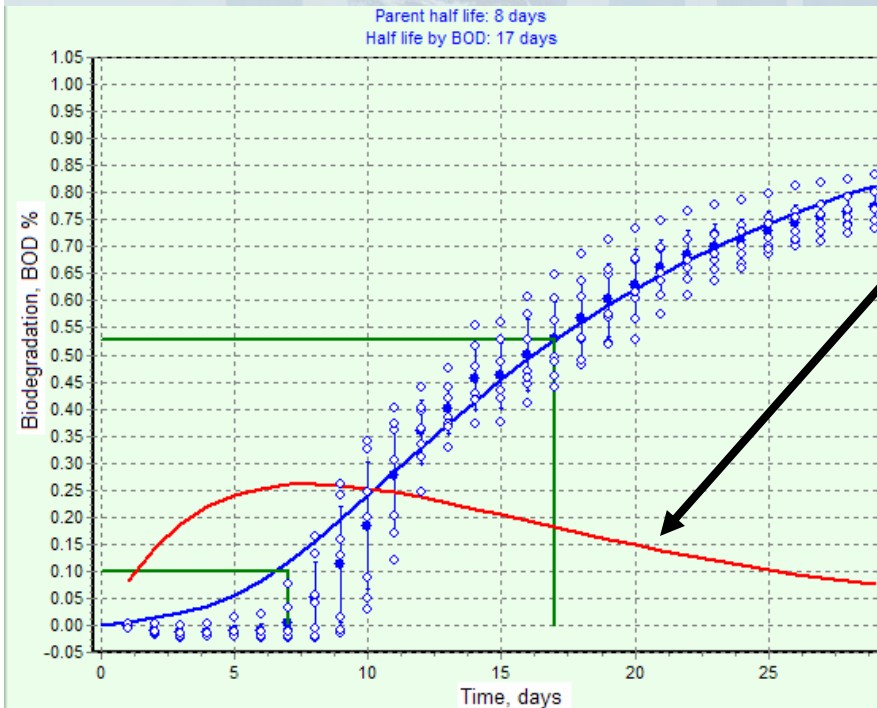
- 1. Primary half-life – half-life of parent chemical**
- 2. Ultimate half-life – half-life by BOD**
- 3. Biodegradation at different days**
- 4. Metabolites quantity at different days**
- 5. Biodegradation within 10 days window**



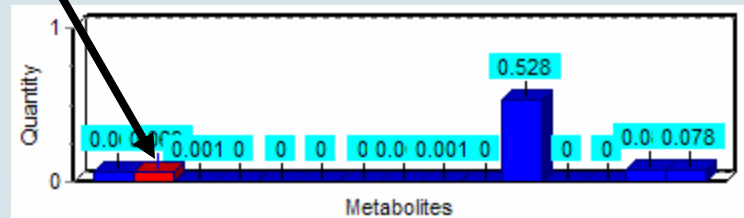
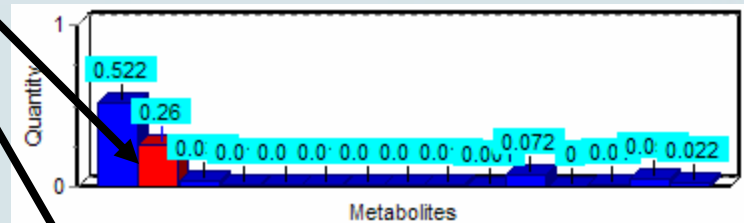
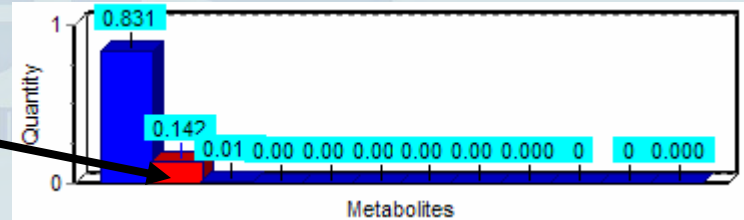
CATABOL^{TD} – time dependent model

The model is able to predict within OECD 301F test:

1. **Primary half-life** – half-life of parent chemical
2. **Ultimate half-life** – half-life by BOD
3. **Biodegradation at different days**
4. **Metabolites quantity at different days**
5. **Biodegradation within 10 days window**



Metabolite



Outlook

- **QSAR and Complexity of Chemical Structure**
- **Toxicity as a result of metabolic activation**
- **Metabolism logic**
- **Probabilistic approach to modeling metabolism**
- **CATABOL for simulating microbial degradation**
- **Performance and reliability of predicted metabolites**
- **Biodegradation kinetic models**
- **Simulating the effect of gene modification on metabolism**

Simulating the effect of gene modification on metabolism

(Microbial degradation)

Depict 2 trees

Compare with single tree # 1

Average similarity : 75.00%

Cell Height 100 Cell Width 100

Cell Height 100 Cell Width 100

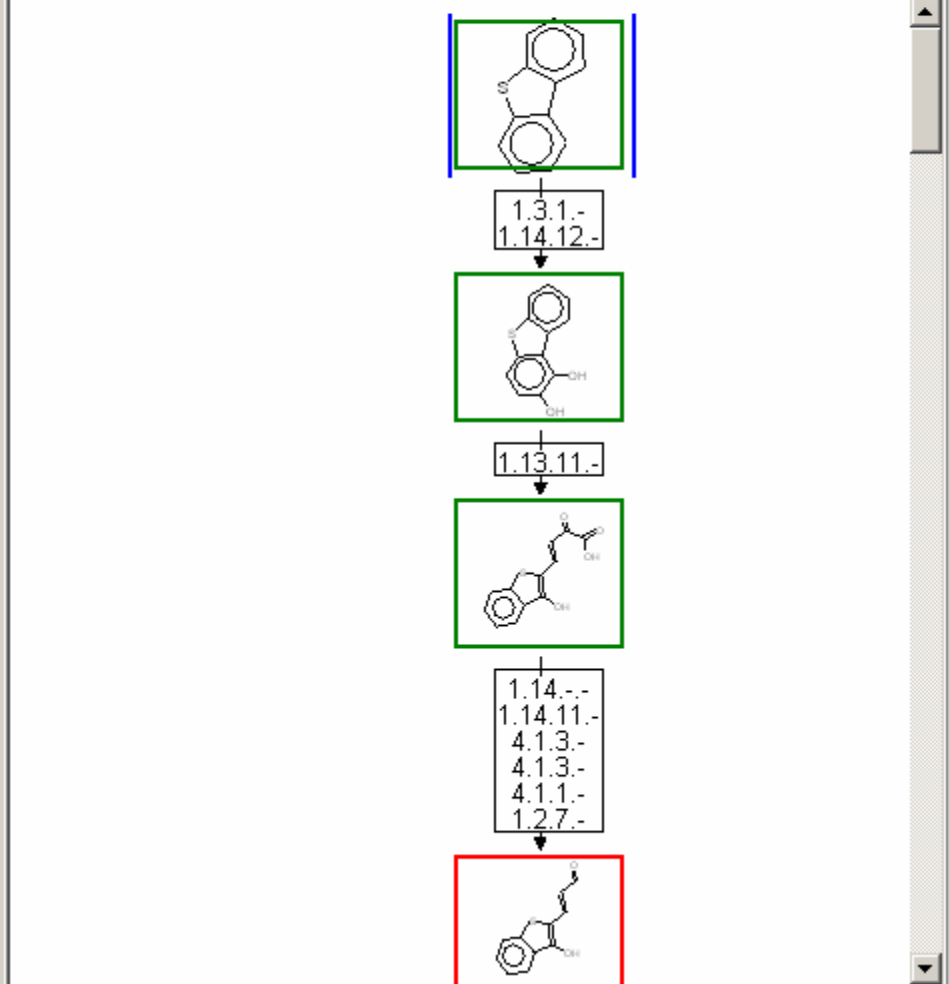
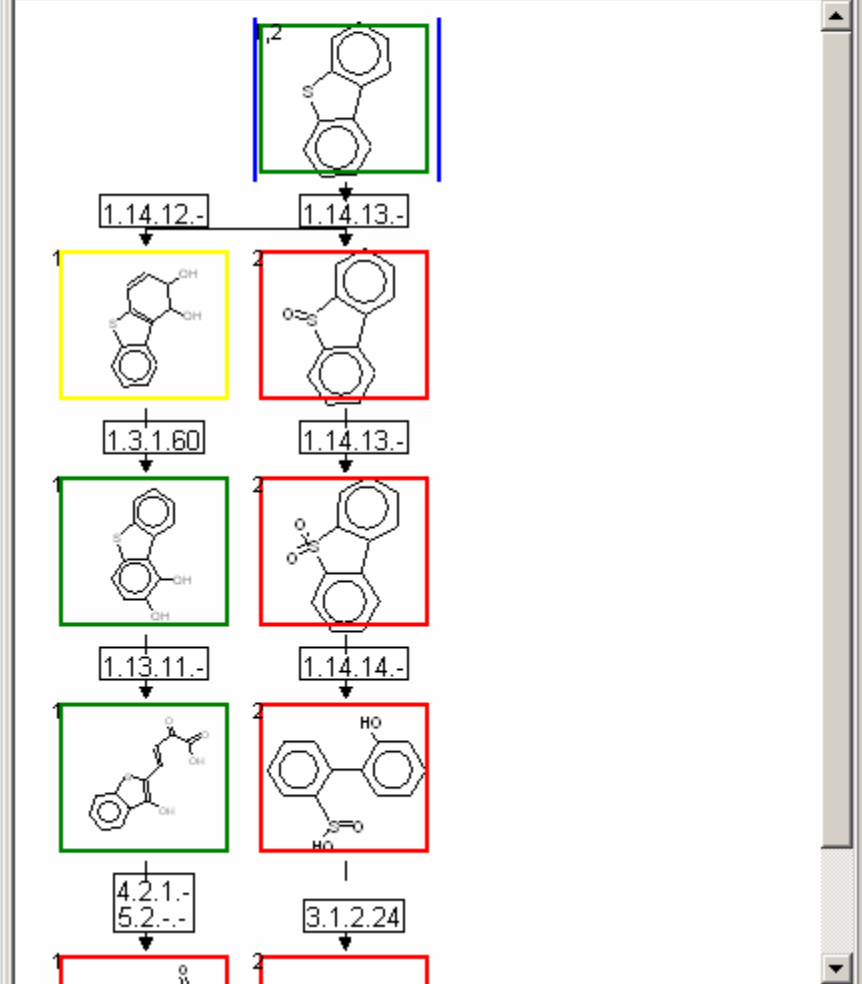
Redraw Print Preview Filter options

Redraw Print Preview Filter options

Reference: http://umbbd.ahc.umn.edu/dbt2/dbt2_map.html
http://umbbd.ahc.umn.edu/dbt/dbt_map.html

L=0. Q=0.000. R=1.000. P(obtain)=0.000. P(metabolize)=0.000

L=0. Q=0.959. R=1.000. P(obtain)=1.000. P(metabolize)=0.000



Depict 2 trees

Compare with single tree #

Average similarity : 75.00%

Cell Height Cell Width

Cell Height Cell Width

Redraw Print Preview Filter options

Redraw Print Preview Filter options

Cell Height Cell Width

Cell Height Cell Width

Redraw Print Preview Filter options

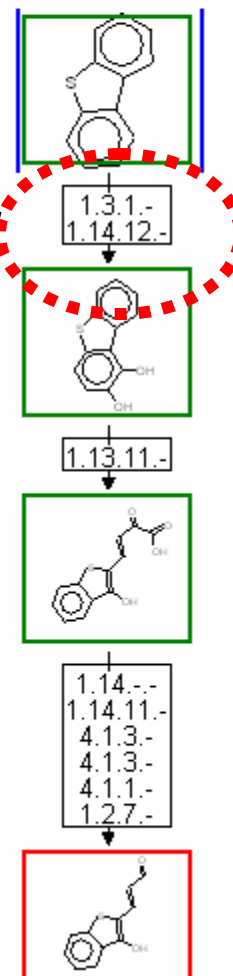
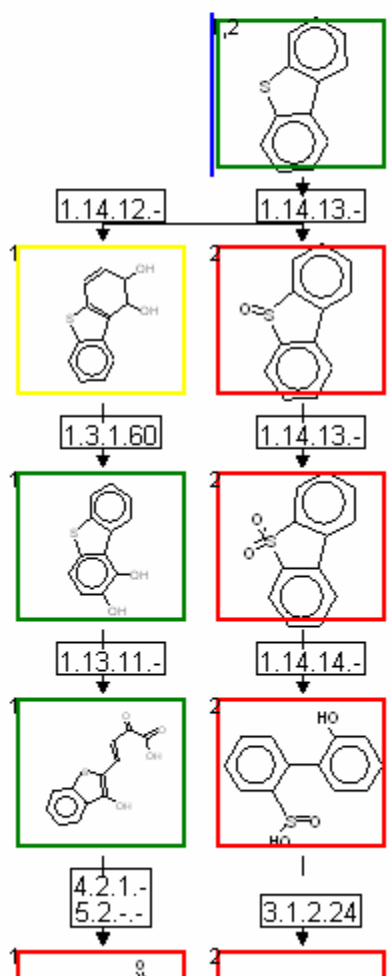
Show most probable pathways Reliability threshold

Reference: "http://umbdd.ahc.umn.edu/dbt2/dbt2_map.html:
 http://umbdd.ahc.umn.edu/dbt/dbt_map.html"

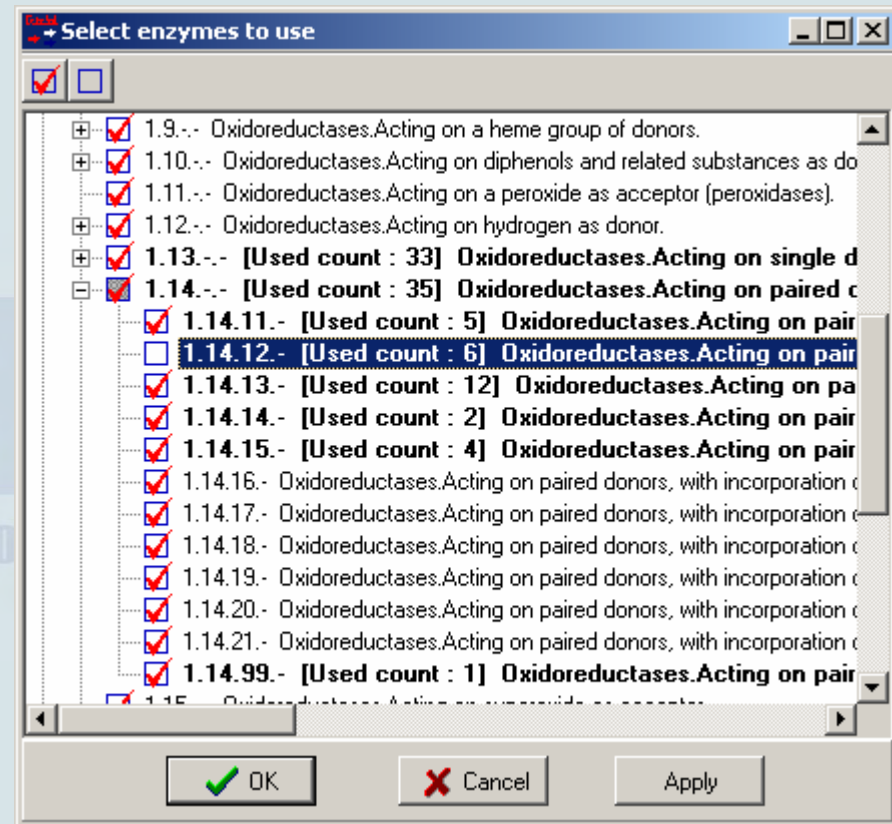
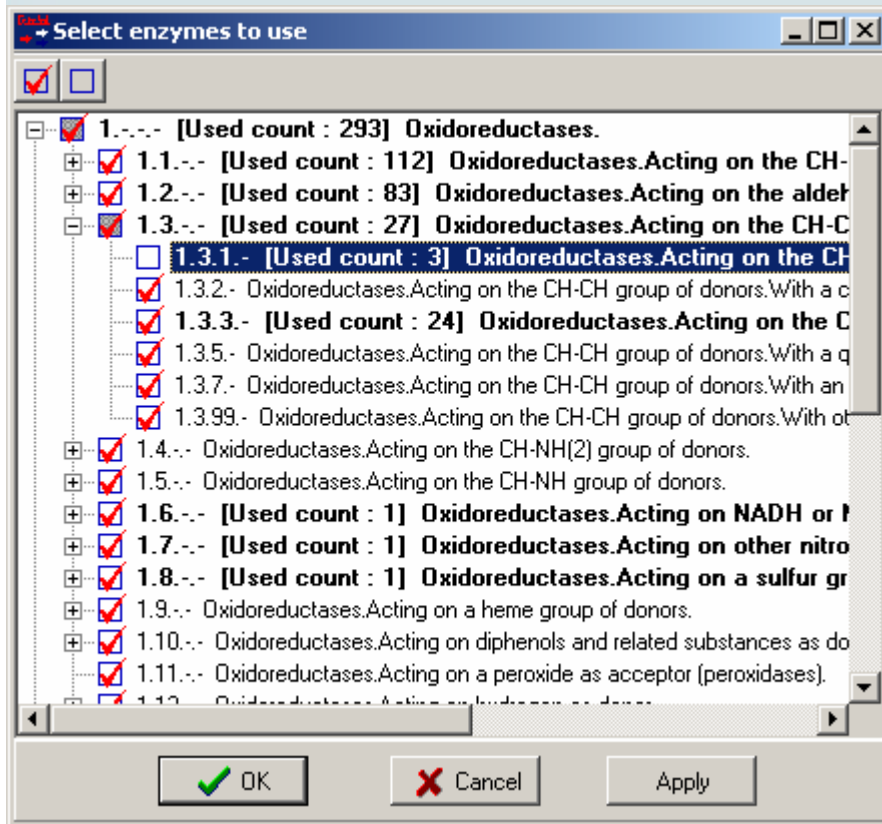
L=0. Q=0.000. R=1.000. P(obtain)=0.000. P(metabolize)=0

L=0. Q=0.959. R=1.000. P(obtain)=1.000. P(metabolize)=0

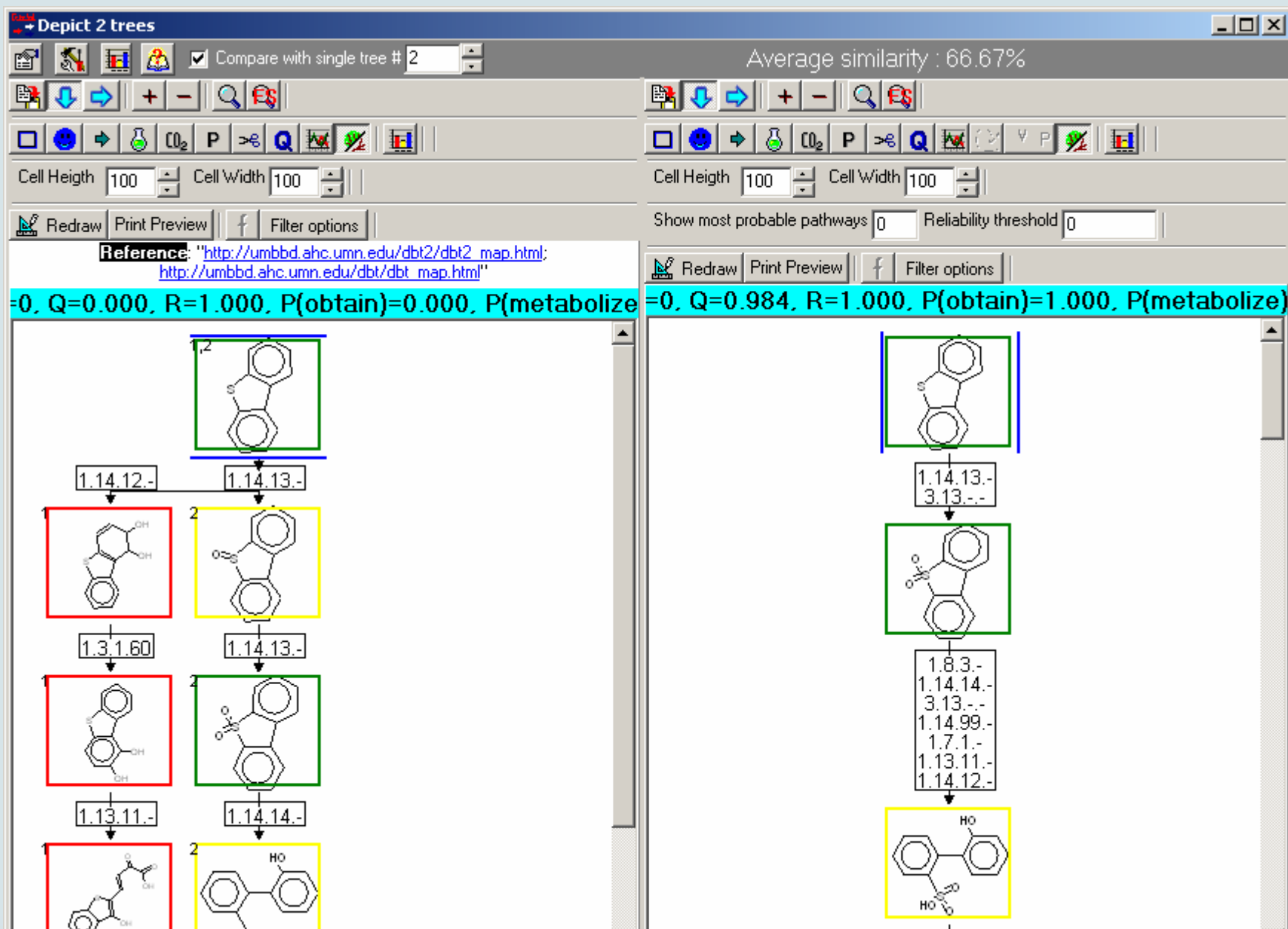
Oxidoreductases. Acting on
 paired donors...



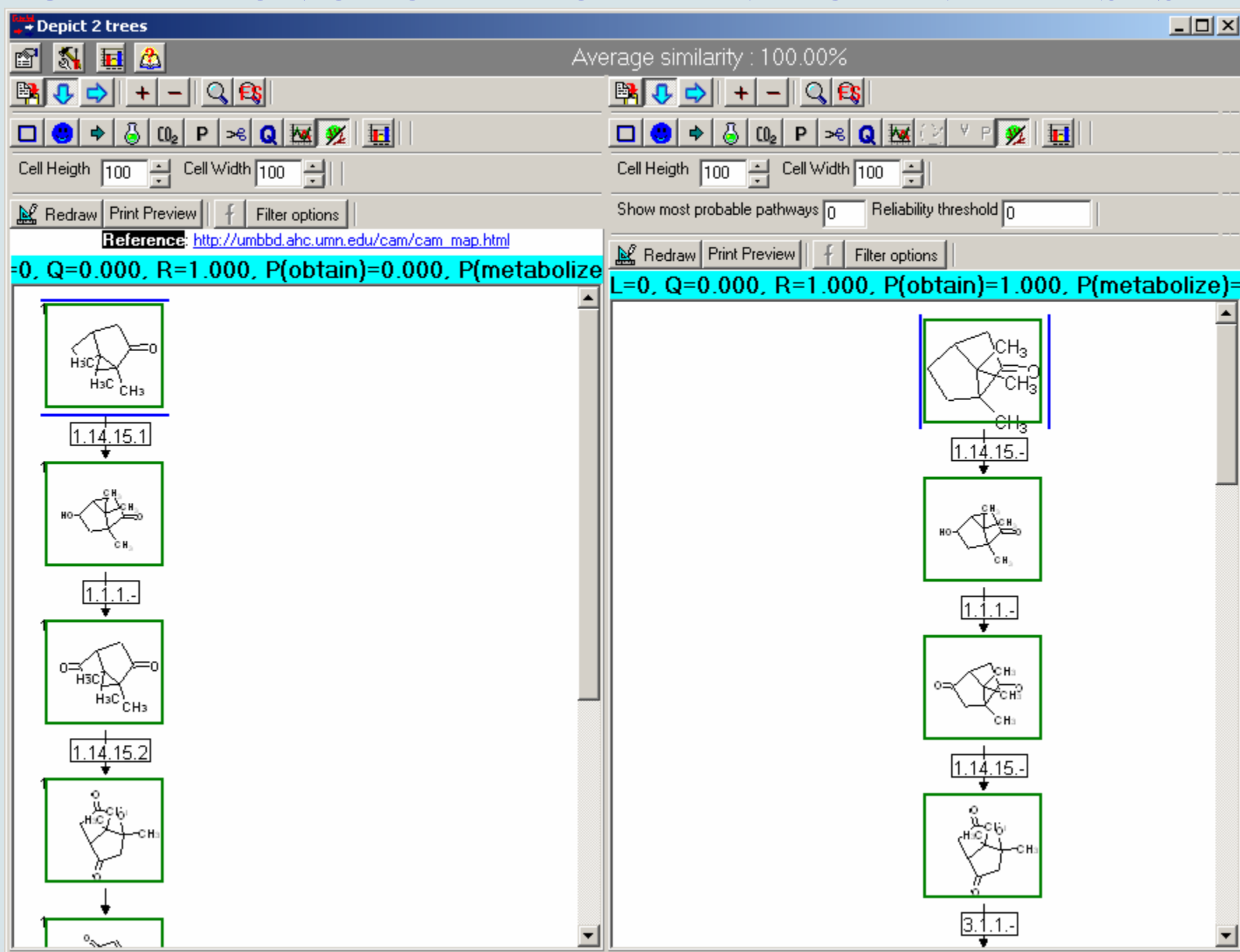
ENZYME MODIFICATIONS IN THE MICROBIAL SYSTEM



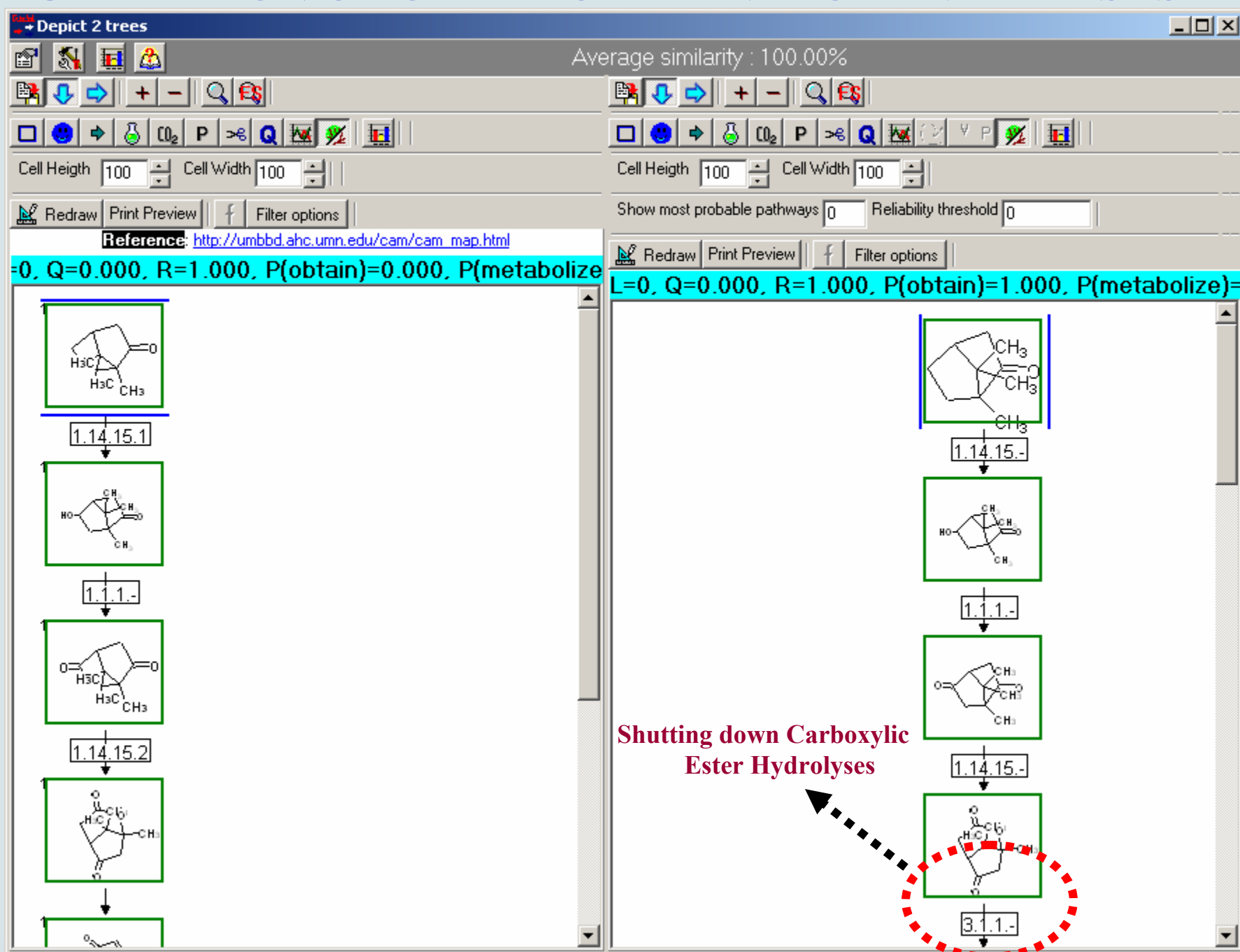
CHANGES IN DEGRADATION PATHWAY DUE TO ENZYME MODIFICATIONS



DEGRADATION OF CAMPHORE – INTACT ENZYME SYSTEM

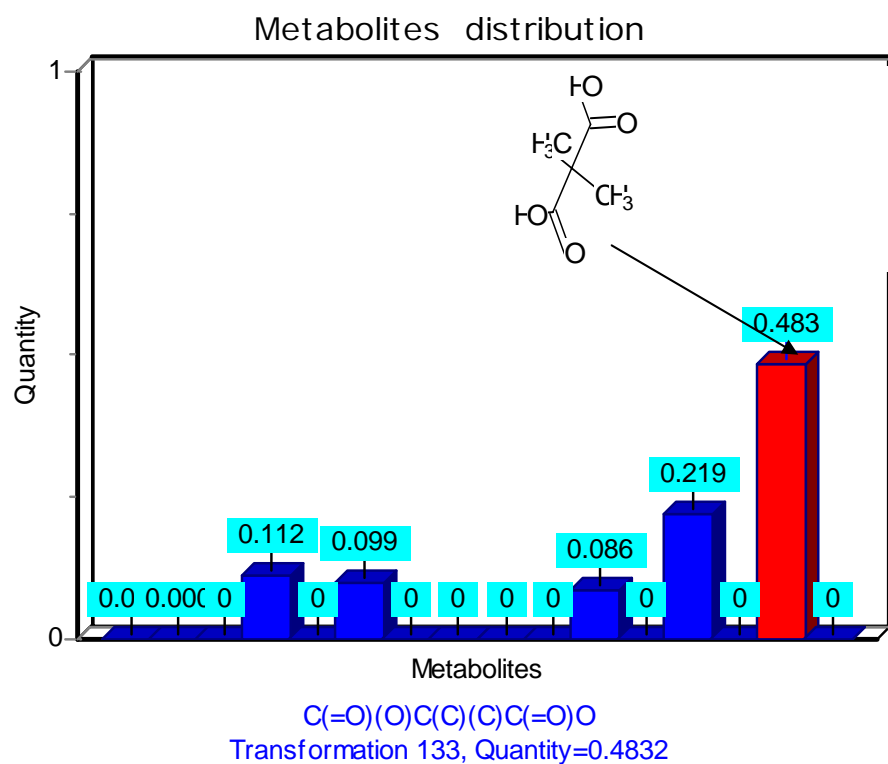


DEGRADATION OF CAMPHORE – INTACT ENZYME SYSTEM

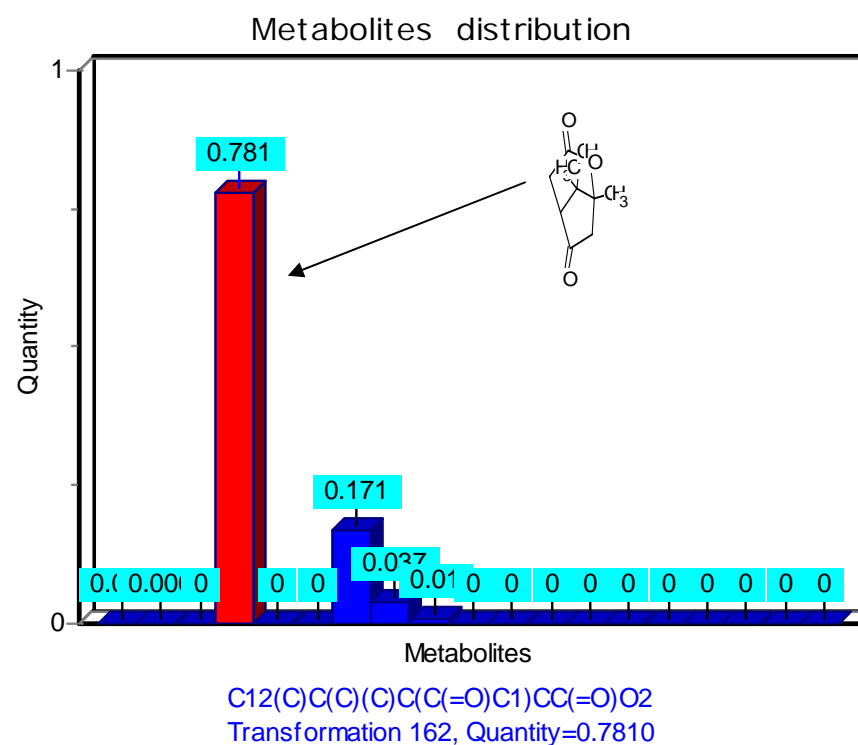


DEGRADATION OF CAMPHORE – ELIMINATING HYDROLASES IN THE ENZYME SYSTEM

INTACT ENZYME SYSTEM



NO HYDROLASES IN THE ENZYME SYSTEM



Conclusions

- **The advantage of CATABOL is the combination of a knowledge system with a predictive capabilities**
- **For environmental risk assessment, its key value is its ability to show possible metabolites resulting from partial biodegradation that might pose issues in the environment**
- **It's complexity requires interpretation of the results by expert**

Questions & Problems

- **MITI (301C) vs. other 301 protocols?**
- **Single protocol vs. most appropriate one (“best scenario”)**
- **Effect of bacterial toxicity on biodegradation**
- **Peer review of biodegradation transformations**
- **Data consistency and acquisition**

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Givaudan



EUROPE

